

Proposed Recommendations

Texas Essential Knowledge and Skills

High School Science, Grades 9-12

Proposed Recommendations
to the State Board of Education

January 5, 2009

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Texas Essential Knowledge and Skills
High School Science, Grades 9-12

Prepared by the State Board of Education TEKS Writing Teams

January 5, 2009

These documents have been combined from grade level team drafts
and formatted for consistency and for ease of review.

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§112.42. Integrated Physics and Chemistry

- (a) **General requirements.** Students shall be awarded one credit for successful completion of this course. Prerequisites: none. This course is recommended for students in Grade 9 or 10.
- (b) **Introduction.**
- (1) In Integrated Physics and Chemistry, students conduct field and laboratory investigations, use scientific methods during investigation, and make informed decisions using critical-thinking and scientific problem-solving. This course integrates the disciplines of physics and chemistry in the following topics: force, motion, energy and matter.
 - (2) **NATURE OF SCIENCE**
Science, as defined by the National Academy of Sciences, is the “use of evidence to construct testable explanations and predictions of natural phenomena, as well as the knowledge generated through this process.” This vast body of changing and increasing knowledge is described by physical, mathematical and conceptual models. Students should know that some questions are outside the realm of science because they deal with phenomena that are not scientifically testable.
 - (3) **SCIENTIFIC INQUIRY**
Scientific inquiry is the planned and deliberate investigation about the natural world. Scientific methods of investigation are experimental, descriptive or comparative. The method chosen should be appropriate to the question being asked.
 - (4) **SCIENCE AND SOCIAL ETHICS**
Scientific decision-making is a way of answering questions about the natural world. Students should be able to distinguish between scientific decision-making methods (scientific methods) and ethical/social decisions that involve science (the application of scientific information).
 - (5) **SCIENCE, SYSTEMS AND MODELS**
A system is a collection of cycles, structures and processes that interact. All systems have basic properties that can be described in space, time, energy and matter. Change and constancy occur in systems as patterns and can be observed, measured and modeled. These patterns help to make predictions that can be scientifically tested. Students should analyze a system in terms of its components and how these components relate to each other, to the whole, and to the external environment.
- (c) **Knowledge and skills.**
- (1) **Scientific processes.** The student, for at least 40% of instructional time, conducts field and laboratory investigations using safe, environmentally appropriate, and ethical practices. The student is expected to:
 - (A) demonstrates safe practices during field and laboratory investigations; and
 - (B) make wise choices in the use and conservation of resources and the disposal or recycling of materials.

- (2) **Scientific processes.** The student uses scientific methods during field and laboratory investigations. The student is expected to:
- (A) plan and implement investigative procedures including asking questions, formulating testable hypotheses, and selecting equipment and technology;
 - (B) collect data and make measurements with precision;
 - (C) organize, analyze, evaluate, make inferences, and predict trends from data; and
 - (D) communicate valid conclusions.
- (3) **Scientific processes.** The student uses critical thinking, scientific reasoning and problem solving to make informed decisions. The student is expected to:
- (A) analyze and evaluate scientific explanations, using empirical evidence, logical reasoning, and experimental and observational testing;
 - (B) draw inferences based on data related to promotional materials for products and services;
 - (C) evaluate the impact of research on scientific thought, society, and the environment;
 - (D) describe connections between physics and chemistry, and future careers; and
 - (E) research and describe the history of physics, chemistry, and contributions of scientists.
- (4) **Science concepts.** The student knows concepts of force and motion evident in everyday life. The student is expected to:
- (A) describe and calculate an object's motion in terms of position, displacement, speed and acceleration;
 - (B) measure and graph distance and speed as a function of time using moving toys;
 - (C) investigate how an object's motion changes only when a net force is applied, including activities and equipment such as toy cars, vehicle restraints, sports activities and classroom objects;
 - (D) assess the relationship between force, mass and acceleration, noting the relationship is independent of the nature of the force, using equipment such as dynamic carts, moving toys, vehicles and falling objects;
 - (E) apply concept of conservation of momentum using action and reaction forces such as is illustrated by students on skateboards;
 - (F) describe the gravitational attraction between objects of different masses at different distances including satellites; and

- (G) examine electrical force as a universal force between any two charged objects; and compare the relative strength of the electrical force and gravitational force.
- (5) **Science concepts.** The student recognizes multiple forms of energy and knows the impact of energy transfer and energy conservation in everyday life. The student is expected to:
- (A) recognize and demonstrate that objects and substances in motion have kinetic energy such as vibration of atoms, water flowing down a stream moving pebbles, and bowling balls knocking down pins;
 - (B) demonstrate common forms of potential energy including gravitational, elastic and chemical, such as a ball on an inclined plane, springs and batteries;
 - (C) demonstrate that moving electric charges produce magnetic forces and moving magnets produce electric forces;
 - (D) investigate the law of conservation of energy;
 - (E) investigate and demonstrate the movement of thermal energy through solids, liquids, and gases by convection, conduction and radiation, such as in weather, living and mechanical systems;
 - (F) evaluate the transfer of electrical energy in series and parallel circuits, and conductive materials;
 - (G) explore the characteristics and behaviors of energy transferred by waves including acoustic, seismic, light and waves on water as they superpose on one another, bend around corners, reflect off surfaces, are absorbed by materials and change direction when entering new materials;
 - (H) analyze energy conversions such as those from radiant, nuclear, and geothermal sources, fossil fuels such as coal, gas, oil, and the movement of water or wind; and
 - (I) critique the advantages and disadvantages of various energy sources and their impact on society and the environment.
- (6) **Science concepts.** The student knows that relationships exist between the structure and properties of matter. The student is expected to:
- (A) examine differences in physical properties of solids, liquids and gases as explained by the arrangement and motion of atoms, ions or molecules of the substances and the strength of the forces of attraction between those particles;
 - (B) relate chemical properties of substances to the arrangement of their atoms or molecules;

- (C) analyze physical and chemical properties of elements and compounds such as, color, density, viscosity, buoyancy, boiling point, freezing point, conductivity, and reactivity;
 - (D) relate the physical and chemical behavior of an element including bonding and classification to its placement on the periodic table; and
 - (E) relate the structure of water to its function as a solvent and investigate the properties of solutions and factors affecting gas and solid solubility including nature of solute, temperature, pressure, pH, and concentration.
- (7) **Science concepts.** The student knows that changes in matter affect everyday life. The student is expected to:
- (A) investigate changes of state as it relates to the arrangement of particles of matter and energy transfer;
 - (B) recognize chemical changes can occur when substances react to form different substances and that these interactions are largely determined by the valence electrons;
 - (C) demonstrate that mass is conserved when substances undergo chemical change and that the number and kind of atoms are the same in the reactants and products;
 - (D) analyze energy changes that accompany chemical reactions such as those occurring in heat packs, cold packs, and glow sticks and classify them as exothermic or endothermic reactions;
 - (E) describe types of nuclear reactions such as fission and fusion and their roles in applications such as medicine and energy production; and
 - (F) research and describe the environmental and economic impact of the end-products of chemical reactions such as those that may result in acid rain, degradation of water and air quality, and ozone depletion.

End of Integrated Physics and Chemistry

§112.43. Biology

- (a) **General requirements.** Students shall be awarded one credit for successful completion of this course. Prerequisites: none. This course is recommended for students in Grades 9, 10, or 11.
- (b) **Introduction.**
- (1) In Biology, students conduct field and laboratory investigations, use scientific methods during investigations, and make informed decisions using critical-thinking and scientific problem solving. Students in Biology study a variety of topics that include: structures and functions of cells and viruses; growth and development of organisms; cells, tissues, and organs; nucleic acids and genetics; biological evolution; taxonomy; metabolism and energy transfers in living organisms; living systems; homeostasis; and ecosystems and the environment.
 - (2) **NATURE OF SCIENCE**
Science, as defined by the National Academy of Sciences, is the “use of evidence to construct testable explanations and predictions of natural phenomena, as well as the knowledge generated through this process.” This vast body of changing and increasing knowledge is described by physical, mathematical and conceptual models. Students should know that some questions are outside the realm of science because they deal with phenomena that are not scientifically testable.
 - (3) **SCIENTIFIC INQUIRY**
Scientific inquiry is the planned and deliberate investigation about the natural world. Scientific methods of investigation are experimental, descriptive or comparative. The method chosen should be appropriate to the question being asked.
 - (4) **SCIENCE AND SOCIAL ETHICS**
Scientific decision-making is a way of answering questions about the natural world. Students should be able to distinguish between scientific decision-making methods (scientific methods) and ethical/social decisions that involve science (the application of scientific information).
 - (5) **SCIENCE, SYSTEMS AND MODELS**
A system is a collection of cycles, structures and processes that interact. All systems have basic properties that can be described in space, time, energy and matter. Change and constancy occur in systems as patterns and can be observed, measured and modeled. These patterns help to make predictions that can be scientifically tested. Students should analyze a system in terms of its components and how these components relate to each other, to the whole, and to the external environment.

(c) **Knowledge and skills.**

- (1) **Scientific processes.** The student, for at least 40% of instructional time, conducts field and laboratory investigations using safe, environmentally appropriate, and ethical practices. The student is expected to:
- (A) demonstrate safe practices during field and laboratory investigations; and
 - (B) make wise choices in the use and conservation of resources and the disposal or recycling of materials.
- (2) **Scientific processes.** The student uses scientific methods and equipment during field and laboratory investigations. The student is expected to:
- (A) know that hypotheses are tentative and testable statements that must be capable of being supported or not supported by observational evidence. Hypotheses of durable explanatory power which have been tested over a wide variety of conditions are incorporated into theories;
 - (B) know scientific theories are based on natural and physical phenomena and are capable of being tested by multiple independent researchers. Unlike hypotheses, scientific theories are well-established and highly reliable explanations, but they may be subject to change as new areas of science and new technologies are developed;
 - (C) distinguish between scientific hypotheses and scientific theories;
 - (D) plan and implement investigative procedures including asking questions, formulating testable hypotheses, and selecting equipment and technology;
 - (E) collect and organize qualitative and quantitative data, and make measurements with accuracy and precision using tools such as calculators, spreadsheet software, data-collecting probes, computers, standard laboratory glassware, microscopes, various prepared slides, stereoscopes, metric rulers, electronic balances, gel electrophoresis apparatuses, micropipettors, hand lenses, Celsius thermometers, hot plates, lab notebooks or journals, timing devices, cameras, Petri dishes, lab incubators, dissection equipment, meter sticks, and models, diagrams or samples of biological specimens or structures;
 - (F) analyze, evaluate, make inferences, and predict trends from data; and
 - (G) communicate valid conclusions supported by the data through methods such as lab reports, labeled drawings, graphic organizers, journals, summaries, oral reports, and technology based reports.

