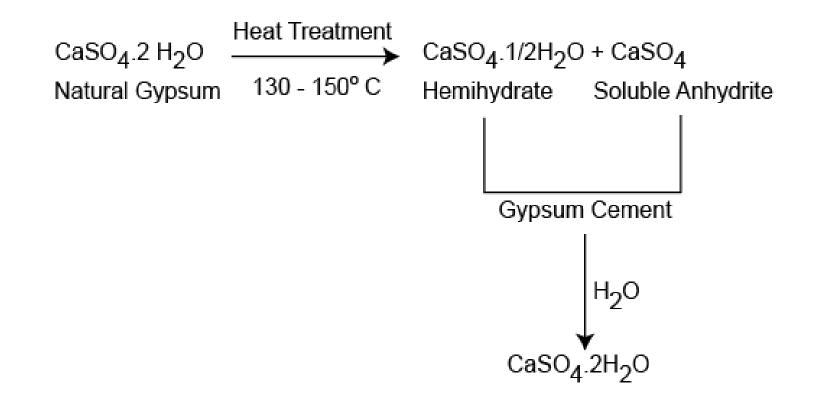


HYDRAULIC CEMENTS

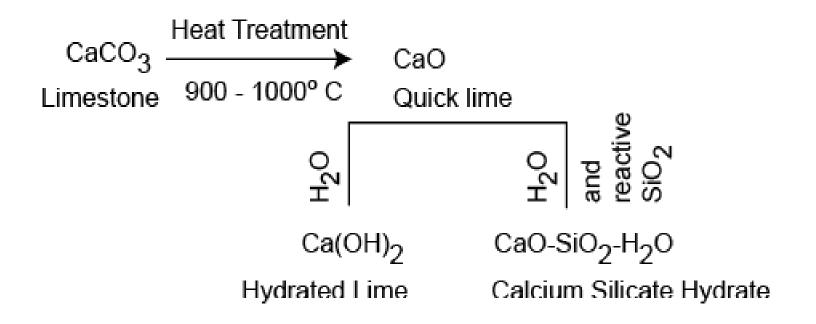


Read Chapter 6





Lime Cement

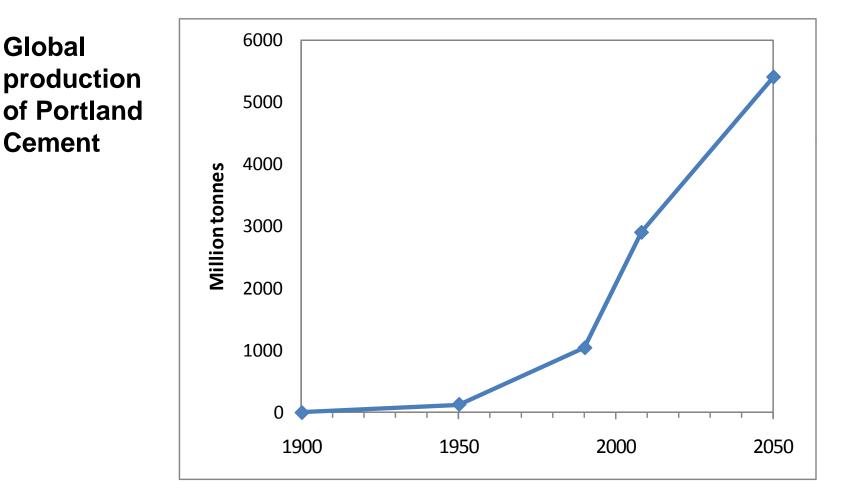


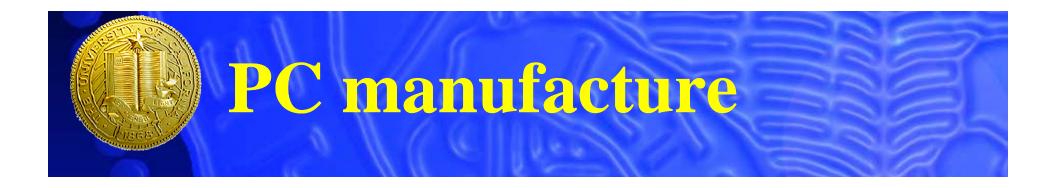


A hydraulic cement capable of setting, hardening and remaining stable under water.

It consists essentially of hydraulic **calcium silicates**, usually containing calcium sulfate.

