

From Purification to Liquefaction Gas Processing with Axens' Technology

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Contaminants in natural gas:

- CO₂
- H_2S
- Mercaptans (RSH)
- Carbonyl sulfide (COS)
- Carbon sulfide (CS₂)
- *H*₂0
- Hg

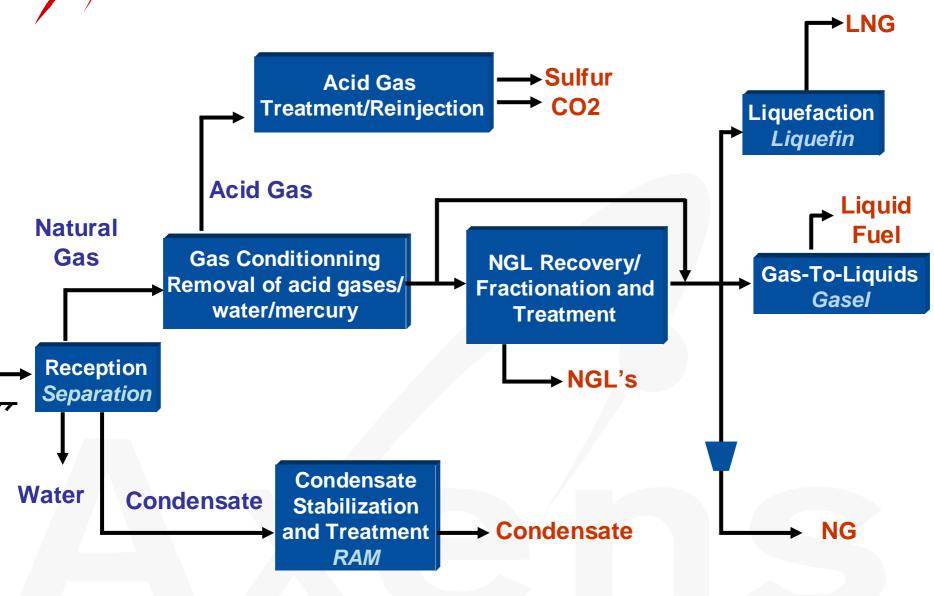
The degree of removal depends on downstream requirements.



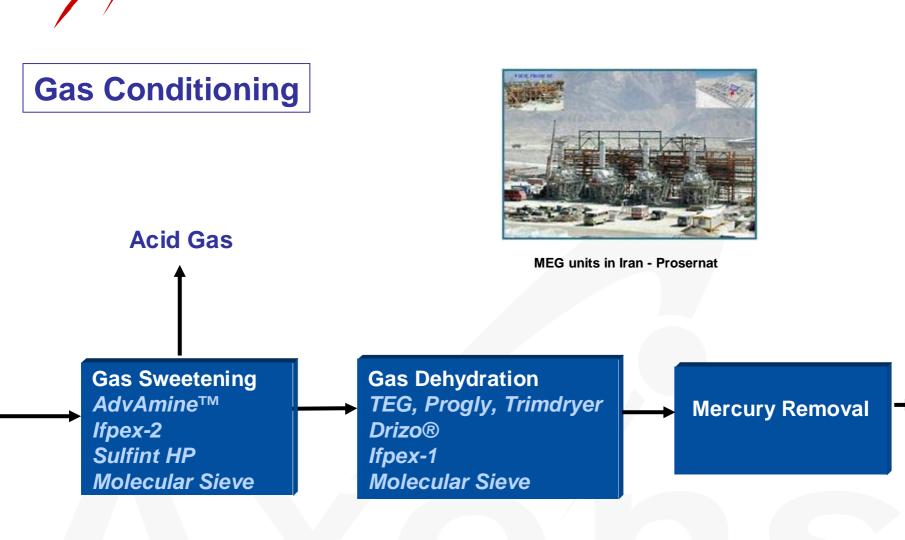
In LNG plants, contaminants must meet the following specifications:

- CO₂ < 100 ppmv (freezing and equipment plugging)
- H₂S < 4 ppmv (sales gas specification)
- H_20 < 1 ppmv (freezing and equipment plugging)
- Hg <0.01 μgr/Nm³ (corrosion in aluminum exchangers)

