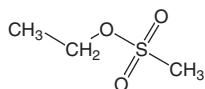


Ethyl Methanesulfonate

CAS No. 62-50-0

Reasonably anticipated to be a human carcinogen
First Listed in the *Third Annual Report on Carcinogens* (1983)



Carcinogenicity

Ethyl methanesulfonate is *reasonably anticipated to be a human carcinogen* based on sufficient evidence of carcinogenicity in experimental animals (IARC 1974, 1987). When administered to newborn mice as a single subcutaneous injection or five daily injections in water, arachis oil, or aqueous gelatin, ethyl methanesulfonate induced adenomas or adenocarcinomas of the lung. When administered as a single intraperitoneal injection, ethyl methanesulfonate induced lung tumors in male mice and lung adenomas in mice of both sexes. Three intraperitoneal injections of ethyl methanesulfonate in arachis oil induced lung and kidney tumors in male mice. In a similar study, ethyl methanesulfonate induced renal carcinomas in female rats and a variety of benign and malignant tumors, including lung carcinomas, in rats of both sexes. When administered as a single intraperitoneal injection, ethyl methanesulfonate induced malignant epithelial and mesenchymal tumors of the kidney in rats of both sexes. When ethyl methanesulfonate was administered as a single intraperitoneal injection to mice of both sexes following a single intraperitoneal injection of dimethylnitrosamine, an additive effect in relation to the incidence of malignant kidney tumors was produced by the combination when compared to the effect of dimethylnitrosamine alone. In a study not reviewed by IARC, when administered in the drinking water, ethyl methanesulfonate induced multiple moderately differentiated mammary adenocarcinomas in young rats of both sexes and mature female rats (Ueo *et al.* 1981).

No adequate data were available to evaluate the carcinogenicity of ethyl methanesulfonate in humans (IARC 1974, 1987).

Properties

Ethyl methanesulfonate is a colorless liquid with a boiling point of 213 to 213.5°C. When heated to decomposition, ethyl methanesulfonate emits toxic fumes of sulfur oxides (Sax and Lewis 1987).

Use

Ethyl methanesulfonate is used experimentally as a mutagen, teratogen, and brain carcinogen and as a research chemical (IARC 1974, HSDB 2000, Merck 1989). It may also be used as a chemosterilant. The monoesters of methanesulfonic acid may be reversible insect and mammalian pest chemosterilants, as well as possible human male contraceptives (Jackson 1964, IARC 1974).

Production

Ethyl methanesulfonate is not produced commercially in the United States, and no import and export data were reported (HSDB 2000, USITC 1986-1991, 1993-1995). Production of ethyl methanesulfonate is likely to be limited to research purposes (IARC 1974). Nineteen U.S. suppliers of the compound were currently listed in Chem Sources (2001).

Exposure

The National Occupational Exposure Survey (1981-1983) estimated that potential occupational exposure to ethyl methanesulfonate is 971 total workers, including 448 female workers (NIOSH 1984). Occupational exposure should be limited to laboratory research and housekeeping personnel. Potential exposure for the general population exists if ethyl

methanesulfonate is approved for the production of chemosterilants for insects and mammalian pests or as a human male contraceptive. Ethyl methanesulfonate is not known to occur in nature (IARC 1974).

Regulations

EPA

Comprehensive Environmental Response, Compensation, and Liability Act

Reportable Quantity (RQ) = 1 lb

Resource Conservation and Recovery Act

Listed Hazardous Waste: Waste codes in which listing is based wholly or partly on substance - U119

Listed as a Hazardous Constituent of Waste

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