Portland Cement

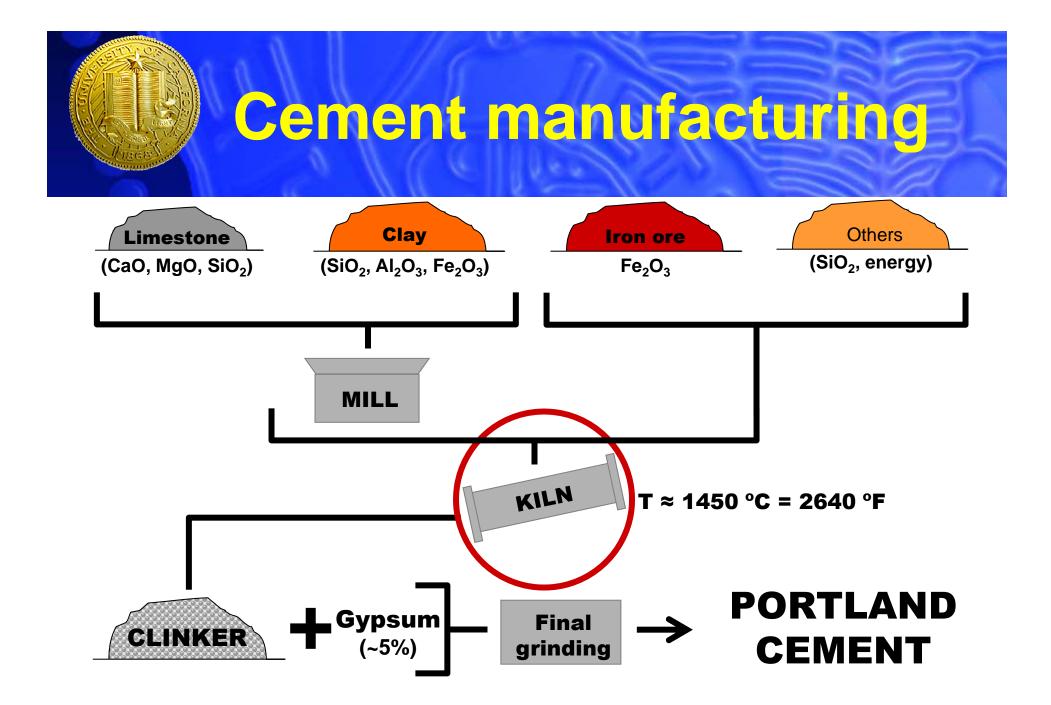




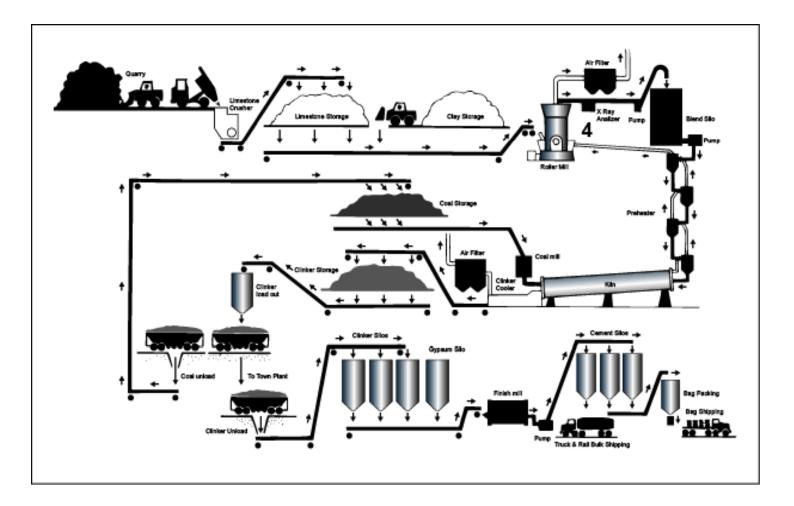
Raw Materials:

2/3 calcareous materials (lime bearing) - limestone

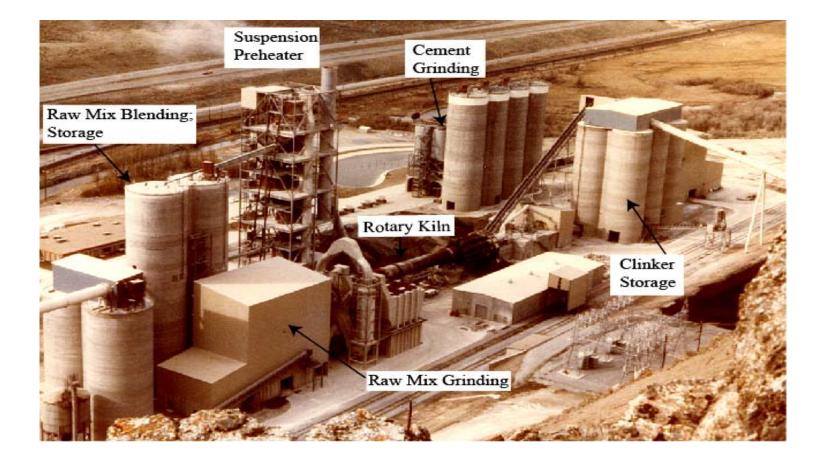
1/3 argillaceous materials (silica, alumina, iron) - clay



