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COMPARISON OF THE RENEWABLE TRANSPORTATION FUELS, LIQUID HYDROGEN AND METHANOL, WITH GASOLINE—ENERGETIC AND ECONOMIC ASPECTS

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Abstract—In this paper, the renewable energy vectors liquid hydrogen (LH_2) and methanol generated from atmospheric CO_2 are compared with the conventional crude oil—gasoline system. Both renewable concepts, liquid hydrogen and methanol, lead to a drastic CO_2 reduction compared to the fossil-based system. The comparison between the LH_2 and methanol vector for the transport sector shows nearly the same fuel cost and energy efficiency but strong infrastructure advantages for methanol. © 1998 International Association for Hydrogen Energy

INTRODUCTION

For the introduction of LH₂ as fuel in the transportation sector a new infrastructure has to be erected with new techniques for hydrogen transportation, distribution and storage. If hydrogen is stored as methanol which is liquid at ambient temperature, it can be utilized directly with minor changes of the existing infrastructure today. Methanol is specially favoured for road transportation in combustion engines as a low emission fuel today, and for future application in fuel cell-powered vehicles as an onboard hydrogen storage medium. Besides other chemical storage systems, such as methylcyclohexane and ammonia, methanol is an interesting option for hydrogen storage and long distance energy transportation.

For the comparison of the renewable energy vectors hydroelectric power \rightarrow LH₂ and hydroelectric power \rightarrow methanol the established crude oil \rightarrow gasoline system is used as reference. For all three vectors the overall energy efficiency and the costs are analysed and discussed. The process design for renewable fuel generation (liquid hydrogen and methanol) is based on technologies available today to get reliable results.

DESCRIPTION OF VECTORS

The selection of the energy vectors was made according to a representative fuel supply in central European countries or Germany in particular. All relevant data for the cost calculations are given in Figs 1–3. The energetic data refers to the lower heating values (LHV) of the fuels.

The crude oil-gasoline vector

This vector comprises five elements: the crude oil production in one of the OPEC countries, the intercontinental tanker transportation, a refinery in Germany, the distribution of gasoline to the filling station and an intermediate passenger car (Fig. 1). The intercontinental crude oil transportation is realised with a conventional tanker. The travelling time for a round trip is 36 days for a distance of 17600 km (Middle East-Germany-Middle East). The refinery represents the mix of refineries in Germany in 1990 for which data on the energy consumption and costs have been available. The distribution of gasoline is realised by a typical tank truck. The reference car shows an average fuel consumption of 8.2 liters gasoline per 100 km. The cost of gasoline which is 1.50 DM l⁻¹ includes 0.98 DM mineral oil tax and 0.20 DM value added tax according to regulations applied in Germany.

The hydropower-liquid hydrogen vector

This vector was mainly taken from the feasibility study on the "Euro-Quebec Hydro-Hydrogen Pilot Project (EQHHPP)" published in 1991 [1]. Some modifications have been made in order to enable a consistent comparison of the energy vectors. A detailed description of production, transport and distribution of this vector is presented in Fig. 2. The intercontinental hydrogen transportation is realised with a specially designed LH₂ barge carrier with a capacity of 1036 t LH₂ and 15 round trips

Crude Oil - Gasoline Vector

OPEC Crude Oil

Crude oil price: 16.5 US\$/barrel

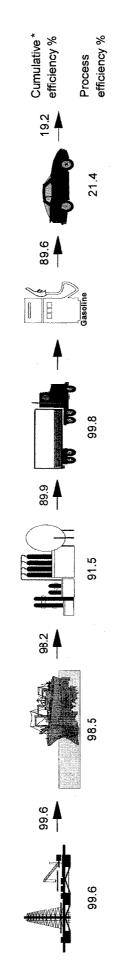
Refinery Germany

Reference plant: mix West Germany 1990 Gasoline price refinery: 387 DM/t

Type: medium size passenger car Fuel consumption: 8.2 1/100 km Fuel: unleaded gasoline Annual km: 15,000 Engine: 85kW

Gasoline Car

Fuel cost (incl. tax): 1.50 DM/1



Intercontinental Crude Oil Transport

Oil Tanker

Capacity: 250,000 tdw (tons dead weight)

Fuel: heavy fuel oil

Distance: 17,600 km

Travel time 36 days

Capital cost: 30,000 DM per day

Crude oil price Germany: 251 DM/t Operating cost: 18,000 DM per day

Filling station commission: 0.025 DM/I

(untaxed)

Fig. 1. Energetic efficiency and process data of the crude oil-gasoline vector.