- (3) **Scientific processes.** The student uses critical thinking, scientific reasoning and problem solving to make informed decisions within and outside the classroom. The student is expected to:
- (A) analyze and evaluate scientific explanations using empirical evidence, logical reasoning, and experimental and observational testing;
 - (B) evaluate promotional claims that relate to biological issues such as product labeling and advertisements;
 - (C) evaluate the impact of scientific research on society and the environment;
 - (D) evaluate models according to their limitations in representing biological objects or events; and
 - (E) communicate and apply scientific information extracted from various sources.
- (4) **Science concepts.** The student knows that cells are the basic structures of all living things with specialized parts that perform specific functions, and that viruses are different from cells. The student is expected to:
- (A) compare and contrast prokaryotic and eukaryotic cells;
 - (B) investigate and explain cellular processes including homeostasis, energy conversions, transport of molecules, and synthesis of new molecules; and
 - (C) compare the structures of viruses to cells, describe viral reproduction, and describe the role of viruses in causing diseases, such as human immunodeficiency virus (HIV) and influenza.
- (5) **Science concepts.** The student knows how an organism grows, and the importance of cell differentiation. The student is expected to:
- (A) describe the stages of the cell cycle, including deoxyribonucleic acid (DNA) replication and mitosis, and the importance of the cell cycle to the growth of organisms;
 - (B) examine specialized cells including roots, stems, and leaves of plants; and animal cells such as blood, muscle, and epithelium;
 - (C) describe the roles of DNA, ribonucleic acid (RNA), and environmental factors in cell differentiation; and
 - (D) recognize that disruptions of the cell cycle lead to diseases such as cancer.

- (6) **Science concepts.** The student knows the mechanisms of genetics including the role of nucleic acids and the principles of Mendelian Genetics. The student is expected to:
- (A) identify components of deoxyribonucleic acid (DNA), and describe how information for specifying the traits of an organism is carried in the DNA;
 - (B) recognize that components that make up the genetic code are common to all organisms;
 - (C) explain the purpose and process of transcription, and translation using models of DNA and ribonucleic acid (RNA);
 - (D) recognize that gene expression is a regulated process;
 - (E) identify and illustrate changes in DNA and evaluate the significance of these changes;
 - (F) predict possible outcomes of various genetic combinations such as monohybrid crosses, dihybrid crosses and non-Mendelian inheritance;
 - (G) recognize the significance of meiosis to sexual reproduction; and
 - (H) describe how techniques such as DNA fingerprinting, genetic modifications and chromosomal analysis are used to study the genomes of organisms.
- (7) **Science concepts.** The student knows evolutionary theory is a scientific explanation for the unity and diversity of life. The student is expected to:
- (A) identify how evidence of common ancestry among groups is provided by the fossil record, biogeography, and homologies including anatomical, molecular, and developmental;
 - (B) recognize that natural selection produces change in populations, not individuals;
 - (C) describe the elements of natural selection including inherited variation, the potential of a population to produce more offspring than can survive, and a finite supply of environmental resources resulting in differential reproductive success;
 - (D) recognize the relationship of natural selection to adaptation, and to the development of diversity in and among species; and
 - (E) recognize the effects of other evolutionary mechanisms including genetic drift, gene flow, mutation, and recombination.

- (8) **Science concepts.** The student knows that taxonomy is a branching classification based on the shared characteristics of organisms and can change as new discoveries are made. The student is expected to:
- (A) define taxonomy and recognize the importance of a standardized taxonomic system to the scientific community;
 - (B) categorize organisms using a hierarchical classification system based on similarities and differences shared among groups; and
 - (C) compare characteristics of taxonomic groups including archaea, bacteria, protists, fungi, plants, and animals.
- (9) **Science concepts.** The student knows the significance of various molecules involved in metabolic processes and energy conversions that occur in living organisms. The student is expected to:
- (A) compare the structures and functions of different types of biomolecules including carbohydrates, lipids, proteins, and nucleic acids;
 - (B) compare the reactants and products of photosynthesis and cellular respiration in terms of energy and matter; and
 - (C) identify and investigate the role of enzymes.
- (10) **Science concepts.** The student knows that biological systems are composed of multiple levels. The student is expected to:
- (A) describe the interactions that occur among systems that perform the functions of regulation, nutrient absorption, reproduction, and defense from injury or illness in animals;
 - (B) describe the interactions that occur among systems that perform the functions of transport, reproduction, and response in plants; and
 - (C) analyze the levels of organization in biological systems and relate the levels to each other and to the whole system.
- (11) **Science concepts.** The student knows that biological systems work to achieve and maintain balance. The student is expected to:
- (A) describe the role of internal feedback mechanisms in the maintenance of homeostasis;
 - (B) investigate and analyze how organisms, populations, and communities respond to external factors;
 - (C) summarize the role of microorganisms in both maintaining and disrupting the health of both organisms and ecosystems; and

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- (D) describe how events and processes that occur during ecological succession can change populations and species diversity.
- (12) **Science concepts.** The student knows that interdependence and interactions occur within an environmental system. The student is expected to:
- (A) interpret relationships including predation, parasitism, commensalism, mutualism, and competition among organisms;
 - (B) compare variations and adaptations of organisms in different ecosystems;
 - (C) analyze the flow of matter and energy through trophic levels using various models including food chains, food webs, and ecological pyramids;
 - (D) recognize that long-term survival of species is dependent on changing resource bases that are limited;
 - (E) describe the flow of matter through the carbon and nitrogen cycles and explain the consequences of disrupting these cycles; and
 - (F) describe how environmental change can impact ecosystem stability.

End of Biology

§112.45. Chemistry

- (a) **General requirements.** Students shall be awarded one credit for successful completion of this course. Required prerequisites: one unit of high school science and Algebra I. Suggested prerequisite: completion of or concurrent enrollment in a second year of math. This course is recommended for students in Grades 10, 11, or 12.
- (b) **Introduction.**
- (1) In Chemistry, students conduct field and laboratory investigations, use scientific methods during investigations, and make informed decisions using critical thinking and scientific problem solving. Students study a variety of topics that include characteristics of matter, use of the periodic table, development of atomic theory and chemical bonding, chemical stoichiometry, gas laws, solution chemistry thermochemistry and nuclear chemistry. Students will investigate how chemistry is an integral part of our daily lives.
 - (2) **NATURE OF SCIENCE**
Science, as defined by the National Academy of Sciences, is the “use of evidence to construct testable explanations and predictions of natural phenomena, as well as the knowledge generated through this process.” This vast body of changing and increasing knowledge is described by physical, mathematical, and conceptual models. Students should know that some questions are outside the realm of science because they deal with phenomena that are not scientifically testable.
 - (3) **SCIENTIFIC INQUIRY**
Scientific inquiry is the planned and deliberate investigation of the natural world. Scientific methods of investigation can be experimental, descriptive, or comparative. The method chosen should be appropriate to the question being asked.
 - (4) **SCIENCE AND SOCIAL ETHICS**
Scientific decision-making is a way of answering questions about the natural world. Students should be able to distinguish between scientific decision-making methods and ethical/social decisions that involve the application of scientific information.
 - (5) **SCIENTIFIC SYSTEMS**
A system is a collection of cycles, structures, and processes that interact. All systems have basic properties that can be described in terms of space, time, energy, and matter. Change and constancy occur in systems as patterns and can be observed, measured, and modeled. These patterns help to make predictions that can be scientifically tested. Students should analyze a system in terms of its components and how these components relate to each other, to the whole, and to the external environment.
- (c) **Knowledge and skills.**
- (1) **Scientific processes.** The student, for at least 40% of instructional time, conducts field and laboratory investigations using safe, environmentally appropriate, and ethical practices. The student is expected to:

- (A) demonstrate safe practices during all investigations; including use of safety showers, eyewash fountains, safety goggles, and fire extinguishers;
 - (B) know specific hazards of chemical substances such as flammability, corrosiveness and radioactivity as summarized on the Material Safety Data Sheets; and
 - (C) demonstrate an understanding of the use and conservation of resources and the proper disposal or recycling of materials.
- (2) **Scientific processes.** The student uses scientific methods to solve investigative questions. The student is expected to:
- (A) know that scientific hypotheses are tentative and testable statements that must be capable of being supported or not supported by observational evidence. Hypotheses of durable explanatory power which have been tested over a wide variety of conditions are incorporated into theories;
 - (B) know that scientific theories are based on natural and physical phenomena and are capable of being tested by multiple independent researchers. Unlike hypotheses, scientific theories are well-established and highly reliable explanations, but may be subject to change as new areas of science and new technologies are developed;
 - (C) distinguish between scientific hypotheses and scientific theories;
 - (D) plan and implement investigative procedures including asking questions, formulating testable hypotheses, and selecting equipment and technology; including graphing calculators, computers and probes, sufficient scientific glassware such as beakers, Erlenmeyer flasks, pipettes, graduated cylinders, volumetric flasks, safety goggles, burettes, electronic balances and an adequate supply of consumable chemicals;
 - (E) collect data and make measurements with accuracy and precision;
 - (F) express and manipulate chemical quantities using scientific conventions and mathematical procedures including dimensional analysis, scientific notation, and significant figures;
 - (G) organize, analyze, evaluate, make inferences, and predict trends from data; and
 - (H) communicate valid conclusions supported by the data through methods such as lab reports, labeled drawings, graphs, journals, summaries, oral reports and technology based reports.

- (3) **Scientific processes.** The student uses critical thinking, scientific reasoning and problem solving skills to make informed decisions within and outside the classroom. The student is expected to:
- (A) analyze and evaluate scientific explanations using empirical evidence, logical reasoning, and experimental and observational testing.
 - (B) make responsible choices in selecting everyday products and services using scientific information;
 - (C) evaluate the impact of research on scientific thought, society, and the environment;
 - (D) describe the connection between chemistry and future careers; and
 - (E) research and describe the history of chemistry and contributions of scientists.
- (4) **Science concepts.** The student knows the characteristics of matter and can analyze the relationships between chemical and physical changes and properties. The student is expected to:
- (A) differentiate between physical and chemical changes and properties. Identify extensive and intensive properties;
 - (B) compare solids, liquids, and gases in terms of compressibility, structure, shape, and volume; and
 - (C) classify matter as pure substances or mixtures through investigation of their properties.
- (5) **Science concepts.** The student understands the historical development of the Periodic Table and can apply its predictive power. The student is expected to:
- (A) explain the use of chemical and physical properties in the historical development of the Periodic Table;
 - (B) use the Periodic Table to identify and explain the properties of chemical families including alkali metals, alkaline earth metals, halogens, noble gases, and transition metals; and
 - (C) use the Periodic Table to identify and explain periodic trends including atomic and ionic radii, electronegativity, and ionization energy.