Manufacturing Process

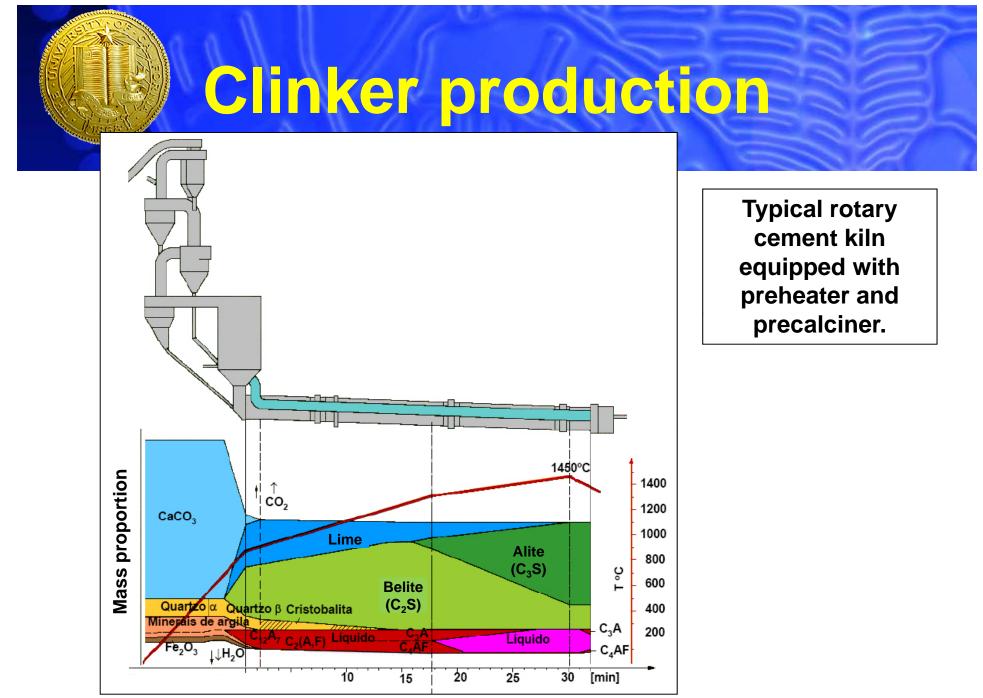


Cement Factory



Cement Factory



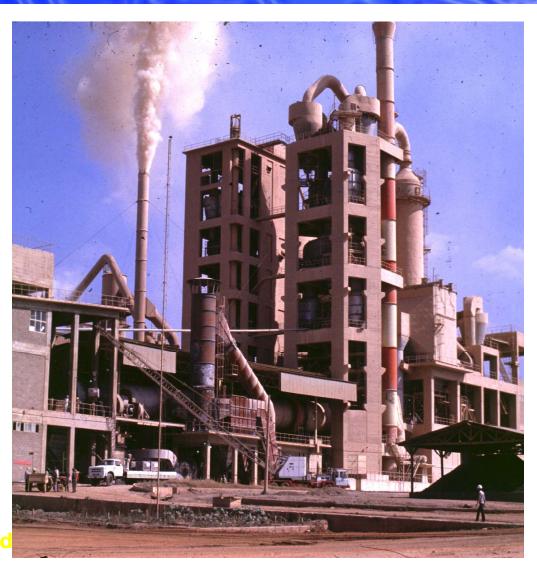


Clinker composition

Raw materials	Compounds	Abbreviation
Limestone	3CaO.SiO ₂ —	→ C ₃ S
$(CaO + SiO_2 + Fe_2O_3 + CO_2)$	2CaO.SiO ₂ —	→ C ₂ S
Clay (SiO ₂ + Al ₂ O ₃	3CaO.Al ₂ O ₃	→ C ₃ A
$+ Fe_2O_3 + H_2O)$	4CaO.Al ₂ O ₃ .Fe ₂ O	$O_3 \longrightarrow C_4 AF$
Other (e.g. Rice husk: SiO ₂)		
•••		