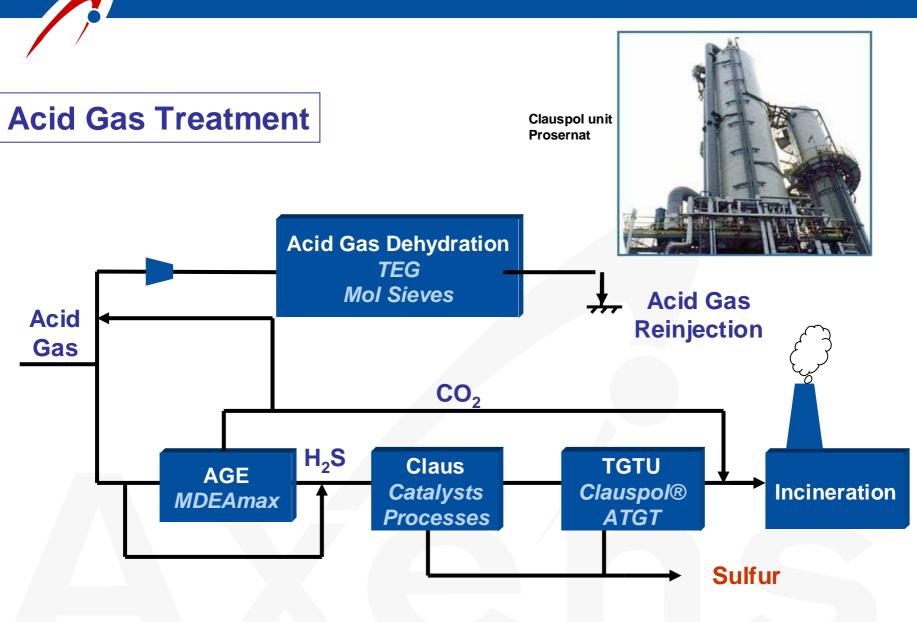
Gas Processing Chain



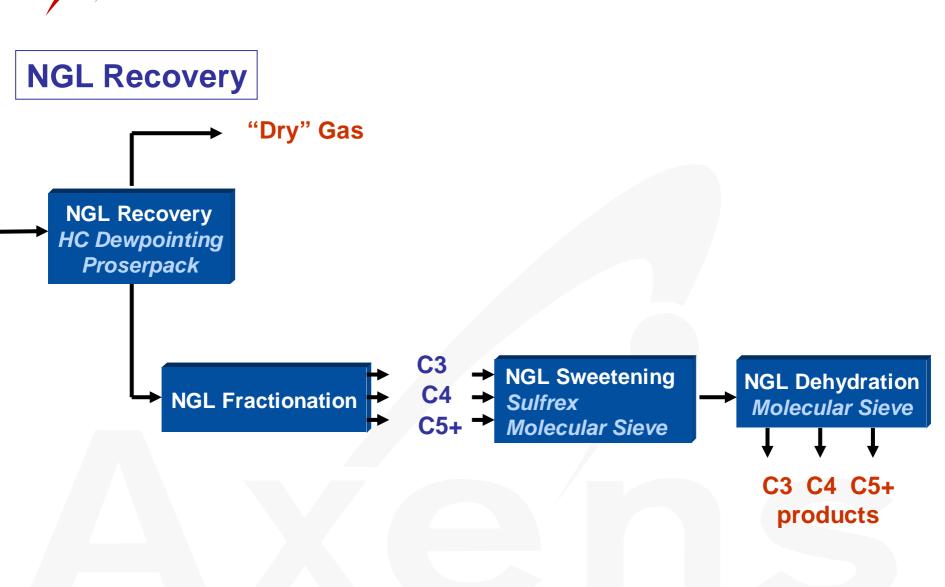
Gas Processing Chain (1/3)



Gas Processing Chain (2/3)



Gas Processing Chain (3/3)





Combining Sulfrex & Molecular Sieves for LPG Sweetening

Synergy between Process and Adsorbent

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Sulfrex Process Objectives

- Obtaining negative Doctor Test on final product
- Odor improvement
- Meet sulfur specifications on LPG or C₃/C₄ cuts
- Meet C₃/C₄ olefinic cuts sulfur specs required for olefins upgrading process (Alkylation, Oligomerization)

LPG Sweetening, Extractive Sulfrex

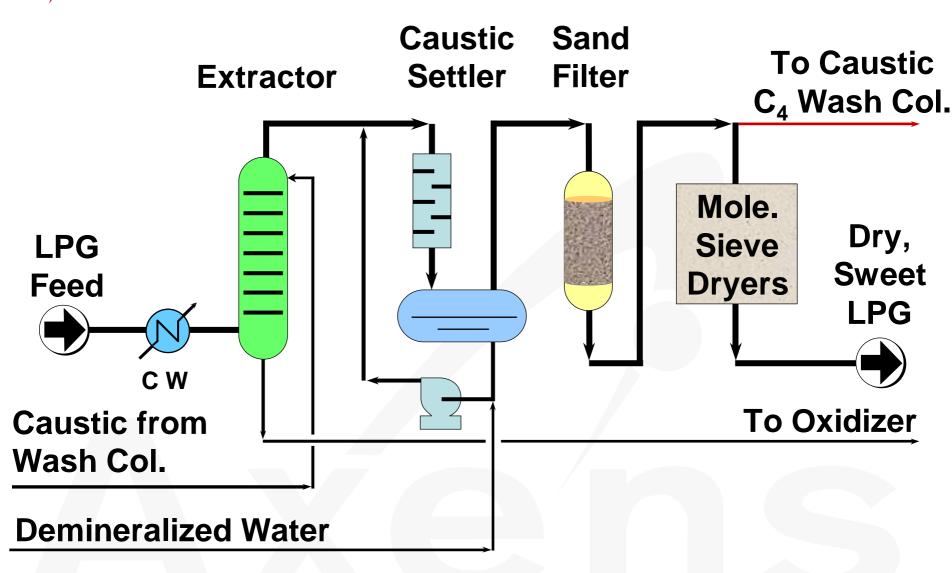
Typical C₄ Cut Impurities

- Mercaptans
 - methyl mercaptan: CH₃-SH up to 1000 ppm mole
 - ethyl mercaptan: C₂H₅-SH up to 5000 ppm mole
- H₂S: traces
- COS: traces (*)
- (*) : In C_3 cut COS content can be as high as several hundred ppm

