

# EDGEWOOD

CHEMICAL BIOLOGICAL CENTER

U.S. ARMY SOLDIER AND BIOLOGICAL CHEMICAL COMMAND

ECBC-TR-229

# IDENTIFICATION, PURIFICATION, AND PARTIAL CHARACTERIZATION OF THE GV-DEGRADING ENZYME FROM ATCC # 29660 ALTEROMONAS UNDINA

Steven P. Harvey Tu-chen Cheng

**RESEARCH AND TECHNOLOGY DIRECTORATE** 

February 2002

Approved for public release; distribution is unlimited.

20020429 050

Aberdeen Proving Ground, MD 21010-5424

### Disclaimer

The findings in this report are not to be construed as an official Department of the Army position unless so designated by other authorizing documents.

<\_

REPORT DOCUMENTATION PAGE			Form Approved OMB No. 0704-0188	
Public reporting burden for this collec searching existing data sources, gath comments regarding this burden estir Washington Headquarters Services, I 22202-4302, and to the Office of Man	tion of information is estimated to ave ering and maintaining the data neede nate or any other aspect of this collec Directorate for Information Operations agement and Budget, Paperwork Red	rage 1 hour per response, d, and completing and revi- tion of information, includi and Reports, 1215 Jeffer duction Project (0704-018)	including the time for reviewing instructions, iewing the collection of information. Send ng suggestions for reducing this burden, to son Davis Highway, Suite 1204, Arlington, VA 8), Washington, DC 20503.	
1. AGENCY USE ONLY (Leave Blank) 2. REPORT DATE 3. REPORT TYPE AND D   2002 February Final; 00 Oct -			DATES COVERED – 01 Sep	
4. TITLE AND SUBTITLE			5. FUNDING NUMBERS	
Identification, Purification, GV-Degrading Enzyme from	and Partial Characterizatio n ATCC # 29660 <i>Alterom</i> o	n of the onas undina	P-106013	
6. AUTHOR(S) Harvey, Steven P., and Ch	eng, Tu-chen		_	
7. PERFORMING ORGANIZATION NAM	E(S) AND ADDRESS(ES)		8. PERFORMING ORGANIZATION REPORT NUMBER	
DIR, ECBC, ATTN: AMSS	ECBC-TR-229			
9. SPONSORING/MONITORING AGENC	Y NAME(S) AND ADDRESS(ES)		10. SPONSORING/MONITORING AGENCY REPORT NUMBER	
11. SUPPLEMENTARY NOTES				
12a. DISTRIBUTION/AVAILABILITY STA	TEMENT		12b. DISTRIBUTION CODE	
Approved for public release	; distribution is unlimited.			
The GV (2-dialkylaminoalky to G and V-type nerve ager or organophosphorus hydro <i>Alteromonas</i> strains posses one of these strains, <i>A. unc</i> Q Sepharose anion exchan from that of <i>A. undina</i> OPA polyacrylamide gel electrop ~20kDa. To our knowledge	I N,N-dialkylphosphonamic ats and is not catalyzed by plase (OPH) enzymes. We assing catalytic activity using <i>dina,</i> has been purified to h ge chromatography. The A following the Q Sepharo horesis of the GV-hydrolyz e, this is the first report of e	iofluoridate) nerve either organophos have screened ar g a GV compound omogeneity by am activity of GV-hydro se column chroma ing enzyme fractio enzymatic detoxific	agent has a toxicity intermediate phorus acid anhydrolases (OPAA) nd identified a number of as substrate. The enzyme from monium sulfate fractionation and olyzing enzyme peak is distinct tography. The SDS- n revealed a single polypeptide of ation of GV.	
14. SUBJECT TERMS Enzymes Hydrolysis	Purification GV		15. NUMBER OF PAGES 18	
			16. PRICE CODE	
17. SECURITY CLASSIFICATION OF REPORT UNCLASSIFIED	18. SECURITY CLASSIFICATION OF THIS PAGE UNCLASSIFIED	IS. SECURITY CLASSIFICA OF ABSTRACT UNCLASSIFIF	TION 20. LIMITATION OF ABSTRACT	
NSN 7540-01-280-5500			Standard Form 298 (Rev. 2-89) Prescribed by ANSI Std. Z39-18 298-102	

.