Production Process

Production of cement is responsible for ~8% CO₂ emissions in the World



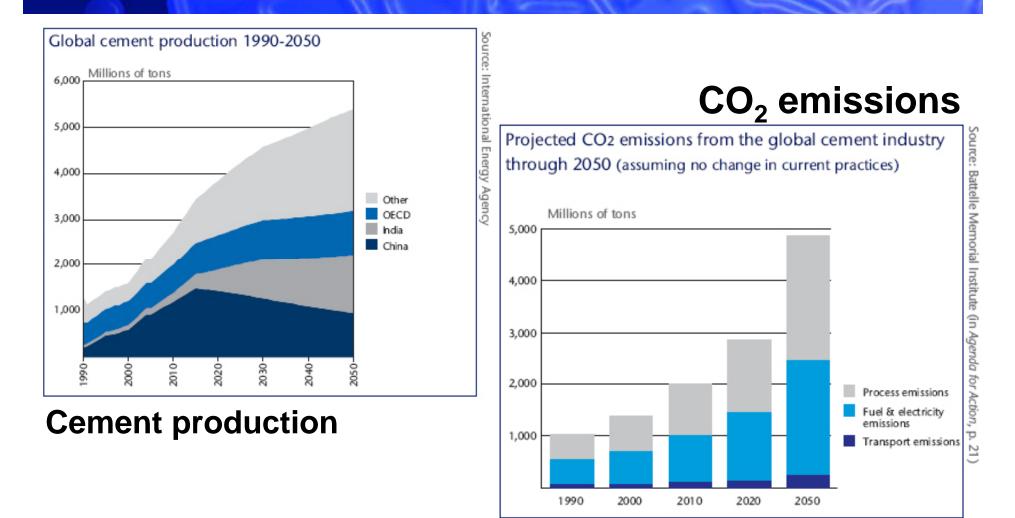
Clinker production: CO₂ emissions

Calcination reaction

 $CaCO_3 \rightarrow CaO + CO_2$ 1kg \rightarrow 0,56kg + 0.44kg

- For 1000kg of $CaCO_3$:
 - 440 kg CO₂ (due to the chemical reaction only)
 - 560kg CaO
 - Energy consumed by the reaction= 3,16 GJ per ton. CaO
 - Coal \rightarrow ~93kg CO₂/GJ
 - 560kg CaO \rightarrow 1,77 GJ \rightarrow ~ 165kg CO₂
 - Total = \sim 605 kg CO₂ (per ton. CaCO₃ calcined)
- About <u>815kg CO₂</u> are generated for every <u>1000kg of clinker</u> produced
 - Calcination (chemical reaction) = ~52%, Energy use = ~48%
- **CE 165: Concrete Materials and Concrete Construction**

Cement production: CO₂ emissions



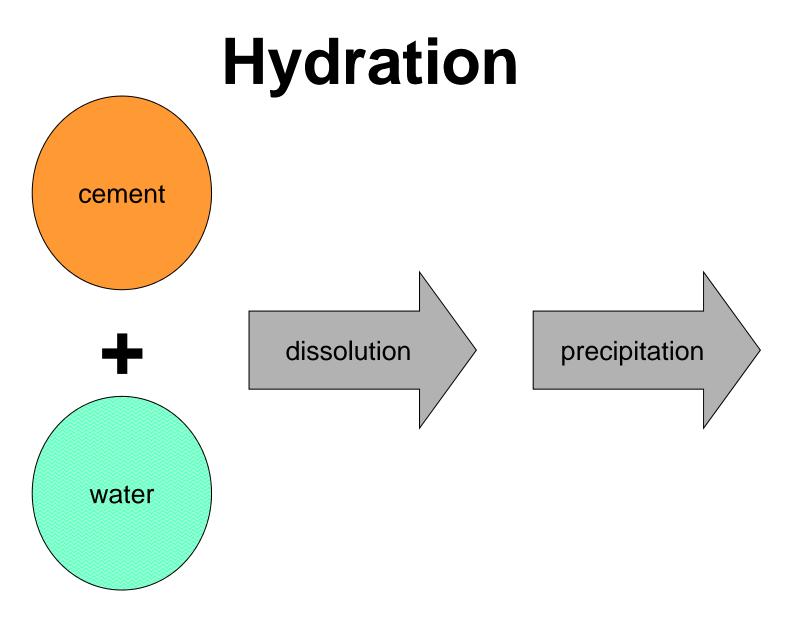


Cement notation:

С	CaO
S	SiO2
Α	Al2O3
\mathbf{F}	Fe2O3
Η	H2O

Cement Composites

 $C_3 S = 3CaO.SiO2$ $C_2 S = 2CaO.SiO2$ $C_3 A = 3CaO.A12O3$ $C_4 AF = 4CaO.A12O3.Fe2O3$



Courtesy: Prof. Karen Scivener, EPFL

CHEMICAL REACTIONS

Important!!!