- (6) **Science concepts.** The student knows and understands the historical development of atomic theory. The student is expected to:
- (A) understand the experimental design and conclusions used in the development of modern atomic theory including Dalton's Postulates, Thomson's discovery of electron properties, Rutherford's nuclear atom, and Bohr's nuclear atom;
 - (B) understand the electromagnetic spectrum and the mathematical relationships between energy, frequency, and wavelength of light;
 - (C) calculate the wavelength, frequency and energy of light using Planck's constant and the speed of light;
 - (D) use isotopic composition to calculate average atomic mass of an element; and
 - (E) express the arrangement of electrons in atoms through electron configurations and Lewis valence electron dot structures.
- (7) **Science concepts.** The student knows how atoms form ionic, metallic and covalent bonds. The student is expected to:
- (A) name ionic compounds containing main group or transition metals, covalent compounds, acids and bases, using IUPAC nomenclature rules;
 - (B) write the chemical formulas of common polyatomic ions, ionic compounds containing main group or transition metals, covalent compounds, acids and bases;
 - (C) construct electron dot formulas to illustrate ionic and covalent bonds;
 - (D) describe the nature of metallic bonding and apply the theory to explain metallic properties such as thermal and electrical conductivity, malleability and ductility; and
 - (E) predict molecular structure for molecules with linear, trigonal planar or tetrahedral electron pair geometries using VSEPR theory.
- (8) **Science concepts.** The student can quantify the changes that occur during chemical reactions. The student is expected to:
- (A) define and utilize the concept of a mole;
 - (B) utilize the mole concept to calculate the number of atoms, ions or molecules in a sample of material;
 - (C) calculate percent composition and empirical and molecular formulas;
 - (D) use the law of conservation of mass to write and balance chemical equations; and

- (E) perform stoichiometric calculations including determination of mass relationships between reactants and products, calculation of limiting reagents and percent yield.
- (9) **Science concepts.** The student understands the principles of ideal gas behavior, kinetic molecular theory and the conditions that influence the behavior of gases. The student is expected to:
- (A) describe and calculate the relations between volume, pressure, number of moles, and temperature for an ideal gas, as described by Boyle's Law, Charles' Law, Avogadro's Law, Dalton's Law of Partial Pressure and the Ideal Gas Law;
 - (B) perform stoichiometric calculations including determination of mass and volume relationships between reactants and products for reactions involving gases; and
 - (C) describe the postulates of kinetic molecular theory.
- (10) **Science concepts.** The student understands and can apply the factors that influence the behavior of solutions. The student is expected to:
- (A) describe the unique role of water in chemical and biological systems;
 - (B) develop and use general rules regarding solubility through investigations with aqueous solutions;
 - (C) calculate the concentration of solutions in units of molarity;
 - (D) use molarity to calculate the dilutions of solutions;
 - (E) distinguish between types of solutions such as electrolytes and nonelectrolytes; and unsaturated, saturated, and supersaturated solutions;
 - (F) investigate factors that influence solubilities and rates of dissolution such as temperature, agitation, and surface area;
 - (G) define acids and bases and distinguish between Arrhenius and Bronsted-Lowery definitions; and predict products in acid base reactions that form water;
 - (H) understand and differentiate among acid-base reactions, precipitation reactions, and oxidation-reduction reactions;
 - (I) define pH and use the hydrogen or hydroxide ion concentrations to calculate the pH of a solution; and
 - (J) distinguish between degrees of dissociation for strong and weak acids and bases.

- (11) **Science concepts.** The student understands the energy changes that occur in chemical reactions. The student is expected to:
- (A) understand energy and its forms including kinetic, potential, chemical and thermal energies;
 - (B) understand the law of Conservation of Energy and the processes of heat transfer;
 - (C) use thermochemical equations to calculate energy changes that occur in chemical reactions and classify reactions as exothermic or endothermic;
 - (D) perform calculations involving heat, mass, temperature change and specific heat; and
 - (E) use calorimetry to calculate the heat of a chemical process.
- (12) **Science concepts.** The student understands the basic processes of nuclear chemistry. The student is expected to:
- (A) describe the characteristics of alpha, beta and gamma radiation;
 - (B) describe radioactive decay process in terms of balanced nuclear equations; and
 - (C) compare fission and fusion reactions.

End of Chemistry

§112.47. Physics

(a) **General requirements.** Students shall be awarded one credit for successful completion of this course. Algebra I is suggested as a prerequisite or co-requisite.

(b) **Introduction**

(1) **PHYSICS**

In Physics, students conduct field and laboratory investigations, use scientific methods during investigations, and make informed decisions using critical thinking and scientific problem solving. Students study a variety of topics that include: laws of motion; changes within physical systems and conservation of energy and momentum; forces; thermodynamics; characteristics and behavior of waves; and atomic, nuclear, and quantum physics. Students who successfully complete Physics will acquire factual knowledge within a conceptual framework, practice experimental design and interpretation, work collaboratively with colleagues, and develop critical thinking skills.

(2) **NATURE OF SCIENCE**

Science, as defined by the National Academy of Sciences, is the “use of evidence to construct testable explanations and predictions of natural phenomena, as well as the knowledge generated through this process.” This vast body of changing and increasing knowledge is described by physical, mathematical, and conceptual models. Students should know that some questions are outside the realm of science because they deal with phenomena that are not scientifically testable.

(3) **SCIENTIFIC INQUIRY**

Scientific inquiry is the planned and deliberate investigation about the natural world. Scientific methods of investigation can be experimental, descriptive, or comparative. The method chosen should be appropriate to the question being asked.

(4) **SCIENCE AND SOCIAL ETHICS**

Scientific decision-making is a way of answering questions about the natural world. Students should be able to distinguish between scientific decision-making methods and ethical/social decisions that involve the application of scientific information.

(5) **SCIENTIFIC SYSTEMS**

A system is a collection of cycles, structures, and processes that interact. All systems have basic properties that can be described in terms of space, time, energy, and matter. Change and constancy occur in systems as patterns and can be observed, measured, and modeled. These patterns help to make predictions that can be scientifically tested. Students should analyze a system in terms of its components and how these components relate to each other, to the whole, and to the external environment.

(c) **Knowledge and skills**

- (1) **Scientific processes.** The student conducts investigations for at least 40% of instructional time using safe, environmentally appropriate, and ethical practices. These investigations must involve actively obtaining and analyzing data with physical equipment, but may also involve experimentation in a simulated environment as well as field observations that extend beyond the classroom. The student is expected to:
- (A) demonstrate safe practices during field and laboratory investigations; and
 - (B) make informed choices in the use and conservation of resources, recycling of materials, and the safe disposal of science classroom or household chemicals.
- (2) **Scientific processes.** The student uses a systematic approach to solve scientific field and laboratory investigative questions. The student is expected to:
- (A) know that scientific hypotheses are tentative and testable statements that must be capable of being supported or not supported by observational evidence. Hypotheses of durable explanatory power which have been tested over a wide variety of conditions are incorporated into theories;
 - (B) know that scientific theories are based on natural and physical phenomena and are capable of being tested by multiple independent researchers. Unlike hypotheses, scientific theories are well-established and highly reliable explanations, but may be subject to change as new areas of science and new technologies are developed;
 - (C) distinguish between scientific hypotheses and scientific theories;
 - (D) design and implement investigative procedures including making observations, asking well-defined questions, formulating testable hypotheses, identifying variables, selecting appropriate equipment and technology, and evaluating numerical answers for reasonableness;
 - (E) demonstrate the use of course apparatus, equipment, techniques, and procedures including multimeters (current, voltage, resistance), triple beam balances, batteries, clamps, dynamics demonstration equipment, collision apparatus, data acquisition probes, discharge tubes with power supply (H, He, Ne, Ar), hand-held visual spectroscope, hot plate, slotted and hooked lab masses, bar magnets, horseshoe magnets, plane mirrors, convex lenses, pendulum support, power supply, ring clamps, ring stands, stopwatches, trajectory apparatus, tuning forks, carbon paper, graph paper, magnetic compass, polarized film, prisms, protractors, resistors, friction blocks, mini lamps (bulbs) and sockets, electrostatics kits, 90 degree rod clamps, metric rulers, spring scales, knife blade switches, Celsius thermometers, meter sticks, scientific calculators, graphing technology, computers, cathode ray tube with horseshoe magnet, ballistic cart or equivalent, resonance tube, spool of nylon thread or string, container of iron filings, roll of white craft paper, copper wire, periodic table, and electromagnetic spectrum charts, slinky springs, wave motion rope, laser pointers.

- (F) uses a wide variety of additional course apparatuses, equipment, techniques, materials, and procedures as appropriate such as ripple tank with wave generator, wave motion rope, micrometer, caliper, radiation monitor, computer, ballistic pendulum, electroscope, inclined plane, optics bench, optics kit, pulley with table clamp, resonance tube, ring stand screen, 4" ring, stroboscope, graduated cylinders, and ticker timer.
 - (G) make measurements with accuracy and precision and record data using scientific notation and International System (SI) units.
 - (H) identify and quantify causes and effects of uncertainties in measured data;
 - (I) organize and evaluate data and make inferences from data including the use of tables, charts, and graphs;
 - (J) communicate valid conclusions supported by the data through various methods such as lab reports, labeled drawings, graphic organizers, journals, summaries, oral reports, and technology based reports; and
 - (K) express and manipulate relationships among physical variables quantitatively including the use of graphs, charts, and equations.
- (3) **Scientific processes.** The student uses critical thinking, scientific reasoning, and problem solving skills to make informed decisions within and outside the classroom. The student is expected to:
- (A) analyze and evaluate scientific explanations by using empirical evidence, logical reasoning, and experimental and observational testing;
 - (B) communicate and apply scientific information extracted from written text;
 - (C) evaluate promotional claims such as product labeling and advertisements;
 - (D) explain the impacts of the scientific contributions of a variety of historical and contemporary scientists on scientific thought and society;
 - (E) research and describe the connections between physics and future careers; and
 - (F) express and interpret relationships symbolically in accordance with accepted theories to make predictions and solve problems mathematically including problems requiring proportional reasoning and graphical vector addition.
- (4) **Science concepts.** The student knows and applies the laws governing motion in a variety of situations. The student is expected to:
- (A) generate and interpret graphs and charts describing different types of motion including the use of real-time technology such as motion detectors or photogates;