Gasoline Distribution

Tank truck

Capacity: 35,000 1

Distance: 200 km

Fuel consumption: 35 l Diesel/100 km

Number of round trips: 500 per year

Towing vehicle: 175,000 DM/6 years Tank trailer: 175,000 DM/10 years Investment/depreciation time:

Total cost: 2.33 DM/km

Gasoline price filling station: 0.33 DM/l

Liquid Hydrogen Vector

Hydroelectric power station

Canada

Full load hours: 8,300 h/a Capacity: 100 MW_e

Electricity cost: 0.05 DM/kWh_p

Intercontinental LH2-Transport

Fuel: heavy fuel oil LH2-barge carrier

Fuel consumption: 2.2 kg LH₂/100 km

Type: medium size passenger car

LH2-Car

Engine: 85 kW Otto

Evaporation losses: 1.7 % per day

Tank volume: 130 l

Average efficiency: 17.8 %

Annual km: 15,000

Filling losses: 5 %

Capacity (volume): $5 \times 3,000 \text{ m}^3_{\text{net}}$

Evaporation losses: 1.4 %

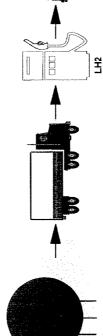
Distance: 11,500 km

Depreciation time: 15 years **Investment:**

Barge carrier: 92 Mill DM

LH, tanks: vacuum isolated, mobile Capacity (mass): 1,036 t LH₂

LH₂-tanks: 48 Mill DM



LH, Distribution

LH, tank truck

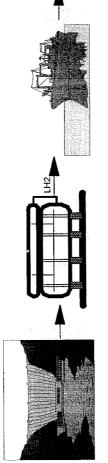
Capacity: 37m3LH, nitrogen casing

Investment/depreciation time:

Towing vecicle: 175,000 DM/6 years Tank trailer: 420,000 DM/20 years

Filling station commission: 0.14 DM/kg LH₂

Fig. 2. Process data of the liquid hydrogen vector.



LH, Production

Electrolysis:

Energy consumption: 4.2 kW_e/Nm³H₂ Advanced alkaline elektrolysis

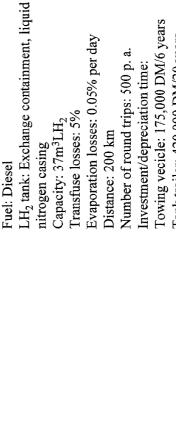
Depreciation time: 15 years

Investment: 1,400 DM/kW,

jauefaction:

Energy consumption: 11 kWhe/kgH2 Depreciation time: 15 years

investment: 1,000 DM/kW_{LHV} H₂



Methanol Vector

Hydroelectric power station

Canada

Full load hours: 8,300 h/a Capacity: 100 MW_e

Electricity cost: 0.05 DM/kWhg

Intercontinental Methanol Transport

Product Carrier: Tanker Capacity: 24,200 tdw

Fuel: heavy fuel oil

Distance: 11,500 km

Depreciation time: 15 years Fravel time: 23 days

Investment: 100 Mill DM

Operating Cost: 17,000 DM per day Fuel Cost: 180 DM/t

(M85): 0.015 DM/I

Cost of blending Methanol and Gasoline

M85-Car

flexible fuel vehicle (M0-M85) Type: medium size passenger car

Engine: 85 kW Otto

Fuel consumption: 14 l M85/100 km Annual km: 15,000

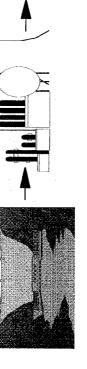
Average efficiency: 23.5%



gasoline blend

Tank truck,

Filling station commission: 0.02 DM/I parameters like gasoline distribution



Methanol production

Capacity: 170 t per day Efficiency: 38.1 % CO, absorber, air flow rate: 8,000 m³/s