2C3S + 6H --> C3S2H3 + 3CH + 120 cal/g 2C2S + 4H --> C3S2H3 + CH + 62 cal/g C3A + CSH2 --> Ettringite + 300 cal/g

Main Components of Portland Cement

Main Components of PC

	amount	notes
C3S	50%	very reactive compound, high heat of
		hydration, high early strength
C2S	25%	low heat of hydration, slow reaction
C3A	10%	problems with sulfate attack, high heat of
		hydration
C4AF	10%	-
gypsum	5%	used to control the set of cement

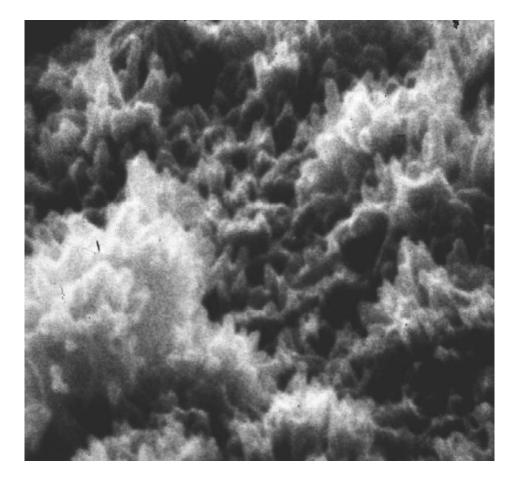
SOLIDS IN CEMENT PASTE

Calcium Silicate Hydrate

Notation: C-S-H C/S Ratio: 1.5 to 2.0 Main Characteristics:

High Surface (100 to 700 m2/g) ----> High Van der Walls Force ----> Strength. Volume: 50% to 60%





SOLIDS IN CEMENT PASTE

Calcium Hydroxide (portlandite)

Ca(OH)2 Volume: 20% to 25% Low Van der Walls force, problems with durability and strength