LPG Sweetening, Extractive Sulfrex Main Reactions Extraction \rightarrow RSNa + H₂O RSH + NaOH **Oxidation and Caustic Regeneration** catalyst 2 NaOH + R-SS-R $2 \text{ RSNa} + \frac{1}{2} \text{ O}_2 + \text{ H}_2 \text{ O}_2$ **Global Reaction** 2 RSH + 1/2 O₂ $R-SS-R + H_2O$

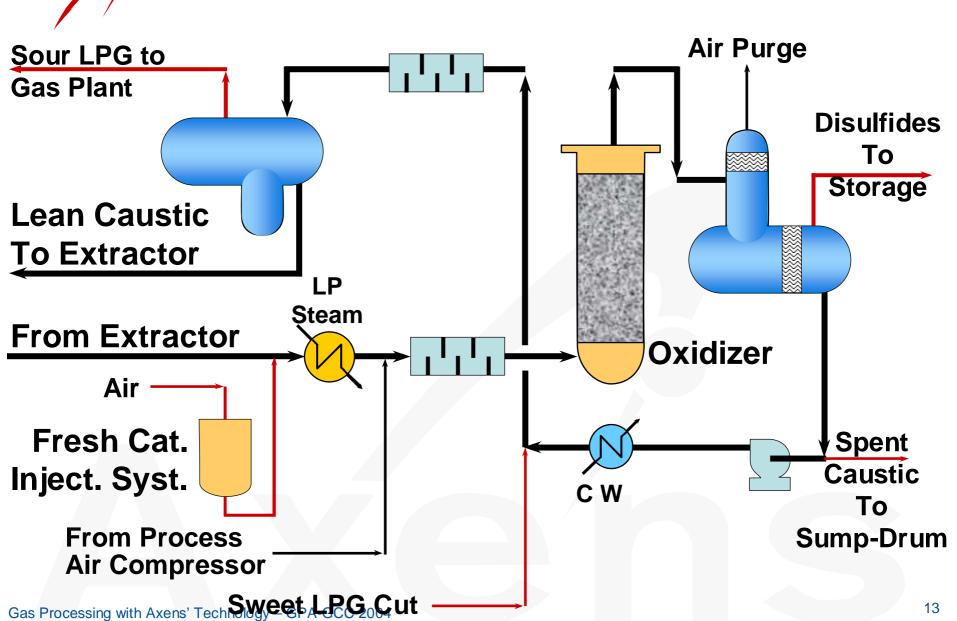
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Extractive Sulfrex, Extraction Section



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Extractive Sulfrex[™] Caustic Regeneration Section





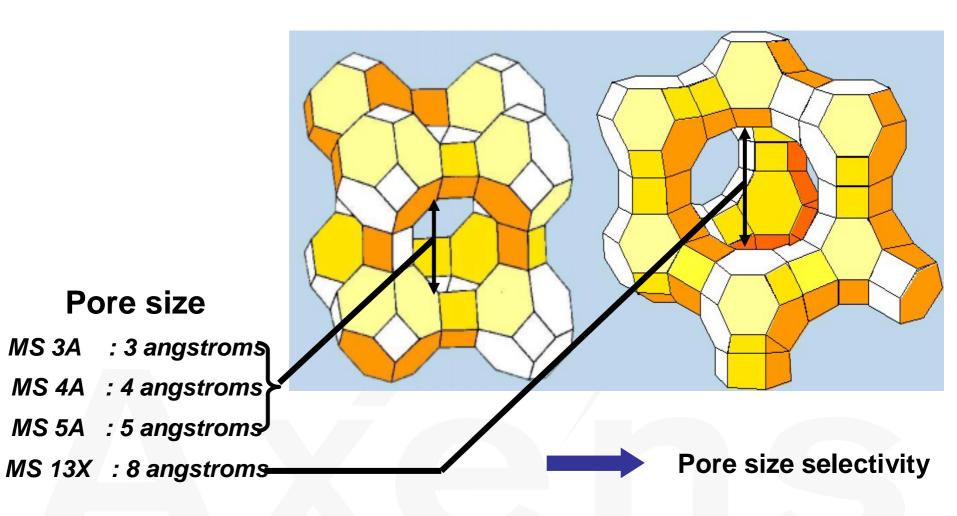
17 licensed units

6.2 millions tons per year of capacity

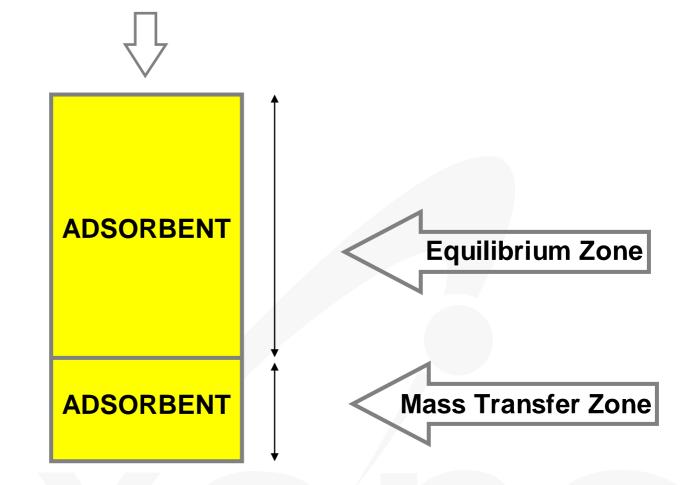
Possibility to add molecular sieves to:

- decrease the size/investment of Sulfrex unit
- insure low RSH outlet concentration

Molecular Sieve Structure



Adsorber Configuration



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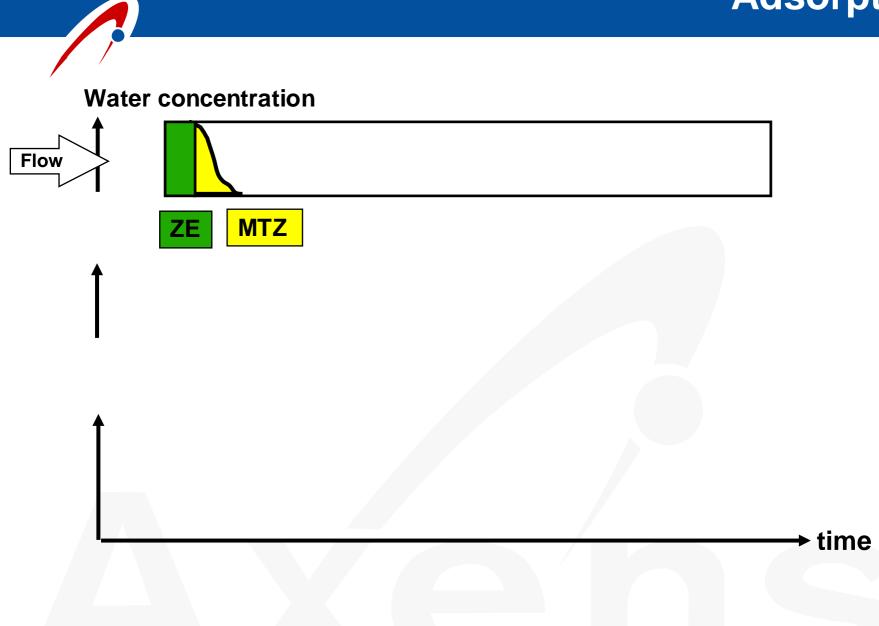


- The adsorbent is saturated with adsorbate at the operating temperature and pressure, according to the isotherm
- The adsorbate concentration is constant throughout the equilibrium zone

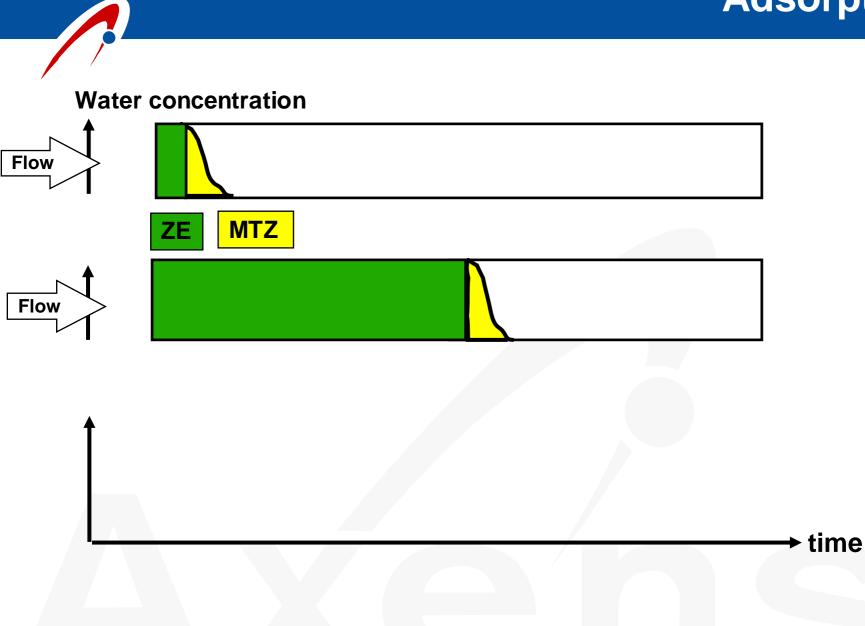
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 The adsorbate concentration is lower than in the equilibrium zone - an almost linear gradient between the concentration at the interface with the equilibrium zone and the adsorbent bed outlet (zero).