3

.

.

<u>}</u>

.

.

Blank

e

.

.

٢,

### PREFACE

The work described in this report was authorized under Project No. 106013, Tech Base Program. This work was started in October 2000 and completed in September 2001.

The use of either trade or manufacturers' names in this report does not constitute an official endorsement of any commercial products. This report may not be cited for purposes of advertisement.

This report has been approved for public release. Registered users should request additional copies from the Defense Technical Information Center; unregistered users should direct such requests to the National Technical Information Center.

Blank

1

.

.

¢

# CONTENTS

1.	INTRODUCTION	
2.	MATERIALS AND METHODS8	
2.1. 2.2.	Growth of Bacterial Strains8 Enzyme Assays8	
3.	RESULTS AND DISCUSSION8	
3.1. 3.2. 3.3. 3.4. 3.4.1 3.4.2	Spontaneous Hydrolysis - Comparison to G-type Nerve Agents8 Screening of Bacterial Strains and Enzymes for GV Activity9 Purification of GVH	
4.	CONCLUSIONS	
	LITERATURE CITED	÷

# FIGURES

1.	Structure of 2-dimethylaminoethyl N, N-dimethylphosphonamidofluoridate	7
2.	Reaction Scheme for the Hydrolytic Defluorination of GV	9
3.	Purification of GVH on Q Sepharose Column with 0 - 1.0 M NaCl Gradient	11
4.	Purification of GVH on Second Q Sepharose Column	12
5.	SDS-PAGE of Fractions from GV Degrading Enzyme Purification	13
6.	pH Profile of GVH	14
7.	Temperature Profile of GVH	15

# TABLES

1.	Spontaneous Hydrolysis Rate of 0.01 M Substrate in 50 mM Bis-tris-propane (pH 7.2) at 25 °C	9
2.	GV Activity (nmol/min/mL) of Lysates	10
3.	Stepwise Elution of GVH Activity	11

# IDENTIFICATION, PURIFICATION, AND PARTIAL CHARACTERIZATION OF THE GV-DEGRADING ENZYME FROM ATCC # 29660 ALTEROMONAS UNDINA

### 1. INTRODUCTION

The enzymes organophoshorus acid anhydrolase (OPAA) from *Alteromonas sp.* JD6.5 and organophosphorus hydrolase (OPH) from *Pseudomonas diminita* have been shown to catalyze the hydrolysis of a number of toxic organophosphorus (OP) compounds including several G-type chemical nerve agents.<sup>1-12</sup> The OPH also has catalytic activity against V-type nerve agents.<sup>13</sup> Both these enzymes have been cloned into *Escherichia coli* and can be produced in large quantities. The OPH enzymatic activity for specific substrates has also been increased by genetic manipulation of the cloned gene.<sup>14</sup> The catalytic activity of these enzymes against nerve agents provides considerable potential for decontamination and detoxification of toxic OPs, and/or *in vivo* prophylaxis.

There is another important class of potential threat agents for which catalytic decontamination has not previously been demonstrated. The GV compounds have the generalized structure of 2-dialkylaminoalkyl N,N-dialkylphosphonamidofluoridate. An example is 2-dimethylaminoethyl N,N-dimethylphosphonamidofluoridate(Figure 1). The GV compounds are generally much more toxic than G agents and are not included in the Chemical Weapons Convention schedules of compounds.



Figure 1. Structure of 2-dimethylaminoethyl N, N-dimethylphosphonamidofluoridate

Neither OPAAs from *A. sp.* JD6.5 or *A. haloplanktis*, nor OPH have any detectable catalytic activity against compound 1 (unpublished data). To date, there have been no reports on the enzymatic detoxification of GV compounds.