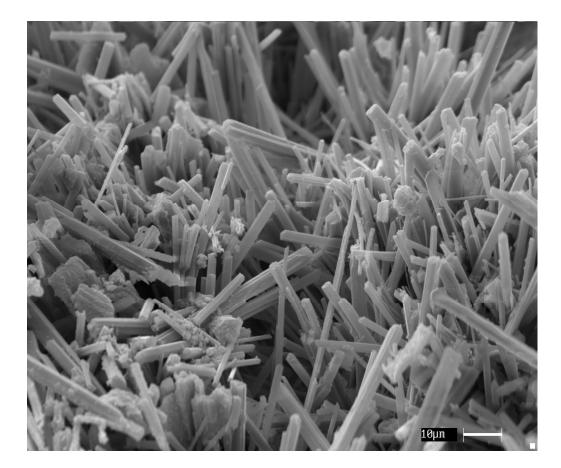


SOLIDS IN CEMENT PASTE

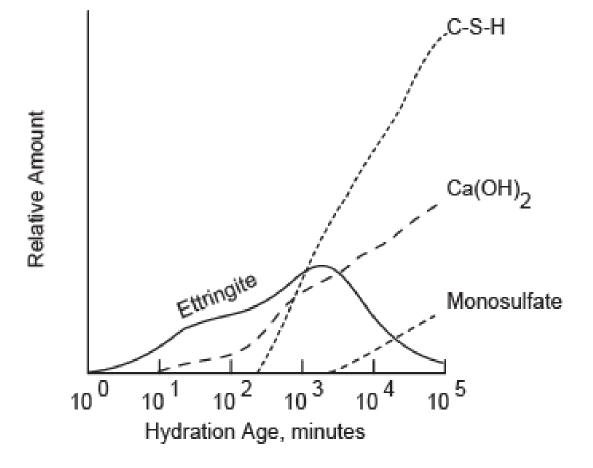
Calcium Sulfoaluminate Hydrates

Volume: 15% to 20% first: ettringite after: monosulfate hydrated





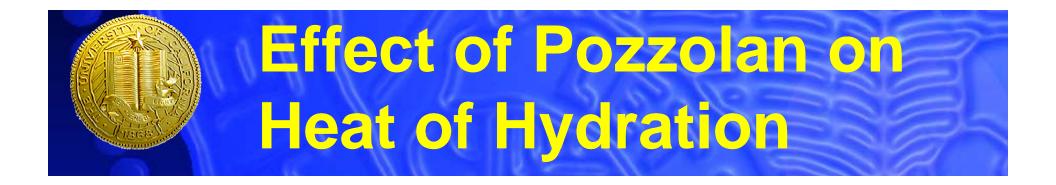


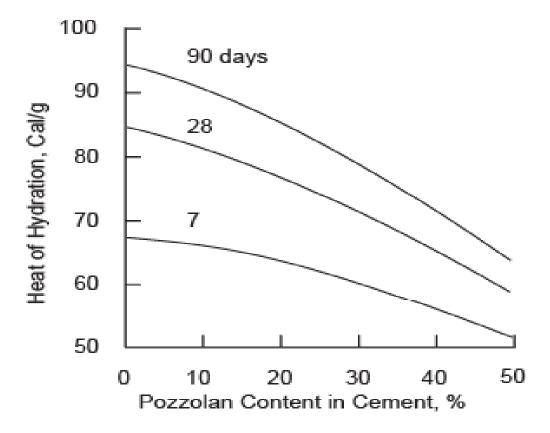


CE 165: Concrete Materials and Concrete Construction

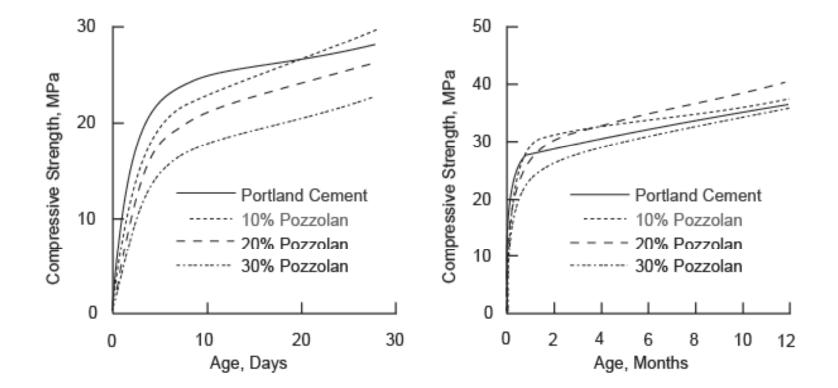
Main Components of PC

	Ν	Iain Components of PCMake a note		
	amount	notes		
C3S	50%	very reactive compound, high heat of		
		hydration, high early strength		
C2S	25%	low heat of hydration, slow reaction		
C3A	10%	problems with sulfate attack, high heat of		
		hydration		
C4AF	10%			
gypsum	5%	used to control the set of cement		





Compressive Strength



Setting and Hardening

Stiffening: loss of consistency by the plastic cement paste and it is associated with the slump loss phenomena in concrete. Setting: Solidification of the plastic cement paste

- Initial Set: Beginning of solidification (point in time when the paste has become unworkable) (>45 min.)
- Final Set: Final solidification (< 375 min.)



Ordinary P.C (England) = Normal P.C. (USA) = Type I ASTM (general purpose PC).



		Compound composition range (%)			
ASTM type	General description	C ₃ S	C2S	C ₃ A	C4AF
I.	General purpose	45-55	20-30	8-12	6-10
II	General purpose with moderate sulfate resistance and moderate heat of hydration	40-50	25-35	5-7	6-10
III	High early strength	50-65	15-25	8-14	6-10
V	Sulfate resistant	40-50	25-35	0-4	10-20

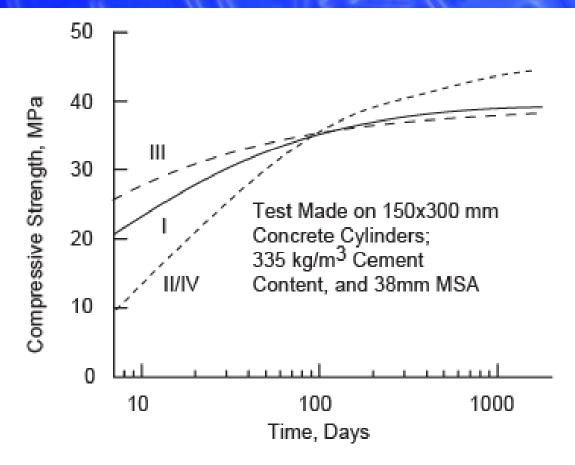
ASTM also has Types I-A, II-A, III-A -- cements with air entrainment