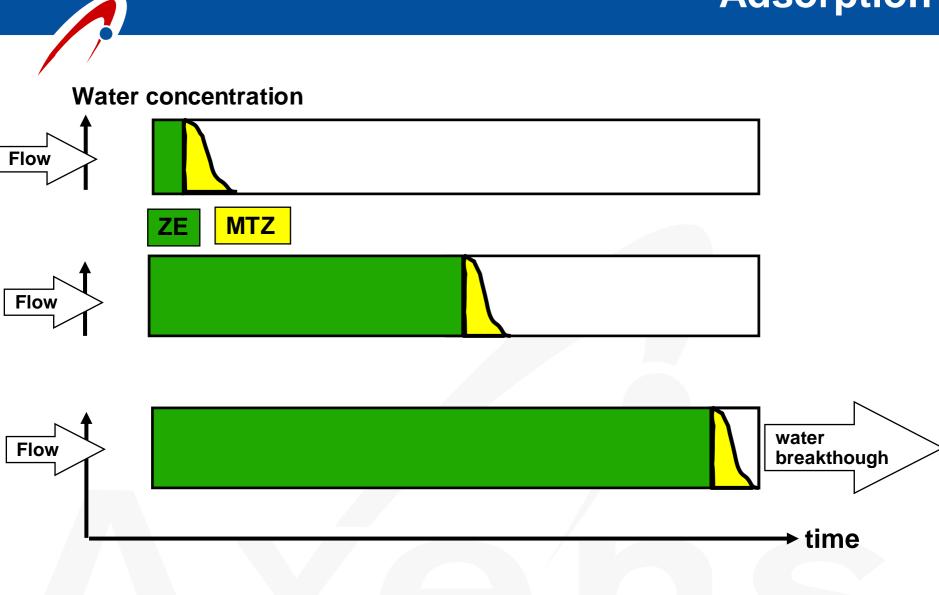
Adsorption



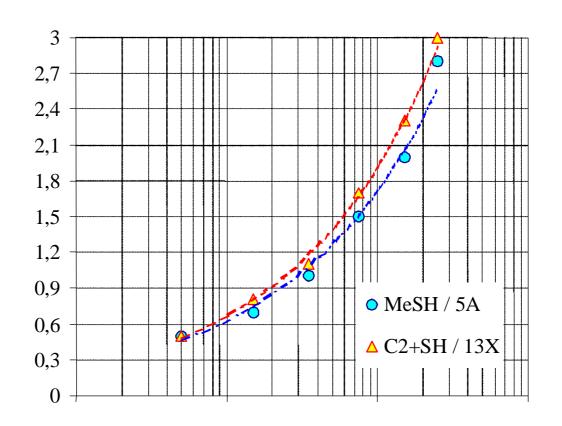
Adsorption



Adsorption



Isotherm RSH Adsorption

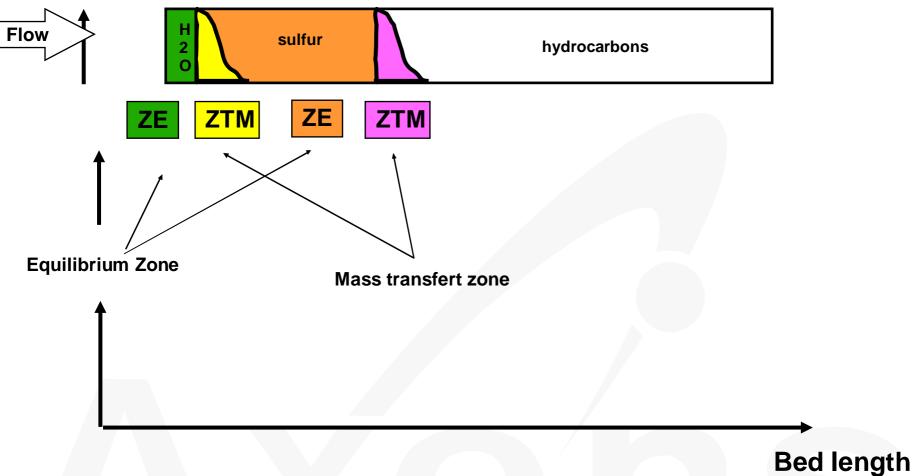


Inlet RSH (ppm)