CO, absorption rate: 65 %

electrodialytic production: 425 t NaOH per day Electrolysis: alkaline pressure electrolysis,

energy consumption: 4.2 kWh_e/Nm³H,

Depreciation time: 15 years

investment:

CO, absorber/enrichment: 19 Mill DM Electrodialysis: 26 Mill DM

Electrolysis: 88 Mill DM

Synthesis: 30 Mill DM

Fig. 3. Process data of the methanol vector.

per year. In Germany LH₂ is transferred to mobile LH₂ tanks which are distributed to filling stations by tank trucks. H₂ evaporation losses at the gas station during the refill of the car tank are fed to a natural gas pipeline (revenues, Fig. 5). The car is based on the reference vehicle for the gasoline vector but requires specific LH₂ equipment. Its average fuel consumption is considered to be 2.2 kg LH₂ per 100 km. Including energetic and material losses the car's efficiency will be 17.9%.

The hydropower-methanol vector

Generation of methanol. A new process has been designed for methanol generation from atmospheric CO₂ [2]. The recovery of higher quantities of CO₂ from the air requires new highly efficient technologies with a low specific energy consumption, due to the low concentration of CO₂ in the air. During the absorption process of CO₂ from the air, Na₂CO₃ is formed from the NaOH scrubbing solution (Fig. 4). The recovery of CO₂ from the carbonate solution produced in the absorption column occurs by acidifying the spent solution with sulphuric acid. For the regeneration of the caustic scrubbing liquid and the acid solution an electrodialytic process with bipolar membranes is used. The methanol production is carried out by a catalytic conversion of CO2 and electrolytically generated hydrogen over a Cu/ZnO catalyst. All energy process data used in this study are based on experimental results and represent a technology available today with an overall efficiency of 38.1% related to the LHV of methanol. CO2 can also be taken from high concentration sources, such as fossil-fired power plants, cement production, etc. If a carbon-based fuel is produced from atmospheric CO₂, the energy carrier will be CO₂-neutral due to the closed carbon cycle without any greenhouse gas emissions. Furthermore, the use of atmospheric CO2 allows methanol synthesis at any location (with renewable energy resources) without the costs for CO₂ transport and storage.

Methanol vector description. For the methanol vector it was assumed that not pure methanol but a blend of 85% methanol and 15% gasoline will be used in cars. This is simply due to the fact that presently only flexible fuel cars are available, as no country with a dense grid of methanol filling stations will be available in the near future. But it is worth noting here that pure methanol cars have already been built and successfully operated in a number of large demonstration projects under commercial conditions, and pure methanol engines show higher efficiencies than gasoline engines. For a better comparison it was assumed that the same hydropower plant in Canada is applied to provide the primary energy input as for the LH₂ vector (Fig. 3). The hydrogen is again produced by an advanced electrolysis. The other components required for methanol production are CO₂ absorption, electrodialysis and methanol synthesis (total investment: 163 million DM). In a second step, methanol is transported from Canada to Germany with a conventional product tanker with a capacity of 24 000 t. For the transportation of the annual methanol produced only

two to three round trips are required. Therefore the costs are calculated on presently applied freight rates. This is different to the EQHHPP-LH2 vector where the size of the barge carrier was selected specially for this project and for continuous operation (15 round trips per year). In Germany, methanol is blended with gasoline to M85 (85 vol.% methanol, 15% gasoline) at a refinery. The cost and energy requirements for providing gasoline are taken from the results of the crude oil-gasoline vector described above. For transporting M85 from the refinery to the filling station, the same reference tank truck is used as for the gasoline vector. The car is again an intermediate passenger car that corresponds to flexible fuel cars produced by different automobile manufacturers, which usually show an increase in efficiency of 10% at M85 operation compared to pure gasoline operation. The cost of adapting the fuel system to M85 in a series production is moderate, in the order of 1000 DM.