The GV poisoning is also refractory to traditional atropine/ reactivator treatments.<sup>15</sup> Enzymes, however, may offer a successful treatment for GV poisoning. *In vivo* circulating enzymes have already been demonstrated to confer protection against toxic OP agents in mice. The LD<sub>50</sub> levels of protection exceed those obtained with atropine/reactivator.<sup>16</sup>

Finally, the gene encoding GV enzymes may offer a good basis for protein engineering for improving VX and G-agent activity due to the similarity of GV and VX structures.

### 2. MATERIALS AND METHODS

### 2.1. <u>Growth of Bacterial Strains</u>.

Bacteria strains were grown in Instant Ocean Medium consisting of 38 g of Instant Ocean (Aquarium Systems, Mentor, OH), 5 g of proteose peptone, and 1 g of yeast extract per liter at 30 °C for 16-20 hr.

### 2.2. Enzyme Assays.

Enzyme assays were conducted with a fluoride electrode attached to a Corning 355 pH/ion analyzer. Fluoride release rates were followed until a stable release rate was observed over a period of several minutes. Reactions were conducted in a temperature-controlled vessel in a total volume of 2.5 mL. Background hydrolysis rates (Table 1) were measured to provide a comparison of the relative stability of the different substrates in an aqueous matrix. Background hydrolysis rates were measured by adding neat substrate directly to 2.5 mL buffer solution to a final concentration of 0.01 M. For assaying enzymatic activity, substrates were diluted into isopropanol for the OPAA assays and into methanol for the OPH assays. The reaction medium contained 50 mM Bis-tris-propane (BTP) pH 7.2, 0.1 mM MnCl<sub>2</sub> for OPAA assays and 0.05 mM CoCl<sub>2</sub> for OPH assays, 3 mM GV compound (1), and 1-5 µL of enzyme sample in a total volume of 2.5 mL. For each assay, the substrate was added last. After a 1-4 min equilibration period, a background hydrolysis rate was measured for 4 min. An identical reaction mixture was then monitored in the presence of enzyme and the background values were subtracted. Assays conditions were adjusted to ensure that <10% of the total substrate was consumed in the assay.

### 3. RESULTS AND DISCUSSION

### 3.1. <u>Spontaneous Hydrolysis – Comparison to G-type Nerve Agents.</u>

The GV is relatively stable in aqueous solution, as compared to GB and GD. Table 1 illustrates the spontaneous hydrolysis (defluorination) rate of 0.01 M GB, GD and GV at pH 7.2 and 25 °C.

Table 1. Spontaneous Hydrolysis Rate of 0.01 M Substrate in 50 mM Bis-tris-propane (pH 7.2) at 25 °C

Substrate	Rate
	(µmol/min)
GB	58
GD	35
GV	4.0

### 3.2. <u>Screening of Bacterial Strains and Enzymes for GV Activity</u>.

Because no enzymes have previously been reported to catalyze the hydrolysis of GV, a screening was conducted to search potential sources. The reaction scheme for GV detoxification is illustrated in Figure 2. Fluoride is the leading group (as HF) and defluorination yields a hydroxylated product (GV-OH). Reactions consisting of 3 mM GF, buffer, and crude cell lysate were monitored by fluoride electrode for the rate of increase in the fluoride ion concentration.



Figure 2. Reaction Scheme for the Hydrolytic Defluorination of GV

Twelve Alteromonas strains were used because they have been shown to possess activity against different OP neurotoxins. These Alteromonas strains were grown in Instant Ocean Medium, lysed twice in a French press, centrifuged and the lysates obtained were assayed for activity against GV compound (1). Assays were conducted in 50 mM Bis-tris-propane (pH 7.2) at 25 °C. Results are shown in Table 2.