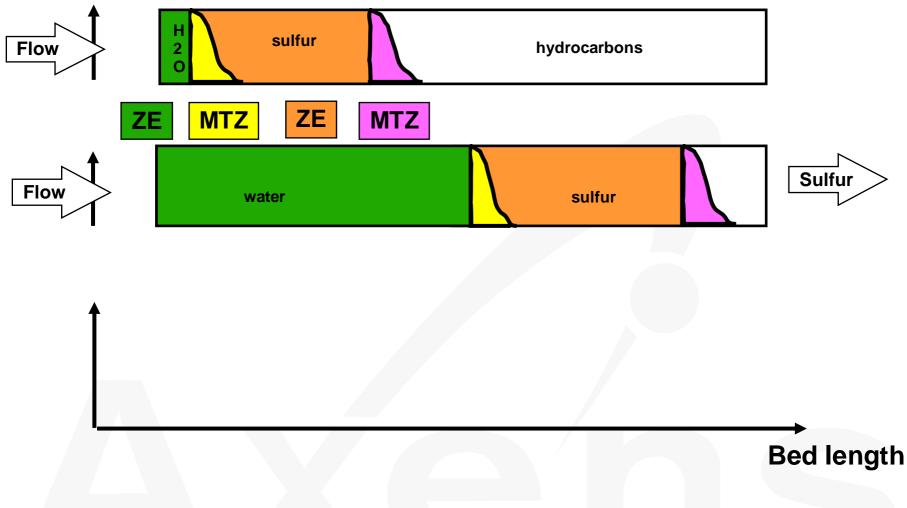
Co-adsorption



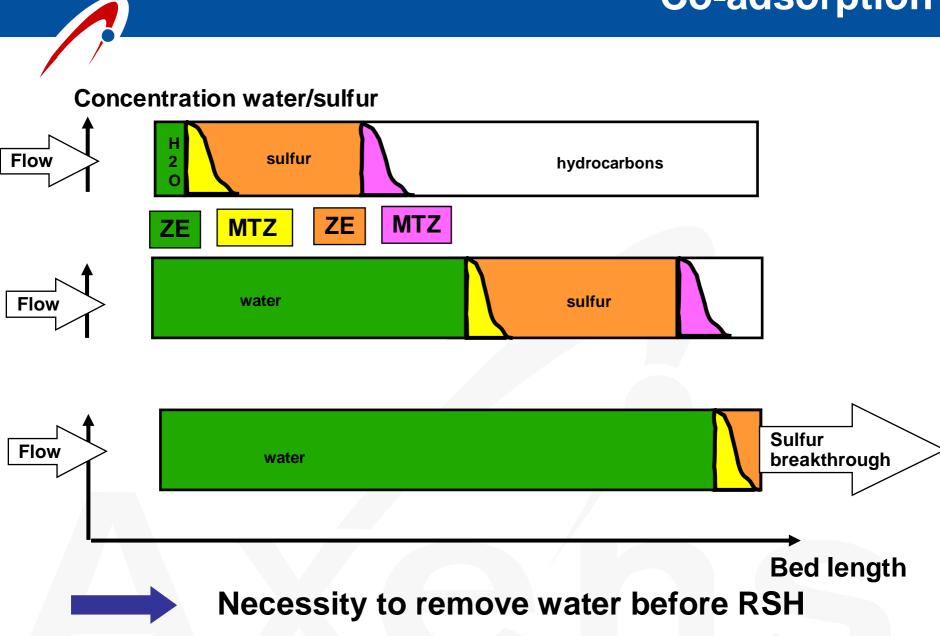


Co-adsorption



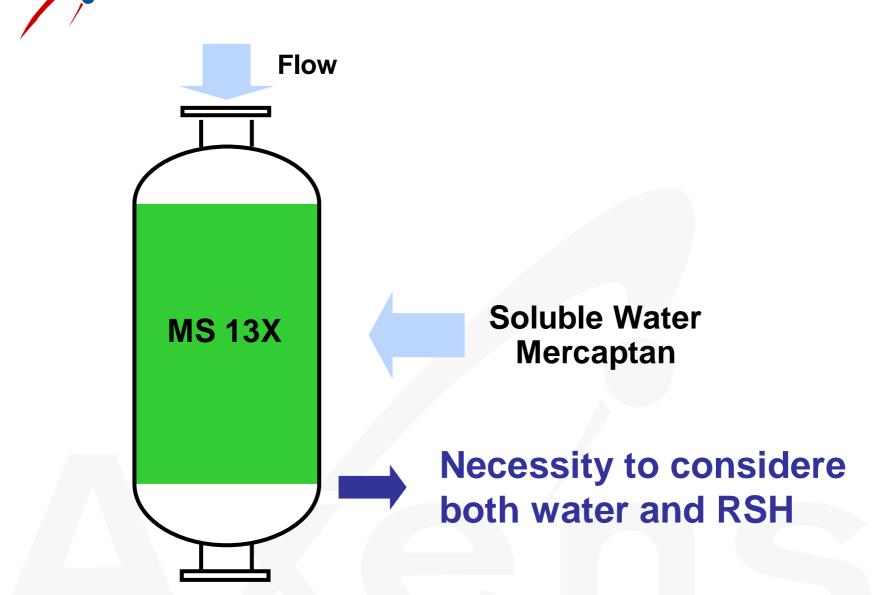


Co-adsorption

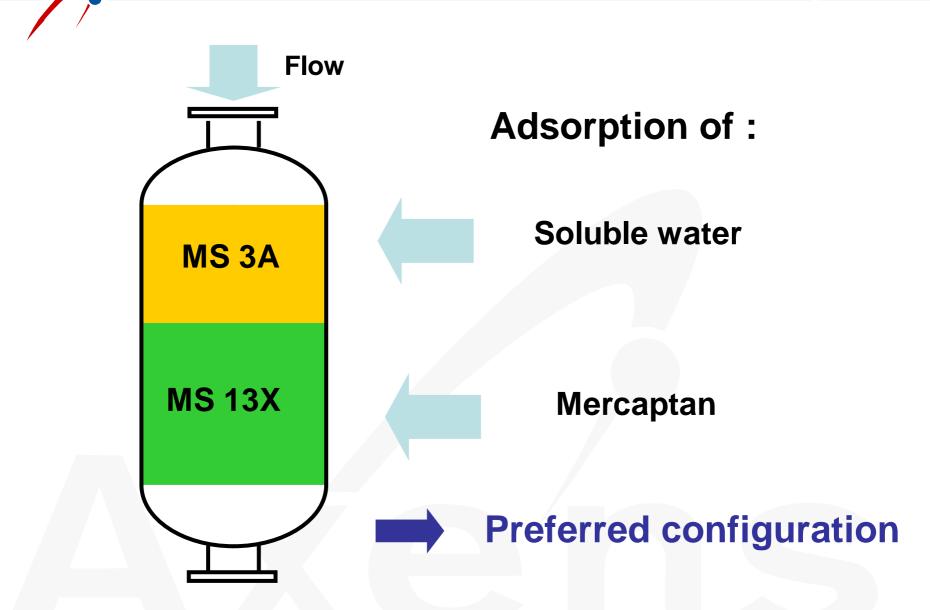


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Molecular Sieves for Removing Water and Mercaptans