COMPARISON OF ENERGY EFFICIENCIES

The crude oil-gasoline vector reaches an overall efficiency for the first four vector elements of almost 90%. This excellent value is due to the marginal energy requirements for crude oil production (0.4%) and the high efficiency of tanker transportation (1.5%), as the fuel consumption of the engines is only 3600 t heavy oil for a round trip, when 250000 t crude oil is transported. Only the refinery requires a higher energy equivalent (about 8.5%) to convert crude oil into products. The distribution of gasoline is again very efficient as the tank truck requires only 701 diesel for one round trip but transports 350001 gasoline. The only crucial element of the gasoline vector is the car with an average efficiency of only 21.4%. Thus, the energy efficiency of the entire gasoline vector reaches 19.2%.

It is obvious that the production efficiency of the renewable energy carriers LH₂ and methanol is lower than gasoline production efficiency, due to the conversion of the primary energy hydroelectric power to a completely different energy carrier. The overall efficiency of the LH₂ production is 57.7% related to the hydroelectricity input and the LHV of hydrogen (Fig. 5). The transport capacity of the barge carrier is 1036 t LH₂, but it consumes 800 t heavy fuel oil during a round trip which is roughly one fourth of the consumption of the crude oil tanker and its transport capacity is 250 000 tons of crude oil. The average efficiency of LH₂ distribution is primarily determined by the losses occurring when filling the mobile tanks. Five per cent of LH2 evaporates and has to be liquefied again. The efficiency of the LII2 car is determined by: filling losses, evaporation losses and the efficiency of the car in motion which has the same efficiency as the gasoline car (21.4%). The car's annual average efficiency was calculated to 17.9% assuming a mileage of 15 000 km and 52 tank refills per year. The total efficiency of the entire LH2 vector related to the overall primary energy input is 8.7%.

As shown in Fig. 6, the efficiency of the methanol

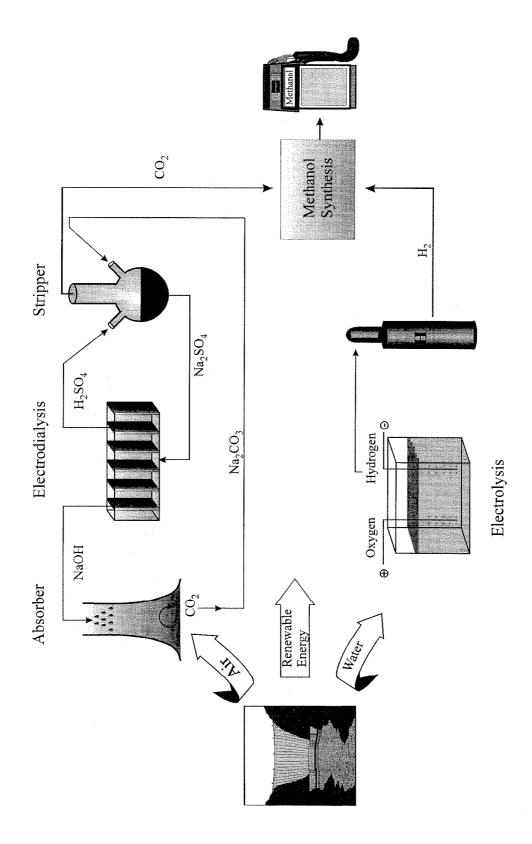


Fig. 4. Schematic process flow sheet of methanol generation from atmospheric carbon dioxide.