ASTM Portland Cements

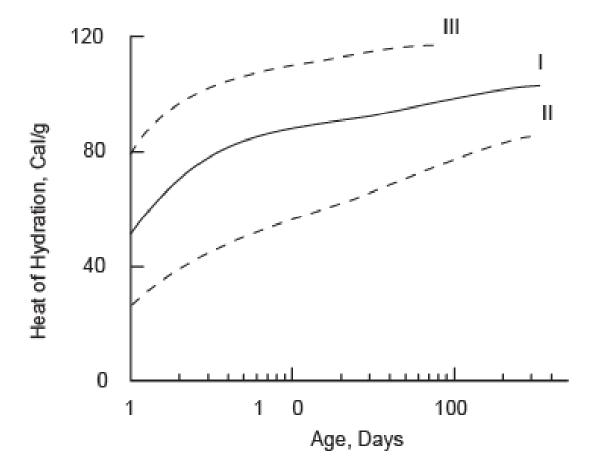
Type I General Purpose

- Type II moderate heat of hydration and sulfate resistance (C3A < 8%): general construction, sea water, mass concrete
- Type III high early strength (C3A < 15%) : emergency repairs, precast, winter construction.
- Type IV low heat (C3S < 35%, C3A < 7%, C2S > 40%) : mass concrete
- Type V sulfate resistant (C3A < 5%): sulfate in soil, sewers.

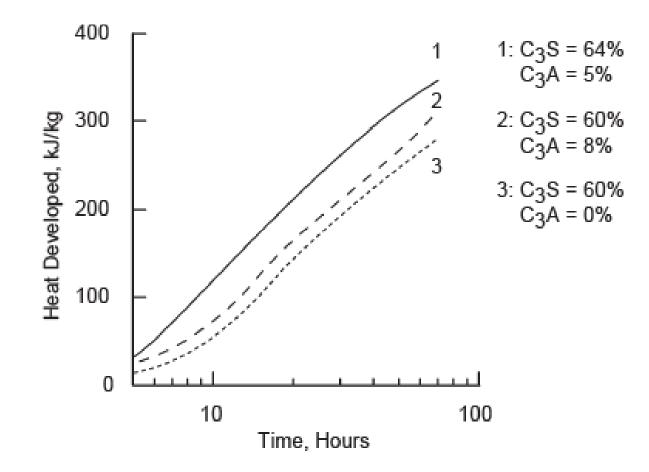
Strength Evolution



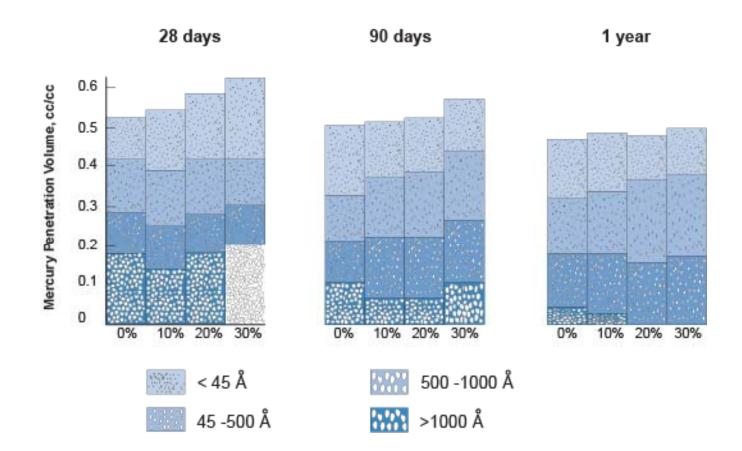
Heat Evolution







Porosity evolution



Cement Requirements

Requirement specified by ASTM C 150	Type I	Type II	Type III	Type V
Fineness:	280	280	None	280
minimum (m ² Ikg)				
Soundness:	0.8	0.8	0.8	0.8
maximum, autoclave expansion (%)				
Time of setting				
Time of setting Initial set	45	45	45	45
	43	45	45	43
minimum (min)	275	275	275	275
Final set	375	375	375	375
maximum (min)				
Compressive strength:				
		1	ninimum [MPa	a]
1 day in moist air	None	None	12.4	None
-				
1 day moist air + 2 days	12.4	10.3	24.1	8.3
water				
1 day moist air + 6 days water	19.3	17.2a	None	15.2

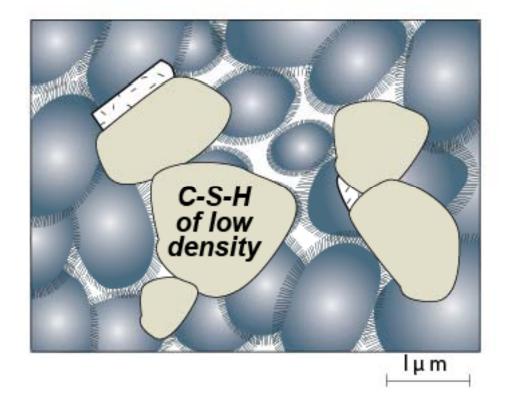
Other hydraulic cements

a) Blended P.Cb) Modified P.C.c) Non-calcium silicate cements

Blended PC

Type I-P P stands for pozzolan. It contains 25 to 30% of fly ash. It has low heat of hydration, develops strength over time.
Type I-S S stands for slag. It contains 50 to 60% of Blast-Furnace Slag.