Multibed Systems for Removing Water and Mercaptans

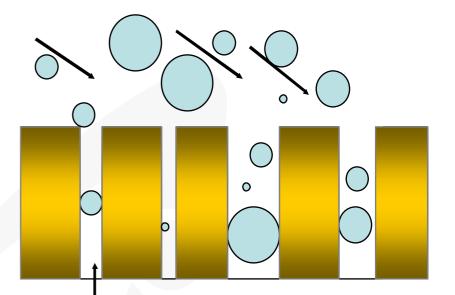


Molecular Sieve vs Activated Alumina

Molecular Sieves cation cation cation **Pore size** MS 3A : 3 angstroms **MS 4A : 4 angstroms MS 5A : 5 angstroms** MS 13X: 8 angstroms

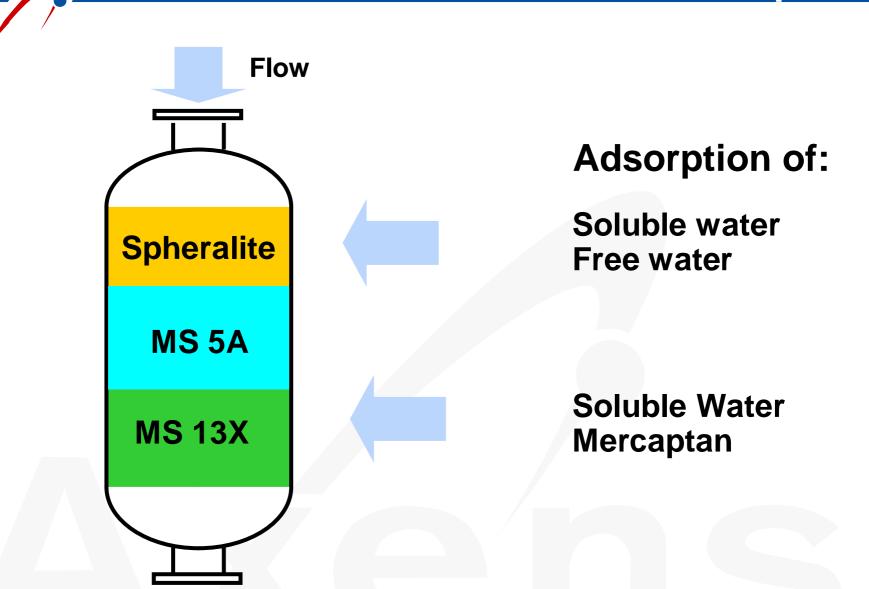
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Activated Aluminas