- (B) describe and analyze motion in one dimension using equations with the concepts of distance, displacement, speed, average velocity, instantaneous velocity, and acceleration;
 - (C) analyze and describe accelerated motion in two dimensions using equations including projectile and circular examples;
 - (D) calculate the effect of forces on objects including the law of inertia, the relationship between force and acceleration, and the nature of force pairs between objects;
 - (E) develop and interpret free-body force diagrams; and
 - (F) identify and describe motion relative to different frames of reference.
- (5) **Science concepts.** The student knows the nature of forces in the physical world. The student is expected to:
- (A) research and describe the historical development of the concepts of gravitational, electromagnetic, weak nuclear, and strong nuclear forces;
 - (B) describe and calculate how the magnitude of the gravitational force between two objects depends on their masses and the distance between their centers;
 - (C) describe and calculate how the magnitude of the electrical force between two objects depends on their charges and the distance between their centers;
 - (D) identify examples of electric and magnetic forces in everyday life;
 - (E) characterize materials as conductors or insulators based on their electrical properties;
 - (F) design, construct, and calculate in terms of current through, potential difference across, resistance of, and power used by electric circuit elements connected in both series and parallel combinations;
 - (G) investigate and describe the relationship between electric and magnetic fields in applications such as generators, motors, and transformers; and
 - (H) describe evidence for and effects of the strong and weak nuclear forces in nature.
- (6) **Science concepts.** The student knows that changes occur within a physical system and applies the laws of conservation of energy and momentum. The student is expected to:
- (A) investigate and calculate with the work-energy theorem in various situations;
 - (B) investigate examples of kinetic and potential energy and their transformations;

- (C) calculate the mechanical energy of, power generated within, impulse applied to, and momentum of a physical system;
 - (D) demonstrate and apply the laws of conservation of energy and conservation of momentum in one dimension;
 - (E) describe how the macroscopic properties of a thermodynamic system such as temperature, specific heat, and pressure are related to the molecular level of matter including kinetic or potential energy of atoms;
 - (F) contrast and give examples of different processes of thermal energy transfer including conduction, convection, and radiation; and
 - (G) analyze and explain everyday examples that illustrate the laws of thermodynamics, including the law of conservation of energy and the law of entropy.
- (7) **Science concepts.** The student knows the characteristics and behavior of waves. The student is expected to:
- (A) examine and describe oscillatory motion and wave propagation in various types of media;
 - (B) investigate and analyze characteristics of waves including velocity, frequency, amplitude, and wavelength and calculate using the relationship between wavespeed, frequency, and wavelength;
 - (C) compare characteristics and behaviors of transverse waves including electromagnetic waves and the electromagnetic spectrum and characteristics and behaviors of longitudinal waves including sound waves;
 - (D) investigate behaviors of waves including reflection, refraction, diffraction, interference, resonance, and the Doppler effect;
 - (E) describe and predict image formation as a consequence of reflection from a plane mirror and refraction through a thin convex lens; and
 - (F) describe the role of wave characteristics and behaviors in medical and industrial applications.
- (8) **Science concepts.** The student knows simple examples of atomic, nuclear, and quantum phenomena. The student is expected to:
- (A) describe the photoelectric effect and the dual nature of light;
 - (B) compare and explain the emission spectra produced by various atoms;

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- (C) describe the significance of mass–energy equivalence and apply it in explanations of phenomena such as nuclear stability, fission, and fusion; and
 - (D) give examples of applications of atomic and nuclear phenomena such as radiation therapy, diagnostic imaging, and nuclear power and examples of quantum phenomena, such as digital cameras. .

End of Physics

§112.44. Environmental Systems

- (a) **General requirements.** Students shall be awarded one credit for successful completion of this course. Suggested prerequisite: one unit high school life science and one unit of high school physical science. This course is recommended for students in Grades 11 or 12.
- (b) **Introduction.**
- (1) In Environmental Systems, students conduct field and laboratory investigations, use scientific methods during investigations, and make informed decisions using critical thinking and scientific problem solving. Students study a variety of topics that include: biotic and abiotic factors in habitats; ecosystems and biomes; interrelationships among resources and an environmental system; sources and flow of energy through an environmental system; relationship between carrying capacity and changes in populations and ecosystems; and changes in environments.
 - (2) **NATURE OF SCIENCE**
Science, as defined by the National Academy of Sciences, is the “use of evidence to construct testable explanations and predictions of natural phenomena, as well as the knowledge generated through this process.” This vast body of changing and increasing knowledge is described by physical, mathematical, and conceptual models. Students should know that some questions are outside the realm of science because they deal with phenomena that are not scientifically testable.
 - (3) **SCIENTIFIC INQUIRY**
Scientific inquiry is the planned and deliberate investigation of the natural world. Scientific methods of investigation can be experimental, descriptive, or comparative. The method chosen should be appropriate to the question being asked.
 - (4) **SCIENCE AND SOCIAL ETHICS**
Scientific decision-making is a way of answering questions about the natural world. Students should be able to distinguish between scientific decision-making methods and ethical/social decisions that involve the application of scientific information.
 - (5) **SCIENTIFIC SYSTEMS**
A system is a collection of cycles, structures, and processes that interact. All systems have basic properties that can be described in terms of space, time, energy, and matter. Change and constancy occur in systems as patterns and can be observed, measured, and modeled. These patterns help to make predictions that can be scientifically tested. Students should analyze a system in terms of its components and how these components relate to each other, to the whole, and to the external environment.
- (c) **Knowledge and skills.**
- (1) **Scientific processes.** The student, for at least 40% of instructional time, conducts hands-on field and laboratory investigations using safe, environmentally appropriate, and ethical practices. The student is expected to:

- (A) demonstrate safe practices during field and laboratory investigations including appropriate first aid responses to accidents that could occur in the field such as insect stings, animal bites, overheating, sprains and breaks; and
 - (B) make wise choices in the use and conservation of resources and the disposal or recycling of materials.
- (2) **Scientific processes.** The student uses scientific methods during field and laboratory investigations. The student is expected to:
- (A) know that scientific hypotheses are tentative and testable statements that must be capable of being supported or not supported by observational evidence. Hypotheses of durable explanatory power which have been tested over a wide variety of conditions are incorporated into theories;
 - (B) know that scientific theories are based on natural and physical phenomena and are capable of being tested by multiple independent researchers. Unlike hypotheses, scientific theories are well-established and highly reliable explanations, but may be subject to change as new areas of science and new technologies are developed;
 - (C) distinguish between scientific hypotheses and scientific theories;
 - (D) follow or plan and implement investigative procedures including making observations, asking questions, formulating testable hypotheses, and selecting equipment and technology;
 - (E) collect data individually or collaboratively, make measurements with precision and accuracy, record values using appropriate units, and calculate statistically relevant quantities to describe data including mean, median, and range;
 - (F) demonstrate the use of course apparatuses, equipment, techniques, and procedures including meter sticks, rulers, pipettes, graduated cylinders, triple beam balances, timing devices, pH meters/probes, thermometers, calculators, computers, internet access, turbidity testing devices, hand magnifiers, work and disposable gloves, compasses, first aid kits, binoculars, field guides, water quality test kits or probes, soil test kits or probes, 100 foot appraiser's tapes, tarps, shovels, trowels, screens, buckets, and rock/mineral samples;
 - (G) use a wide variety of additional course apparatuses, equipment, techniques, materials, and procedures as appropriate such as air quality testing devices, cameras, flow meters, GPS units, GIS software, computer models, densimeters, clinometers, and field journals;
 - (H) organize, analyze, evaluate, build models, make inferences, and predict trends from data;
 - (I) perform calculations using dimensional analysis, significant digits, and scientific notation; and

- (J) communicate valid conclusions supported by the data through methods such as lab reports; labeled drawings, graphic organizers, journals, summaries, oral reports and technology based reports.
- (3) **Scientific processes.** The student uses critical thinking, scientific reasoning, and problem solving to make informed decisions within and outside the classroom. The student is expected to:
- (A) analyze and evaluate scientific explanations using empirical evidence, logical reasoning, and experimental and observational testing;
 - (B) communicate and apply scientific information extracted from written text;
 - (C) make responsible choices in selecting everyday products and services using scientific information;
 - (D) evaluate the impact of research on scientific thought, society, and the environment;
 - (E) describe the connection between environmental science and future careers; and
 - (F) research and describe the history of environmental science and contributions of scientists.
- (4) **Science concepts.** The student knows the relationships of biotic and abiotic factors within habitats, ecosystems, and biomes. The student is expected to:
- (A) identify native plants and animals using a dichotomous key;
 - (B) assess the role of native plants and animals within a local ecosystem, and compare them to plants and animals in ecosystems within four other biomes;
 - (C) diagram abiotic cycles including the rock, hydrologic, carbon, and nitrogen cycles;
 - (D) make observations and compile data about fluctuations in abiotic cycles and evaluate the effects of abiotic factors on local ecosystems and local biomes;
 - (E) measure the concentration of solute, solvent, and solubility of dissolved substances such as dissolved oxygen, chlorides, and nitrates and describe their impact on an ecosystem;
 - (F) predict how the introduction or removal of an invasive species may alter the food chain and affect existing populations in an ecosystem;
 - (G) predict how species extinction may alter the food chain and affect existing populations in an ecosystem; and

- (H) research and explain the causes of species diversity and predict changes that may occur in an ecosystem if species and genetic diversity is increased or reduced.
- (5) **Science concepts.** The student knows the interrelationships among the resources within the local environmental system. The student is expected to:
- (A) summarize methods of land use and management, and describe its effects on land fertility;
 - (B) identify source, use, quality, management, and conservation of water;
 - (C) document the use and conservation of both renewable and non-renewable resources as they pertain to sustainability;
 - (D) identify renewable and non-renewable resources that must come from outside an ecosystem such as food, water, lumber, and energy;
 - (E) analyze and evaluate the economic significance and interdependence of resources within the environmental system; and
 - (F) evaluate the impact of waste management methods such as reduction, reuse, recycling, and composting on resource availability.
- (6) **Science concepts.** The student knows the sources and flow of energy through an environmental system. The student is expected to:
- (A) define and identify the components of the geosphere, hydrosphere, cryosphere, atmosphere, and biosphere and the interactions among them;
 - (B) describe and compare renewable and nonrenewable energy derived from natural and alternative sources such as oil, natural gas, coal, nuclear, solar, geothermal, hydroelectric, and wind;
 - (C) explain the flow of energy in an ecosystem including conduction, convection and radiation;
 - (D) investigate and explain the effects of energy transformations in terms of the laws of thermodynamics within an ecosystem; and
 - (E) investigate and identify energy interactions in an ecosystem.
- (7) **Science concepts.** The student knows the relationship between carrying capacity and changes in populations and ecosystems. The student is expected to:
- (A) relate carrying capacity to population dynamics;
 - (B) calculate birth rates and exponential growth of populations;
 - (C) analyze and predict the effects of non-renewable resource depletion; and