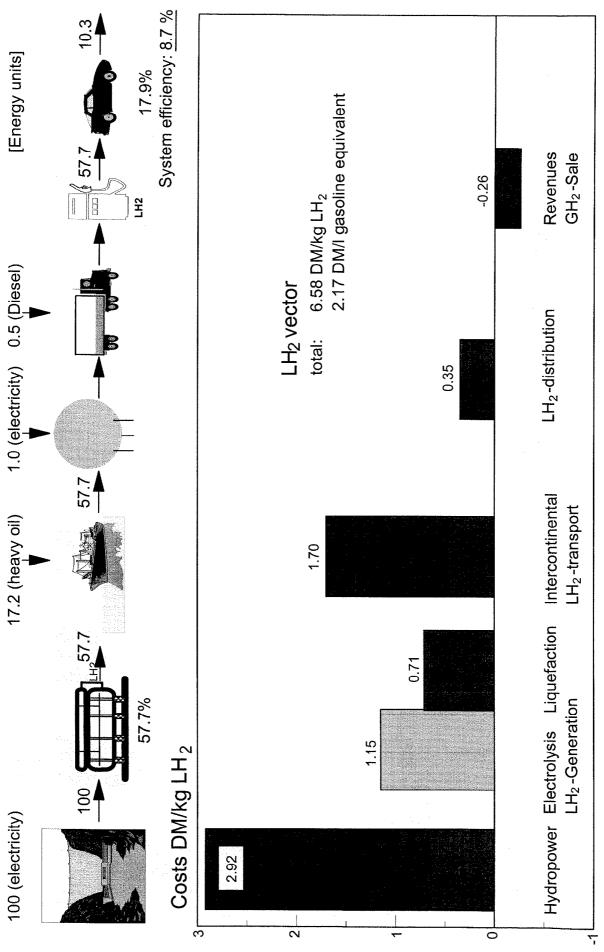


Fig. 5. Energetic efficiency of the liquid hydrogen vector and LH_2 cost calculation.

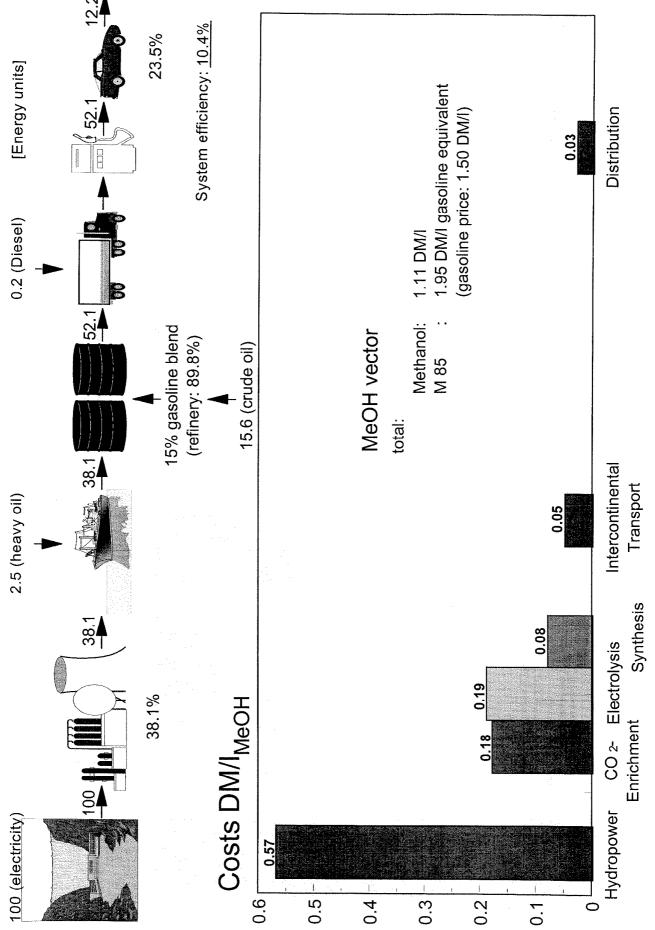


Fig. 6. Energetic efficiency of the methanol vector and methanol cost calculation.

vector strongly depends on the production process which amounts to 38.1%. The efficiency of methanol transportation is much higher compared to LH₂, as the capacity of the tanker is higher and no evaporation and filling losses occur. As methanol is then blended with gasoline to M85, efficiencies of distribution and of the methanol car (23.5%) are related to M85. Because M85 consists of 15% gasoline and the efficiency of gasoline production is rather high (89.9%, Fig. 1), this positively influences the overall efficiency of the M85 vector, which will be 10.4%. For a pure methanol vector the efficiency of fuel production and distribution would be a little lower. But on the other hand the efficiency of a pure methanol car could be in the range of 25%. The overall efficiency of the pure methanol (M100) vector therefore would be in the order of 9.3%.