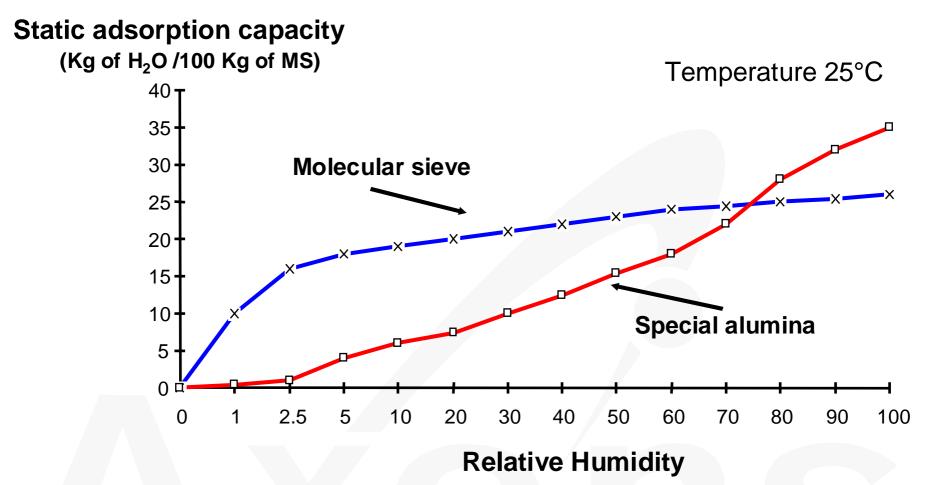
Pore size : 20- 60 angstroms

Multibed Systems for Removing Water and Mercaptans

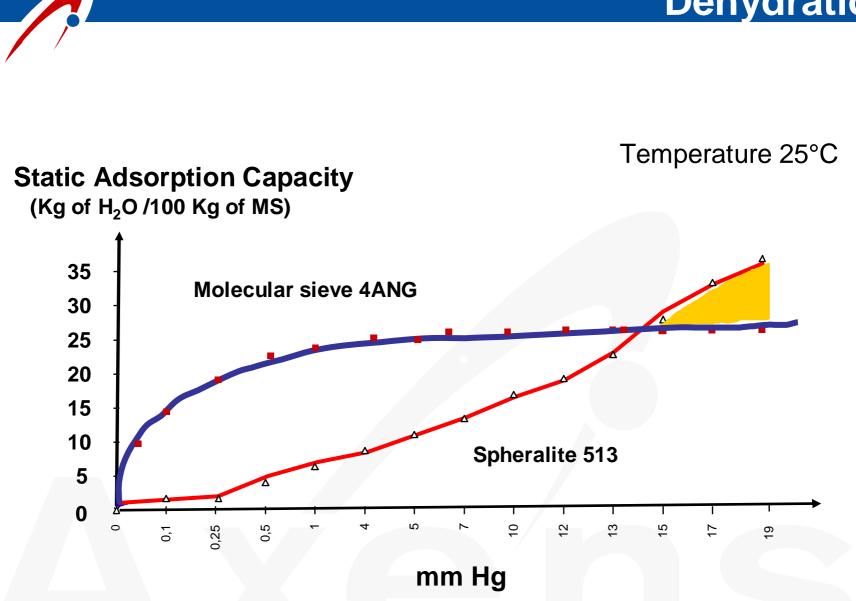


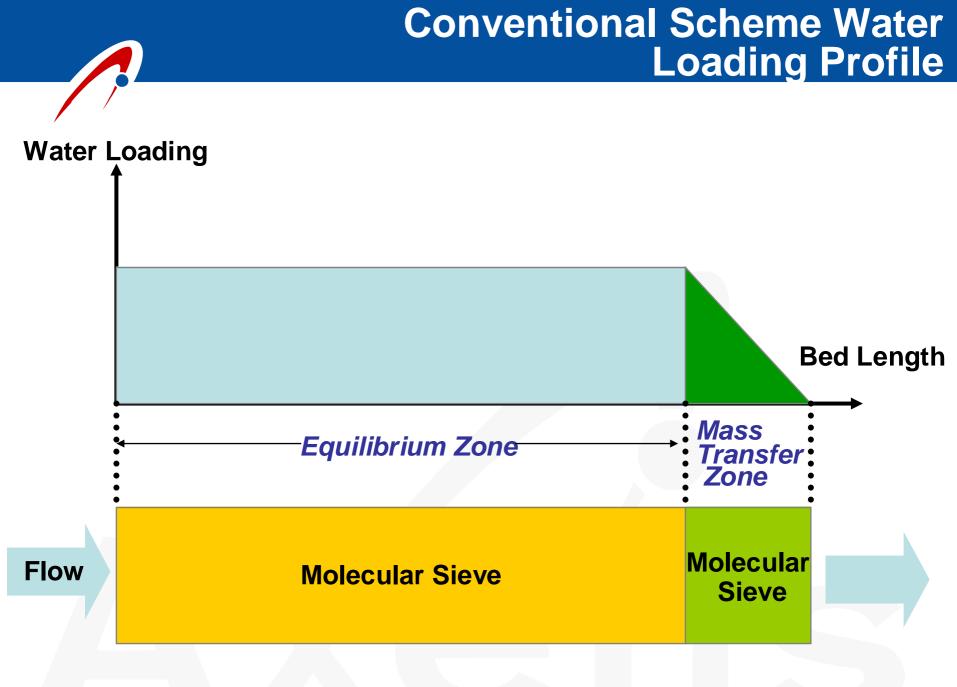
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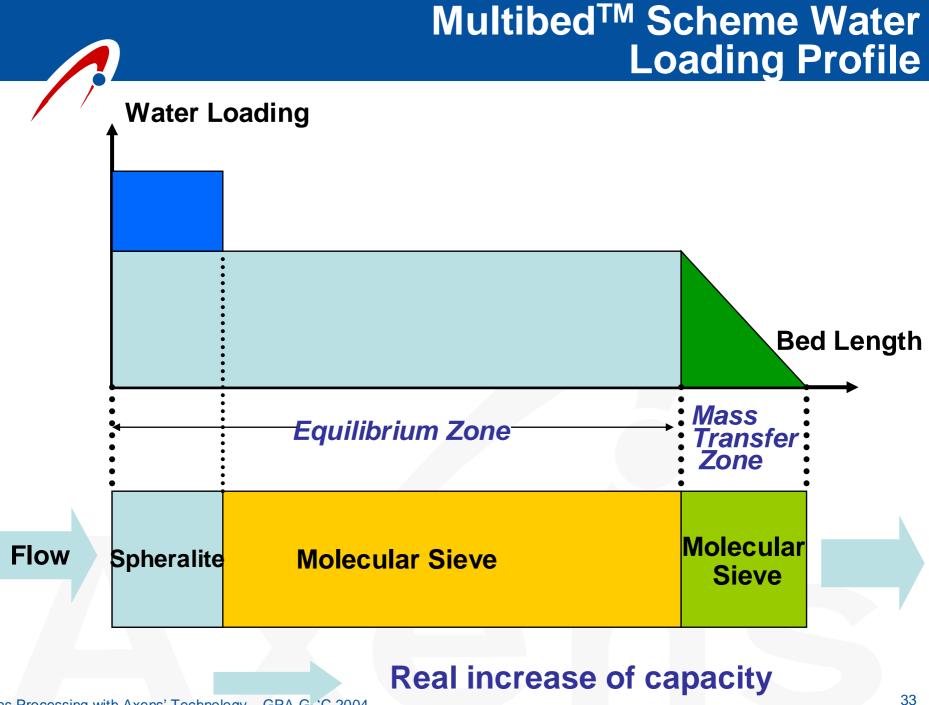
Isotherm Water Adsorption











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The Spheralite used in the upper section of the equilibrium zone does not increase the co-adsorption of hydrocarbons as compared to a molecular sieve system



- Axens' complete offer for gas treatment, together with Prosernat
- Proven technologies that allows optimizing project's economics and performances
 - H2S removal : Amine Treatment & Molecular Sieves
 - Drying unit : Glycol Treatment & Molecular Sieves



- Appropriate solution for any contaminant
- Efficient adsorbent technologies for drying and purification
- Wide experience in feedstock purification for petrochemicals, refining, natural gas and chemical industries
- On-going research and development