- (D) analyze and make predictions about the impact on populations of geographic locales due to diseases, birth and death rates, urbanization, and natural events, such as migration and seasonal changes.
- (8) **Science concepts.** The student knows that environments change naturally. The student is expected to:
- (A) analyze and describe the effects on areas impacted by natural events such as tectonic movement, volcanic events, fires, tornados, hurricanes, flooding, tsunamis, and population growth;
 - (B) explain how regional changes in the environment may have a global effect;
 - (C) examine how natural processes such as succession and feedback loops restore habitats and ecosystems;
 - (D) describe how temperature inversions impact weather conditions including El Niño and La Niña oscillations; and
 - (E) analyze the impact of temperature inversions on global warming, icecap and glacial melting, and changes in ocean currents and surface temperatures.
- (9) **Science concepts.** The student knows the impact of human activities on the environment. The student is expected to:
- (A) identify causes of air, soil and water pollution including point and nonpoint sources;
 - (B) investigate the types of air, soil, and water pollution such as chlorofluorocarbons, carbon dioxide, pH, pesticide runoff, thermal variations, metallic ions, heavy metals and nuclear waste;
 - (C) examine the concentrations of air, soil, and water pollutants using appropriate units;
 - (D) describe the effect of pollution on global warming, glacial/ice cap melting, greenhouse effect, ozone layer, and aquatic viability;
 - (E) evaluate the effect of human activities including habitat restoration projects, species preservation efforts, nature conservancy groups, hunting, fishing, ecotourism, all terrain vehicles, and small personal water craft on the environment;
 - (F) evaluate cost-benefit trade-offs of commercial activities such as municipal development, farming, deforestation, over-harvesting, and mining;
 - (G) discuss the positive and negative influence of commonly held ethical beliefs on scientific practices such as methods used to increase food production or the existence of global warming;

- (H) discuss the impact of research and technology on social ethics and legal practices in situations such as the design of new buildings, recycling, or emission standards;
- (I) research the advantages and disadvantages of “going green” such as organic gardening/farming, natural methods of pest control, hydroponics, xeriscaping, energy efficient homes/appliances and hybrid cars;
- (J) analyze past and present local, state and national legislation including Texas automobile emissions regulations, the National Park Service Act, the Clean Air Act, the Clean Water Act, the Soil and Water Resources Conservation Act, and the Endangered Species Act, and
- (K) analyze past and present international treaties/protocols such as the environmental Antarctic Treaty System, Montreal Protocol, and Kyoto Protocol.

End of Environmental Systems

§112.46. Aquatic Science

- (a) **General requirements.** Students shall be awarded one credit for successful completion of this course. Required prerequisite: one unit of high school Biology. Recommended prerequisite: Chemistry or concurrent enrollment in Chemistry. This course is recommended for students in Grades 10, 11, or 12.
- (b) **Introduction.**
- (1) In Aquatic Science, students study the interactions of biotic and abiotic components in aquatic environments including impacts on aquatic systems. Investigations and field work in this course may emphasize fresh water or marine aspects of aquatic science depending primarily upon the natural resources available for study near the school. Students who successfully complete Aquatic Science will acquire knowledge about a variety of aquatic systems, conduct investigations and observations of aquatic environments, work collaboratively with peers, and develop critical thinking and problem solving skills.
- (2) **NATURE OF SCIENCE**
Science, as defined by the National Academy of Sciences, is the “use of evidence to construct testable explanations and predictions of natural phenomena, as well as the knowledge generated through this process.” This vast body of changing and increasing knowledge is described by physical, mathematical, and conceptual models. Students should know that some questions are outside the realm of science because they deal with phenomena that are not scientifically testable.
- (3) **SCIENTIFIC INQUIRY**
Scientific inquiry is the planned and deliberate investigation of the natural world. Scientific methods of investigation can be experimental, descriptive, or comparative. The method chosen should be appropriate to the question being explored.
- (4) **SCIENCE AND SOCIAL ETHICS**
Scientific decision-making is a way of answering questions about the natural world. Students should be able to distinguish between scientific decision-making methods and ethical/social decisions that involve the application of scientific information.
- (5) **SCIENTIFIC SYSTEMS**
A system is a collection of cycles, structures, and processes that interact. All systems have basic properties that can be described in terms of space, time, energy, and matter. Change and constancy occur in systems as patterns and can be observed, measured, and modeled. These patterns help to make predictions that can be scientifically tested. Students should analyze a system in terms of its components and how these components relate to each other, to the whole, and to the external environment.
- (c) **Knowledge and skills.**
- (1) **Scientific processes.** The student, for a minimum of 40% of instructional time, conducts field and laboratory investigations using safe, environmentally appropriate, and ethical practices. The student is expected to:

- (A) demonstrate safe practices during field and laboratory investigations, including chemical, electrical, and fire safety and safe handling of live and preserved organisms; and
 - (B) make wise choices in the use and conservation of resources and the disposal or recycling of materials.
- (2) **Scientific processes. The student will use scientific methods during field and laboratory investigations. The student is expected to:**
- (A) know that scientific hypotheses are tentative and testable statements that must be capable of being supported or not supported by observational evidence. Hypotheses of durable explanatory power which have been tested over a wide variety of conditions are incorporated into theories;
 - (B) know that scientific theories are based on natural and physical phenomena and are capable of being tested by multiple independent researchers. Unlike hypotheses, scientific theories are well-established and highly reliable, but they may be subject to change as new areas of science and new technologies are developed;
 - (C) distinguish between scientific hypotheses and scientific theories;
 - (D) plan and implement investigative procedures including asking questions, formulating testable hypotheses, and selecting, handling, and maintaining appropriate equipment and technology;
 - (E) collect data individually or collaboratively, make measurements with precision and accuracy, record values using appropriate units, and calculate statistically relevant quantities to describe data including mean, median, and range;
 - (F) demonstrate the use of course apparatuses, equipment, techniques and procedures
 - (G) organize, analyze, evaluate, build models, make inferences, and predict trends from data;
 - (H) perform calculations using dimensional analysis, significant digits, and scientific notation; and
 - (I) communicate valid conclusions using essential vocabulary and multiple modes of expression, such as such a lab reports, labeled drawings, graphic organizers, journals, summaries, oral reports and technology based reports.
- (3) Scientific processes. Uses critical thinking, scientific reasoning, and problem solving to make informed decisions within and outside the classroom. The student is expected to:
- (A) analyze and evaluate scientific explanations using empirical evidence, logical reasoning, and experimental and observational testing;

- (B) communicate and apply scientific information extracted from written text
 - (C) make responsible choices in selecting everyday products and services using scientific information;
 - (C) evaluate the impact of research and technology on scientific thought, society, and the environment;
 - (D) describe the connection between aquatic science and future careers; and
 - (E) explore and describe the history of aquatic science and contributions of scientists.
- (4) **Science concepts.** Students will know that aquatic environments are the product of Earth systems interactions. The student is expected to:
- (A) identify key features and characteristics of atmospheric, geological, hydrological, and biological systems as they relate to aquatic environments;
 - (B) apply systems thinking to the examination of aquatic environments including positive and negative feedback cycles; and
 - (C) collect and evaluate global environmental data using technology, such as maps, visualizations, satellite data, GPS/GIS, weather balloons, buoys, etc.
- (5) **Science concepts.** The student conducts long term studies on local aquatic environments. Local natural environments are to be preferred over artificial or virtual environments. The student is expected to:
- (A) evaluate data over a period of time from an established aquatic environment documenting seasonal changes and the behavior of organisms;
 - (B) collect baseline quantitative data including pH, salinity, temperature, mineral content, nitrogen compounds, and turbidity from an aquatic environment;
 - (C) analyze interrelationships among producers, consumers, and decomposers in a local aquatic ecosystem; and
 - (D) identify the interdependence of organisms in an aquatic environment such as in a pond, river, lake, ocean, or aquifer, and the biosphere.
- (6) **Science concepts.** The student knows the role of cycles in an aquatic environment. The student is expected to:
- (A) identify the role of carbon, nitrogen, water, and nutrient cycles, including upwellings, and turnovers in an aquatic environment; and
 - (B) examine the interrelationships between aquatic systems and climate & weather including El Niño and La Niña, currents, and hurricanes.

- (7) **Science concepts.** The student knows the origin and use of water in a watershed. The student is expected to:
- (A) identify sources and determine the amounts of water in a watershed including rainfall, groundwater and surface water;
 - (B) identify factors that contribute to how water flows through a watershed; and
 - (C) identify water quantity and quality in a local watershed.
- (8) **Science concepts.** The student knows that geological phenomena and fluid dynamics affect aquatic systems. The student is expected to:
- (A) demonstrate basic principles of fluid dynamics including hydrostatic pressure, density, salinity, and buoyancy;
 - (B) identify interrelationships between ocean currents, climates, and geologic features; and
 - (C) describe and explain fluid dynamics in an upwelling and lake turnover.
- (9) **Science concepts.** The student knows the types and components of aquatic ecosystems. The student is expected to:
- (A) differentiate among freshwater, brackish, and saltwater ecosystems;
 - (B) identify the major properties and components of different marine and freshwater life zones; and
 - (C) identify biological, chemical, geological, and physical components of an aquatic life zone as they relate to the organisms in it.
- (10) **Science concepts.** The student knows environmental adaptations of aquatic organisms. The student is expected to:
- (A) classify different aquatic organisms using tools, such as dichotomous keys;
 - (B) compare and describe how adaptations allow an organism to exist within an aquatic environment; and
 - (C) compare differences in adaptations of aquatic organisms to fresh water and marine environments.

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- (11) **Science concepts.** The student knows about the interdependence and interactions that occur in aquatic environments. The student is expected to:
- (A) identify how energy flows and matter cycles through both fresh water and salt water aquatic systems including food webs, chains, and pyramids; and
 - (B) evaluate the factors affecting aquatic population cycles.
- (12) **Science concepts.** The student understands how human activities impact aquatic environments. The student is expected to:
- (A) predict effects of chemical, organic, physical, and thermal changes from humans on the living and nonliving components of an aquatic ecosystem;
 - (B) analyze the cumulative impact of human population growth on an aquatic system;
 - (C) investigate the role of humans in unbalanced systems such as invasive species, fish farming, cultural eutrophication, or red tides;
 - (D) analyze and discuss how human activities such as fishing, transportation, dams, and recreation influence aquatic environments; and
 - (E) understand the impact of various laws and policies (such as The Endangered Species Act, right of capture laws, or Clean Water Act) on aquatic systems.