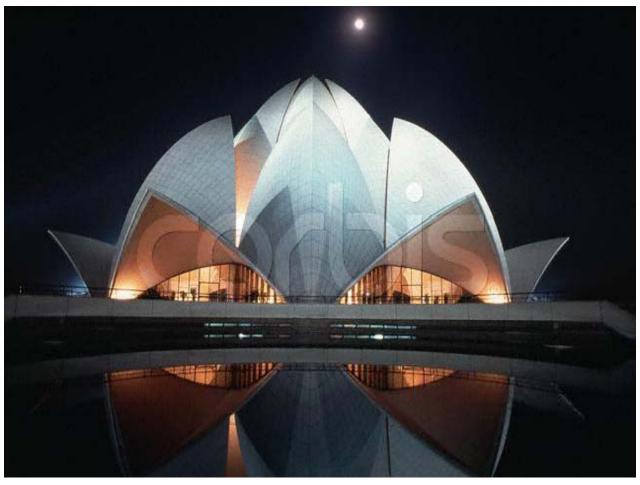
Blended Cement



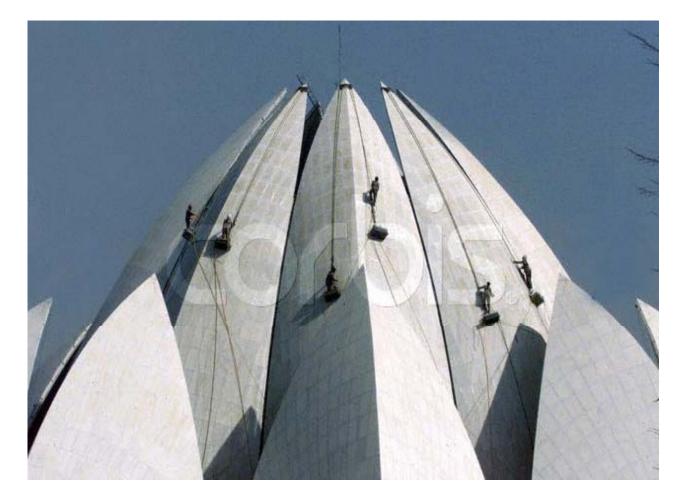


- Type K: shrinkage compensating cement
- Jet Set Cement: Fast Setting (3-5 min)
- Oil Well Cement
- White Cement

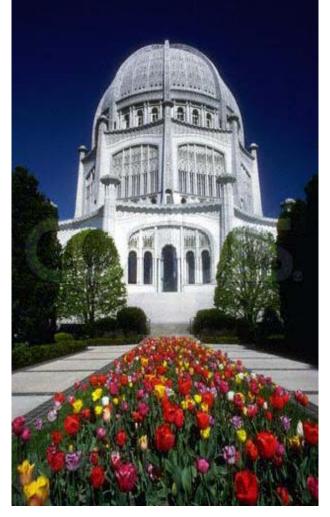


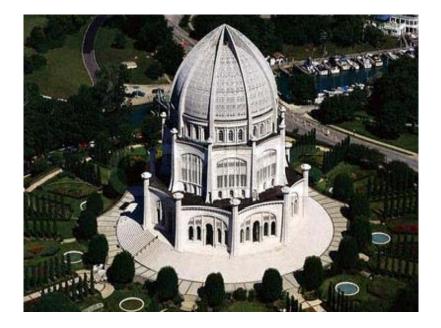


















Calcium Aluminate Cement

- high early strength
- hardening even at low temperatures
- superior durability to sulfate attack
- fast hydration



$CA + 10 H \rightarrow CAH10$

Conversion

C3AH6 + 2AH3 + 18H



Hot Cement

We are dealing with a cement shortage in our region, and during summer we must use cement which is quite hot. The cement comes directly from a cement plant located only 10 miles away from our concrete plant, and in the summer it is often so hot that it hurts to hold it in hand. This causes us many problems with our resistances. The compression resistances of our concretes are correct, though very erratic. Sometimes, for example, the resistance after 7 days is as high as it should have been after 28 days. What can we do about this?