COMPARISON OF COSTS

The costs are mainly determined by the energy input and the capital cost of the production plants. For both renewable energy vectors it was assumed that hydroelectricity is available at 0.05 DM kWh⁻¹ and 8300 h y⁻¹. Capital costs are calculated on a real interest rate of 8% and depreciation periods (usually 15 years) correspond to about 50% of the expected technical lifetime of the plants (see Figs 5 and 6). These parameters have been selected according to practical standards in the energy industry. On these assumptions it follows that LH₂ provided to the consumer costs 6.58 DM kg⁻¹ (untaxed) which corresponds to 2.17 DM l⁻¹ gasoline equivalent. This value is calculated for an annual base taking into account evaporation losses that occur even when the car is not in use. As Fig. 5 shows, about 25% of the LH₂ cost is determined by the overseas transportation, which could be considerably reduced if transportation capacities increase in the future.

The cost of producing and providing methanol to the consumer can be calculated as 1.11 DM l⁻¹ methanol (untaxed) which is 2.30 DM related to the energy content of one liter of gasoline. Taking into account the 10% higher operation efficiency of M85 cars compared to gasoline cars, the value has to be reduced correspondingly to 2.08 DM l⁻¹ gasoline equivalent. Assuming gasoline costs of 1.50 DM l⁻¹ for blending, the price of M85 is 1.95 DM l⁻¹ gasoline equivalent.

CONCLUSIONS

Comparing the untaxed gasoline equivalent costs of LH₂ (2.17 DM) with that of the methanol in the M85 vector (2.08 DM) or the resulting cost of M85 (1.95 DM), in the frame of calculation precision the costs are almost equal. The higher costs of the renewable fuels, liquid hydrogen and methanol, will become more comparable to the fossil fuel costs, if environment damage costs and CO₂ taxes are included in calculations in the future.

The investment and operation costs of the methanol synthesis, based on CO₂ and H₂, are equivalent to a conventional plant but without the steam reformer costs, which are more than 50% of the total investment. The costs for methanol production from CO₂ and the hydrogen generation costs are reliable but the CO₂ enrichment costs are somewhat uncertain due to the non-existence of such plants today. The CO₂ enrichment costs are only 16% while the energy costs are more than 50% of the overall methanol costs, therefore higher CO₂ enrichment costs would not lead to a drastically increased methanol price.

It is also worth noting that fuel costs represent only one aspect for the consumer. Of higher importance is the energy service cost, i.e. driving a distance of 100 km with a passenger car. To determine this cost one has to consider the investment of the car, the decrease in value, operation and maintenance costs, tax and insurance. Here the methanol has clear advantages compared to LH₂ as only minor modifications of the car have to be done while LH₂ cars require complex and costly equipment. Considering the advantage of the existing infrastructure for liquid fuels and the cost disadvantage for the liquid hydrogen powered car (minimum 20% additional car investment costs for the LH₂ equipment), the methanol vector shows a considerable advantage over the LH₂ vector, even if the fuel costs are almost equal.

The comparison of the LH₂ with the methanol vector shows an energetic advantage for the methanol concept. This is mainly due to the evaporation losses in the LH₂ concept and due to higher efficiency of the methanol car. Even if LH₂ evaporation losses could be reduced near to zero, this would not change the result in principle. The efficiency of the methanol vector was calculated to be 38.1%. With an advanced technology, the process efficiency can be improved to 44% [2]. A further efficiency improvement will be possible with high temperature processes for CO₂ enrichment and synthesis [3].

Efficiencies will be even higher if CO₂ is taken from concentrated emissions (e.g. fossil-fired power plants or cement production). Cost calculations based on these assumptions lead to 30% lower methanol prices compared to the atmospheric CO₂ path. Further cost reductions are feasible by using other carbon sources as feedstocks such as biomass or waste for synthesis gas production via gasification and subsequent methanol generation. Synfuels from renewable hydrogen and carbonaceous resources are the link to conventional fuels, which allows the continuous transition from fossil to renewable energy carriers.

For both concepts, LH₂ and methanol, an increase of the car efficiency is possible in the future. A pure methanol (M100) car with an internal combustion engine has an improved efficiency of 30% compared to a gasoline car. A further improvement can be achieved by fuel cell-powered cars. Thus, the overall energy efficiency of both renewable vectors will approach the crude oil vector efficiency.

The main advantage of the renewable concepts is related to the greenhouse gas emission reduction poten-

tial. Both renewable energy vectors, LH₂ and methanol, can contribute to a substantial reduction of CO₂ emissions in the transport sector.

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