End of Aquatic Science

§112.48. Astronomy

- (a) **General requirements.** Students shall be awarded one credit for successful completion of this course. Suggested prerequisite: one unit of high school science. This course is recommended for students in Grades 11 or 12.
- (b) **Introduction.**
- (1) In Astronomy, students conduct field and laboratory investigations, use scientific methods, and make informed decisions using critical thinking and scientific problem solving. Students study the following topics: astronomy in civilization, patterns and objects in the sky, our place in space, the moon, reasons for the seasons, planets, the sun, stars, galaxies, cosmology, and space exploration. Students who successfully complete Astronomy will acquire knowledge within a conceptual framework, conduct observations of the sky, work collaboratively, and develop critical thinking skills.
- (2) **NATURE OF SCIENCE**
Science, as defined by the National Academy of Sciences, is the “use of evidence to construct testable explanations and predictions of natural phenomena, as well as the knowledge generated through this process.” This vast body of changing and increasing knowledge is described by physical, mathematical, and conceptual models. Students should know that some questions are outside the realm of science because they deal with phenomena that are not scientifically testable.
- (3) **SCIENTIFIC INQUIRY**
Scientific inquiry is the planned and deliberate investigation of the natural world. Scientific methods of investigation can be experimental, descriptive, or comparative. The method chosen should be appropriate to the question being asked.
- (4) **SCIENCE AND SOCIAL ETHICS**
Scientific decision-making is a way of answering questions about the natural world. Students should be able to distinguish between scientific decision-making methods and ethical/social decisions that involve the application of scientific information.
- (5) **SCIENTIFIC SYSTEMS**
A system is a collection of cycles, structures, and processes that interact. All systems have basic properties that can be described in terms of space, time, energy, and matter. Change and constancy occur in systems as patterns and can be observed, measured, and modeled. These patterns help to make predictions that can be scientifically tested. Students should analyze a system in terms of its components and how these components relate to each other, to the whole, and to the external environment.
- (c) **Knowledge and skills.**
- (1) **Scientific processes.** The student, for at least 40% of instructional time, conducts field and laboratory investigations using safe, environmentally appropriate, and ethical practices. The student is expected to:

- (A) demonstrate safe practices during field and laboratory investigations; and
 - (B) make wise choices in the use and conservation of resources and the disposal or recycling of materials.
- (2) **Scientific processes.** The student uses scientific methods during field and laboratory investigations. The student is expected to:
- (A) know that scientific hypotheses are tentative and testable statements that must be capable of being supported or not supported by observational evidence. Hypotheses of durable explanatory power which have been tested over a wide variety of conditions are incorporated into theories;
 - (B) know that scientific theories are based on natural and physical phenomena and are capable of being tested by multiple independent researchers. Unlike hypotheses, scientific theories are well-established and highly reliable explanations, but may be subject to change as new areas of science and new technologies are developed;
 - (C) distinguish between scientific hypotheses and scientific theories;
 - (D) plan and implement investigative procedures including making observations, asking questions, formulating testable hypotheses, and selecting equipment and technology;
 - (E) collect data and make measurements with accuracy and precision;
 - (F) organize, analyze, evaluate, make inferences, and predict trends from data including making new revised hypotheses when appropriate;
 - (G) communicate valid conclusions in writing, oral presentations, and through collaborative projects; and
 - (H) utilize astronomical technology such as telescopes, binoculars, sextants, computers and software.
- (3) **Scientific processes.** The student uses critical thinking, scientific reasoning and problem solving skills to make informed decisions within and outside the classroom. The student is expected to:
- (A) analyze and evaluate scientific explanations using empirical evidence, logical reasoning, and experimental and observational testing;
 - (B) draw inferences based on data related to promotional materials for products and services;

- (C) evaluate the impact of research on scientific thought, society, and the environment; and
- (D) describe the connection between astronomy and future careers.
- (4) **Science concepts.** The student recognizes the importance and uses of astronomy in civilization. The student is expected to:
 - (A) research and describe the use of astronomy in ancient civilizations such as the Egyptians, Mayan, Aztecs, Europeans, and the native Americans;
 - (B) research and describe the contributions of scientists to our changing understanding of astronomy including Ptolemy, Copernicus, Tycho Brahe, Kepler, Galileo, Newton, Einstein, and Hubble, and the contribution of women astronomers including Maria Mitchell and Henrietta Swan Leavitt;
 - (C) describe and explain the historical origins of the perceived patterns of constellations and the role of constellations in ancient and modern navigation; and
 - (D) explain the contributions of modern astronomy to today's society including the identification of potential asteroid/comet impact hazards and the Sun's effects on communication, navigation, and high-tech devices.
- (5) **Science concepts.** The student develops a familiarity with the sky. The student is expected to:
 - (A) observe and record the apparent movement of the sun and moon during the day;
 - (B) observe and record the apparent movement of the moon, planets, and stars in the nighttime sky; and
 - (C) recognize and identify constellations such as Ursa Major, Ursa Minor, Orion, Cassiopeia, and constellations of the zodiac.
- (6) **Science concepts.** The student knows our place in space. The student is expected to:
 - (A) compare and contrast the scale, size and distance of the Earth, Moon, and Sun system through the use of data and modeling;
 - (B) compare and contrast the scale, size and distance of objects in the solar system, such as the sun and planets through the use of data and modeling;
 - (C) examine the scale, size and distance of the stars, Milky Way, and other galaxies through the use of data and modeling;

- (D) relate apparent versus absolute magnitude to the distances of celestial objects; and
 - (E) demonstrate the use of units of measurement in astronomy including Astronomical Units and light years.
- (7) **Science concepts.** The student knows the role of the Moon in the Earth, Moon, and Sun system. The student is expected to:
- (A) observe and record data about lunar phases and use that information to model the Earth, Moon, and Sun system;
 - (B) illustrate the cause of lunar phases by showing positions of the moon relative to the earth and the sun for each phase including new moon, waxing crescent, first quarter, waxing gibbous, full moon, waning gibbous, third quarter, and waning crescent;
 - (C) identify and differentiate the causes of lunar and solar eclipses including differentiating between lunar phases and eclipses; and
 - (D) identify the effects of the moon on tides.
- (8) **Science concepts.** The student knows the reasons for the seasons. The student is expected to:
- (A) recognize that seasons are caused by the tilt of the earth's axis;
 - (B) explain how latitudinal position affects the length of day and night throughout the year;
 - (C) recognize that the angle of incidence of sunlight determines the concentration of solar energy received on the earth at a particular location; and
 - (D) examine the relationship of the seasons to equinoxes, solstices, the tropics and the equator.
- (9) **Science concepts.** The student knows that planets of different size, composition, and surface features orbit around the sun. The student is expected to:
- (A) compare and contrast the factors essential to life on Earth such as temperature, water, mass, and gases to conditions on other planets;
 - (B) compare the planets in terms of orbit, size, composition, rotation, atmosphere, natural satellites, and geological activity;
 - (C) relate the role of Newton's Law of Universal Gravitation to the motion of the planets around the sun and to the motion of natural and artificial satellites around the planets; and

- (D) explore the origins and significance of small solar system bodies including asteroids, comets, and Kuiper belt objects.
- (10) **Science concepts.** The student knows the role of the sun as the star in our solar system. The student is expected to:
- (A) identify the approximate mass, size, motion, temperature, structure, and composition of the sun;
 - (B) distinguish between nuclear fusion and nuclear fission, and identify the source of energy within the sun as nuclear fusion of hydrogen to helium;
 - (C) describe the eleven year solar cycle and the significance of sunspots; and
 - (D) analyze solar magnetic storm activity including coronal mass ejections, prominences, flares, and sunspots.
- (11) **Science concepts.** The student knows the characteristics and life cycle of stars. The student is expected to:
- (A) identify the characteristics of main sequence stars including surface temperature, age, relative size, and composition;
 - (B) characterize star formation in stellar nurseries from giant molecular clouds, to protostars, to the development of main sequence stars;
 - (C) evaluate the relationship between mass and fusion on the dying process and properties of stars;
 - (D) differentiate among the end states of stars including white dwarfs, neutron stars, and black holes;
 - (E) compare how the mass and gravity of a main sequence star will determine its end state as a white dwarf, neutron star, or black hole;
 - (F) relate the use of spectroscopy in obtaining physical data on celestial objects such as temperature, chemical composition, and relative motion; and
 - (G) utilize the Hertzsprung-Russell diagram to plot and examine the life cycle of stars from birth to death.
- (12) **Science concepts.** The student knows the variety and properties of galaxies. The student is expected to:
- (A) describe characteristics of galaxies;
 - (B) recognize the type, structure, and components of our Milky Way galaxy and location of our solar system within it; and

- (C) compare and contrast the different types of galaxies including spiral, elliptical, irregular, and dwarf.
- (13) **Science concepts.** The student knows the scientific theories of cosmology. The student is expected to:
- (A) research and describe the historical development of the Big Bang Theory including red shift, cosmic microwave background radiation, and other supporting evidence;
 - (B) research and describe current theories of the evolution of the universe including estimates for the age of the universe; and
 - (C) research and describe scientific hypotheses of the fate of the universe including open and closed universes and the role of dark matter and dark energy.
- (14) **Science concepts.** The student recognizes the benefits and challenges of space exploration to the study of the universe. The student is expected to:
- (A) identify and explain the contributions of human space flight, and future plans and challenges;
 - (B) recognize the advancement of knowledge in astronomy through robotic space flight;
 - (C) analyze the importance of ground-based technology in astronomical studies;
 - (D) recognize the importance of space telescopes to the collection of astronomical data across the electromagnetic spectrum; and
 - (E) demonstrate an awareness of new developments and discoveries in astronomy.

End of Astronomy

§112.XX. Earth and Space Science

- (a) **General requirements.** Students shall be awarded one credit for successful completion of this course. Required prerequisites: three units of science, one of which may be taken concurrently, and three units of mathematics, one of which may be taken concurrently. This course is recommended for students in grade 12 but may be taken by students in grade 11.
- (b) **Introduction.**
- (1) Earth and Space Science (ESS) is a capstone course designed to build on students' prior scientific and academic knowledge and skills to develop understanding of the Earth System in space and time.
 - (2) **NATURE OF SCIENCE**
Science, as defined by the National Academy of Sciences, is the "use of evidence to construct testable explanations and predictions of natural phenomena, as well as the knowledge generated through this process." This vast body of changing and increasing knowledge is described by physical, mathematical, and conceptual models. Students should know that some questions are outside the realm of science because they deal with phenomena that are not scientifically testable.
 - (3) **SCIENTIFIC INQUIRY**
Scientific inquiry is the planned and deliberate investigation of the natural world. Scientific methods of investigation can be experimental, descriptive, or comparative. The method chosen should be appropriate to the question being asked.
 - (4) **SCIENCE AND SOCIAL ETHICS**
Scientific decision-making is a way of answering questions about the natural world. Students should be able to distinguish between scientific decision-making methods and ethical/social decisions that involve the application of scientific information.
 - (5) **EARTH AND SPACE SCIENCE THEMES**
An Earth systems approach to the themes of Earth in Space and Time, Solid Earth, and Fluid Earth defined the selection and development of the concepts discussed below.
 - (A) **Earth in Space and Time.** Earth has a long, complex, and dynamic history. Advances in technologies continue to further our understanding of the origin, evolution, and properties of the Earth and planetary systems within a chronological framework. The origin and distribution of resources that sustain life on Earth are the result of interactions among Earth's subsystems over billions of years.

- (B) **Solid Earth.** The geosphere is a collection of complex, interacting, dynamic subsystems linking Earth’s interior to its surface. The geosphere is composed of materials that move between subsystems at various rates driven by the uneven distribution of thermal energy. These dynamic processes are responsible for the origin and distribution of resources as well as geologic hazards that impact society.
- (C) **Fluid Earth.** The fluid Earth consists of the hydrosphere, cryosphere, and atmosphere subsystems. These subsystems interact with the biosphere and geosphere resulting in complex biogeochemical and geochemical cycles. The global ocean is the thermal energy reservoir for surface processes and, through interactions with the atmosphere, influences climate. Understanding these interactions and cycles over time has implications for life on Earth.

(6) **EARTH AND SPACE SCIENCE STRANDS**

The course has three strands used throughout each of the three themes: systems, energy, and relevance.

- (A) **Systems.** A system is a collection of interacting physical, chemical and biological processes that involves the flow of matter and energy on different temporal and spatial scales. The Earth system is composed of interdependent and interacting subsystems of the geosphere, hydrosphere, atmosphere, cryosphere, and biosphere within a larger planetary and stellar system. Change and constancy occur in Earth’s system and can be observed, measured as patterns and cycles, and described or presented in models used to predict how the Earth system changes over time.
- (B) **Energy.** The uneven distribution of Earth’s internal and external thermal energy is the driving force for complex, dynamic and continuous interactions and cycles in Earth subsystems. These interactions are responsible for the movement of matter within and between the subsystems resulting in, for example, plate motions and ocean-atmosphere circulation.
- (C) **Relevance.** The interacting components of the Earth system change by both natural and human-influenced processes. Natural processes include hazards such as flooding, earthquakes, volcanoes, hurricanes, meteorite impacts, and climate change. Some human-influenced processes, such as pollution and unsustainable use of Earth’s natural resources, may damage the Earth system. Examples include climate change, soil erosion, air and water pollution, and biodiversity loss. The time scale of these changes and their impact on human society must be understood to make wise decisions concerning the use of the land, water, air, and natural resources. Proper stewardship of Earth will prevent unnecessary degradation and destruction of Earth’s subsystems and diminish detrimental impacts to individuals and society.

(c) **Knowledge and skills.**

- (1) **Scientific processes.** The student conducts laboratory and field investigations for at least 40% of instructional time using safe, environmentally appropriate, and ethical practices. The student is expected to:
- (A) demonstrate safe practices during laboratory and field investigations;
 - (B) make wise choices in the use and conservation of resources and the disposal or recycling of materials; and
 - (C) use the school's technology and information systems in a wise and ethical manner.
- (2) **Scientific processes.** The student uses scientific methods during laboratory and field investigations. The student is expected to:
- (A) know that scientific hypotheses are tentative and testable statements that must be capable of being supported or not supported by observational evidence. Hypotheses of durable explanatory power which have been tested over a wide variety of conditions are incorporated into theories.
 - (B) know that scientific theories are based on natural and physical phenomena and are capable of being tested by multiple independent researchers. Unlike hypotheses, scientific theories are well-established and highly reliable explanations, but may be subject to change as new areas of science and new technologies are developed.
 - (C) distinguish between scientific hypotheses and scientific theories.
 - (D) demonstrate the use of course equipment, techniques, and procedures, including computers and web-based computer applications;
 - (E) use a wide variety of additional course apparatuses, equipment, techniques, and procedures as appropriate, such as satellite imagery and other remote sensing data, Geographic Information Systems (GIS), Global Positioning System (GPS), scientific probes, microscopes, telescopes, modern video and image libraries, weather station, fossil and rock kits, bar magnets, coiled springs, wave simulators, tectonic plate models, and planetary globes;
 - (F) organize, analyze, evaluate, make inferences, and predict trends from data;
 - (G) use mathematical procedures such as algebra, statistics, scientific notation, and significant figures to analyze data using SI units; and
 - (H) communicate valid conclusions supported by the data using several formats, such as technical reports, lab reports, labeled drawings, graphic organizers, journals, presentations, and technical posters.

- (3) **Scientific processes.** The student uses critical thinking, scientific reasoning, and problem solving to make informed decisions within and outside the classroom. The student is expected to:
- (A) analyze and evaluate scientific explanations using empirical evidence, logical reasoning, and experimental and observational testing;
 - (B) analyze, evaluate, communicate and apply scientific information extracted from written text such as current events, news reports, published journal articles, and marketing materials;
 - (C) evaluate the impact of research on scientific thought, society, and public policy;
 - (D) explore careers and collaboration among scientists in Earth and space sciences;
 - (E) learn and understand the contributions of scientists to the historical development of the Earth and space sciences; and
- (4) **Earth in space and time.** The student knows how Earth-based and space-based astronomical observations reveal the structure, scale, composition, origin, and history of the universe. The student is expected to:
- (A) evaluate the evidence concerning the Big Bang model, such as red shift and cosmic microwave background radiation, and the concept of an expanding universe that originated about 14 billion years ago;
 - (B) explain how the Sun and other stars transform matter into energy through nuclear fusion; and
 - (C) investigate the process by which a supernova can lead to the formation of successive generation stars and planets.
- (5) **Earth in space and time.** The student knows that Earth's place in the solar system is explained by the solar nebular accretionary disk model. The student is expected to:
- (A) analyze how gravitational condensation of solar nebular gas and dust can lead to the accretion of planetesimals and protoplanets;
 - (B) investigate sources of heat, including kinetic heat of impact accretion, gravitational compression, and radioactive decay, which allows protoplanet differentiation into layers;
 - (C) contrast the characteristics of comets, asteroids, meteoroids, and their positions in the solar system including the orbital regions of the terrestrial planets, the asteroid belt, gas giants, Kuiper Belt, and Oort Cloud;
 - (D) explore the historical and current hypotheses for the origin of the Moon, including the collision of Earth with a Mars-sized planetesimal;

- (E) compare terrestrial planets to gas giant planets in the solar system, including structure, composition, size, density, orbit, surface features, tectonic activity, temperature, and suitability for life; and
 - (F) compare extra-solar planets with planets in our solar system, and describe how such planets are detected.
- (6) **Earth in space and time.** The student knows the evidence for how Earth's atmospheres, hydrosphere, and geosphere formed and changed through time. The student is expected to:
- (A) analyze the changes of Earth's atmosphere through time from the original hydrogen-helium atmosphere, the carbon dioxide-water vapor-methane atmosphere, and the current nitrogen-oxygen atmosphere;
 - (B) evaluate the role of volcanic outgassing and impact of water-bearing comets in developing Earth's atmosphere and hydrosphere;
 - (C) investigate how the formation of atmospheric oxygen and the ozone layer impacted the formation of the geosphere and biosphere; and
 - (D) evaluate how Earth's cooling led to tectonic activity, resulting in continents and ocean basins.
- (7) **Earth in space and time.** The student knows that scientific dating methods of fossils and rock sequences are used to construct a chronology of Earth's history expressed in the geologic time scale. The student is expected to:
- (A) evaluate relative dating methods using original horizontality, rock superposition, lateral continuity, cross-cutting relationships, unconformities, index fossils, and biozones based on fossil succession to determine chronological order;
 - (B) apply radiometric dating methods that can be used to calculate the ages of igneous rocks from Earth and Moon, and meteorites; and
 - (C) understand how multiple dating methods are used to construct the geologic time scale, which represents Earth's approximate 4.6 billion year history;
- (8) **Earth in space and time.** The student knows that fossils provide evidence for geological and biological evolution. Students are expected to:
- (A) evaluate a variety of fossil types, transitional fossils, fossil lineages, and significant fossil deposits with regard to their appearance, completeness, and rate and diversity of evolution;
 - (B) explain how sedimentation, fossilization, and speciation affect the degree of completeness of the fossil record; and

- (C) evaluate the significance of the terminal Permian and Cretaceous mass extinction events, including adaptive radiations of organisms after the events.
- (9) **Solid Earth.** The student knows Earth’s interior is differentiated chemically, physically and thermally. The student is expected to:
- (A) evaluate heat transfer through Earth’s subsystems by radiation, convection, and conduction; include its role in plate tectonics, volcanism, ocean circulation, weather, and climate;
 - (B) examine the chemical, physical, and thermal structure of Earth’s crust, mantle, and core including the lithosphere and asthenosphere;
 - (C) explain how scientists use geophysical methods, such as seismic wave analysis, gravity, and magnetism to interpret Earth’s structure; and
 - (D) describe the formation and structure of the Earth’s magnetic field including its interaction with charged solar particles to form the Van Allen belts and auroras.
- (10) **Solid Earth.** The student knows that plate tectonics is the global mechanism for major geologic processes and that heat transfer, governed by the principles of thermodynamics, is the driving force. The student is expected to:
- (A) investigate how new conceptual interpretations of data and innovative geophysical technologies led to the current theory of plate tectonics;
 - (B) describe how heat and rock composition affect density within Earth’s interior and how density influences the development and motion of Earth’s tectonic plates;
 - (C) explain how plate tectonics accounts for geologic processes and features, including sea floor spreading, ocean ridges and rift valleys, subduction zones, earthquakes, volcanoes, mountain ranges, hot spots, and hydrothermal vents;
 - (D) calculate the motion history of tectonic plates using equations relating rate, time, and distance to predict future motions, locations, and resulting geologic features;
 - (E) distinguish the location, type, and relative motion of convergent, divergent, and transform plate boundaries using evidence from the distribution of earthquakes and volcanoes; and
 - (F) evaluate the role of plate tectonics with respect to long-term global changes in Earth’s subsystems, such as continental buildup, glaciation, sea level fluctuations, mass extinctions, and climate change.

- (11) **Solid Earth.** The student knows that the geosphere continuously changes over a range of time scales involving dynamic and complex interactions among Earth's subsystems. The student is expected to:
- (A) compare the roles of erosion, deposition, and igneous activity by lava, water, wind, ice, and gravity in constantly reshaping Earth's surface;
 - (B) explain how plate tectonics accounts for geologic surface processes and features including folds, faults, sedimentary basin formation, mountain building, and continental accretion;
 - (C) analyze changes in continental plate configurations, such as Pangaea, and their impact on the biosphere, atmosphere and hydrosphere through time;
 - (D) interpret Earth surface features using a variety of methods such as satellite imagery, aerial photography, topographic and geologic maps using appropriate technologies; and
 - (E) evaluate the impact of changes in Earth's subsystems on humans, such as earthquakes, tsunamis, volcanic eruptions, hurricanes, flooding, and storm surges; and the impact of humans on Earth's subsystems, such as population growth, fossil fuel burning, and use of fresh water.
- (12) **Solid Earth.** The student knows that Earth contains energy, water, mineral, and rock resources, and that use of these resources impacts Earth's subsystems. The student is expected to:
- (A) evaluate how the use of energy, water, mineral, and rock resources affects Earth's subsystems;
 - (B) describe the formation of fossil fuels, including petroleum and coal;
 - (C) discriminate between renewable and nonrenewable resources based upon rate of formation and use;
 - (D) analyze the economics of resources from discovery to disposal, including technological advances, resource type, concentration and location, waste disposal and recycling, and environmental costs; and
 - (E) explore careers that involve the exploration, extraction, production, use, and disposal of Earth's resources.
- (13) **Fluid Earth.** The student knows that the fluid Earth is composed of the hydrosphere, cryosphere, and atmosphere subsystems that interact on various time scales with the biosphere and geosphere. The student is expected to:
- (A) quantify the components and fluxes within the hydrosphere, such as changes in polar ice caps and glaciers, salt water incursions, and groundwater levels in response to precipitation events or excessive pumping;

- (B) analyze how global ocean circulation is the result of wind, tides, the Coriolis effect, water density differences, and the shape of the ocean basins;
 - (C) analyze the empirical relationship between the emissions of carbon dioxide, atmospheric carbon dioxide levels, and the average global temperature trends over the past 150 years;
 - (D) discuss mechanisms and causes, such as selective absorbers, major volcanic eruptions, solar luminance, giant meteorite impacts, and human activities that result in significant changes in Earth's climate;
 - (E) investigate the causes and history of eustatic sea-level changes that result in transgressive and regressive sedimentary sequences; and
 - (F) discuss scientific hypotheses for the origin of life by abiotic chemical processes in an aqueous environment through complex geochemical cycles.
- (14) **Fluid Earth.** The student knows that Earth's global ocean stores solar energy, and is a major driving force for weather and climate through complex atmospheric interactions. The student is expected to:
- (A) analyze the uneven distribution of solar energy on Earth's surface, including differences in atmospheric transparency, surface albedo, Earth's tilt, duration of insolation, and differences in atmospheric and surface absorption of energy;
 - (B) investigate how the atmosphere is heated from Earth's surface due to absorption of solar energy, which is re-radiated as thermal energy and trapped by selective absorbers; and
 - (C) explain how thermal energy transfer between the ocean and atmosphere drives surface currents, thermohaline currents, and evaporation that influences climate.
- (15) **Fluid Earth.** The student knows that interactions among Earth's five subsystems influence climate and resource availability, which affect Earth's habitability. The student is expected to:
- (A) describe how changing surface-ocean conditions, including El Nino-Southern Oscillation, affect global weather and climate patterns;
 - (B) investigate evidence, such as ice cores, glacial striations, and fossils, for climate variability and its use in developing computer models to explain present and predict future climates;
 - (C) quantify the dynamics of surface and groundwater movement, such as recharge, discharge, evapotranspiration, storage, residence time, and sustainability;
 - (D) explain the global carbon cycle, including how carbon exists in different forms within the five subsystems and how these forms affect life; and

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- (E) analyze recent global ocean temperature data to predict the consequences of changing ocean temperature on evaporation, sea level, algal growth, coral bleaching, hurricane intensity, and biodiversity.

End of Earth and Space Science

Proposed Recommendations

§112.XX. Engineering Design and Problem Solving

- (a) **General requirements.** Students shall be awarded one credit for successful completion of this course. Prerequisites: Geometry, Algebra II, Chemistry, and Physics. This course is recommended for students in Grades 11-12.
- (b) **Introduction.**
- (1) Engineering design is the creative process of solving problems by identifying needs and then devising solutions. This solution may be a product, technique, structure, process, or many other things depending on the problem. Science aims to understand the natural world, while engineering seeks to shape this world to meet human needs and wants. Engineering design takes into consideration limiting factors – or "design under constraint."
 - (2) Engineering Design and Problem Solving reinforces and integrates skills learned in previous mathematics and science courses. This course emphasizes solving problems -- moving from well-defined toward more open-ended -- with real world application. Students apply critical thinking skills to justify a solution from multiple design options. Additionally, the course promotes interest in, and understanding of, career opportunities in engineering.
 - (3) This course is intended to stimulate students' ingenuity, intellectual talents, and practical skills in devising solutions to engineering design problems. Students use the engineering design process cycle to investigate, design, plan, create, and evaluate solutions. At the same time, this course fosters awareness of the social and ethical implications of technological development.
 - (4) Importantly, this course will be directed largely by the resources of local school districts and the skills and talents of the instructors, influenced by the interests of the students and current events. The specific knowledge and skills below give considerable latitude to the districts and teachers in formulating or adopting the curriculum and the materials used in the course. It is expected, however, that the course be rigorous and go into considerable depth within a major engineering discipline related to the project. Various engineering disciplines address a broad spectrum of design problems utilizing specific concepts from the sciences and mathematics to derive a solution. However, the design process and problem-solving that are at the core of these standards are inherent to all engineering disciplines.
- (c) **Knowledge and skills.**
- (1) The student, for at least 40% of instructional time, conducts engineering field and laboratory activities using safe, environmentally appropriate, and ethical practices. The student is expected to:
 - (A) demonstrate safe practices during engineering field and laboratory activities; and
 - (B) make informed choices in the use and conservation of resources, recycling of materials, and the safe and legal disposal of materials.

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- (2) The student applies knowledge of science and mathematics and the tools of technology to solve engineering design problems. The student is expected to:
- (A) apply scientific processes and concepts outlined in the biology, chemistry, or physics TEKS relevant to engineering design problems;
 - (B) apply concepts, procedures, and functions outlined in the Algebra I, Geometry, or Algebra II TEKS relevant to engineering design problems;
 - (C) select appropriate mathematical models to develop solutions to engineering design problems;
 - (D) integrate advanced math and science skills as necessary to develop solutions to engineering design problems;
 - (E) judge the reasonableness of mathematical models and solutions;
 - (F) investigate and apply relevant chemical, mechanical, biological, electrical, and physical properties of materials to engineering design problems;
 - (G) identify the inputs, processes, outputs, control and feedback associated with open and closed systems;
 - (H) describe the difference between open-loop and closed-loop control systems;
 - (I) make measurements and specify tolerances with minimum necessary accuracy and precision;
 - (J) use appropriate measurement systems, including customary and International System (SI) units; and
 - (K) use conversions between measurement systems to solve real-world problems.
- (3) The student communicates through written documents, presentations, and graphic representations using the tools and techniques of professional engineers. The student is expected to:
- (A) communicate visually by sketching and creating technical drawings utilizing established engineering graphic tools, techniques, and standards;
 - (B) read and comprehend technical documents including specifications and procedures;
 - (C) prepare written documents such as memorandums, e-mails, design proposals, procedural directions, letters, and technical reports using the formatting and terminology conventions of technical documentation;
 - (D) organize information for visual display and analysis using appropriate formats for various audiences including but not limited to graphs and tables;

- (E) evaluate the quality and relevance of sources and cite appropriately; and
 - (F) defend a design solution in a presentation.
- (4) The student recognizes the history, development, and practices of the engineering professions. The student is expected to:
- (A) identify and describe career options, working conditions, earnings, and educational requirements of various engineering disciplines, such as those listed by the Texas Board of Professional Engineers;
 - (B) recognize that engineers are guided by established codes emphasizing high ethical standards;
 - (C) explore the differences, similarities, and interactions among engineers, scientists, and mathematicians;
 - (D) describe how technology has evolved in the field of engineering and consider how it will continue to be a useful tool in solving engineering problems;
 - (E) discuss the history and importance of engineering innovation on the American economy and quality of life; and
 - (F) describe the importance of patents and the protection of intellectual property rights.
- (5) The student creates justifiable solutions to open-ended problems using engineering design practices and processes. The student is expected to:
- (A) identify and define an engineering problem;
 - (B) formulate goals, objectives, and requirements to solve an engineering problem;
 - (C) determine the design parameters associated with an engineering problem, such as materials, personnel, resources, funding, manufacturability, feasibility, and time;
 - (D) establish and evaluate constraints pertaining to a problem, including but not limited to health, safety, social, environmental, ethical, political, regulatory, and legal;
 - (E) identify or create alternative solutions to a problem using a variety of techniques, such as brainstorming, reverse engineering, and researching engineered and natural solutions;
 - (F) test and evaluate proposed solutions using methods such as models, prototypes, mock-ups, simulations, critical design review, statistical analysis, or experiments;
 - (G) apply structured techniques to select and justify a preferred solution to a problem, such as a decision tree, design matrix, or cost-benefits analysis;

- (H) predict performance, failure modes, and reliability of a design solution; and
 - (I) prepare a project report that clearly documents the designs, decisions, and activities during each phase of the engineering design process.
- (6) The student manages an engineering design project. The student is expected to:
- (A) participate in the design and implementation of a real or simulated engineering project;
 - (B) develop a plan and time-line for completion of a project;
 - (C) work in teams and share responsibilities, acknowledging, encouraging, and valuing contributions of all team members;
 - (D) compare and contrast the roles of a team leader and other team responsibilities;
 - (E) identify and manage the resources needed to complete a project;
 - (F) utilize a budget to determine effective strategies to meet cost constraints;
 - (G) create a risk assessment for an engineering design project;
 - (H) analyze and critique the results of an engineering design project; and
 - (I) maintain an engineering notebook that chronicles work, such as ideas, concepts, inventions, sketches, and experiments.

End of Engineering Design and Problem Solving