		GV			GV
ATCC #	Strain	Activity	ATCC #	Strain	Activity
27127	A. haloplanktis B	169	14393	A. haloplanktis Z	77
23821	A. haloplanktis C	24		A. espejiana	132
19648	A. haloplanktis J	186	33043	A. sp. G	192
	A. haloplanktis M	444	33524	A. sp. M	238
33492	A. luteoviolaceans	5	29332	A. sp. P	40
35358	A. undina	155	19375	A. niger	29

Table 2. GV Activity (nmol/min/mL) of Lysates

In addition to the lysates shown in Table 2, three different purified enzymes (OPAA cloned from *A. sp.* JD6.5 and *A. haloplanktis*, and OPH from *P. diminita*) were also tested. Previously, all three of these enzymes had been demonstrated to possess significant activity against several G-type agents.<sup>7,9,10,11</sup> However, none of the three had any detectable GV activity under the conditions used for the assays (MATERIALS AND METHODS). Limits of detection for the assays were at least in the low nmol/min/mg range.

It should be noted that enzymes with higher specific activity are sometimes obtained from strains with lower crude lysate activity. In addition, the crude lysate enzymatic activity is usually dependent on growth conditions. Based on previous enzyme purification experience with *A. undina* # 35358 strain, it was selected for further purification of its GV-hydrolyzing enzyme(s), designated as GVH.

### 3.3. <u>Purification of GVH</u>.

A single colony was grown overnight in Instant Ocean Medium (pH 7.0) at room temperature in a 100 mL shake flask. The overnight cultures were transferred to a BioFlow 3000 bioreactor (New Brunswick Scientific Co) with a 6 L working volume of the same medium and cultivated for 7.5 hr at a pH of 7. A total of 53 g wet cells were harvested and resuspended in 200 mL 10 mM bis-tris-propane pH 7.2. After cells were broken in a French pressure cell (SLM Aminco), cellular debris was removed by centrifugation at 5,000 xg. The cell lysate, which contained the GVH activity, was further purified by 40-65% ammonium sulfate fractionation. The pellet was resuspended in a minimal volume of 10 mM bis-tris-propane pH 7.2 and dialyzed against several changes of the same buffer. The dialyzed sample was then applied to a Q Sepharose (Pharmacia) column previously equilibrated with 10 mM bis-tris-propane pH 7.2, washed with 75 mL of the same buffer, and eluted stepwise with 0.1 M and 1 M NaCl.

Fraction	OD <sub>280</sub>	GVH Activity (uM F-/min/50 uL)	Specific Acivity (uM F-/min/50 uL/OD <sub>280</sub> )
10 mM BTP Wash	2.82	1.93	0.687
0.1 M NaCl Eluate	0.365	10.8	29.6
1 M NaCl Eluate	1.39	6.76	4.87

Table 3. Stepwise Elution of GVH A	ctivity
------------------------------------	---------

The 0.1 M NaCl eluate with the highest enzymatic activity was dialyzed overnight against 10 mM BTP pH 7.2 and again loaded on a second Q Sepharose column. After the nonbinding protein was removed by washing, elution was carried out with a linear gradient of 0 - 0.6 M NaCl in bis-tris propane buffer and analyzed for GVH activity. Figure 3 shows the GVH activity, DFP activity (OPAA) and OD<sub>280</sub> profile. The OPAA (DFP) activity peak is clearly distinct from the GVH activity peak, indicating these enzymes are not the same.





The fractions with GVH activity (fraction number 16-20) were pooled and loaded on a second smaller (5 mL) Q Sepharose column and eluted with a 0-0.15 M NaCI gradient (Figure 4).



Figure 4. Purification of GVH on Second Q Sepharose Column

The enzyme fraction with highest activity (fraction #18) from this second Q Sepharose chromatogram was analyzed by sodium dodecyl sulfate polyacrylamide gel electrophoresis (SDS-PAGE). Results (Figure 5) show a single band with a molecular weight of approximately 20 kilo Daltons (kDa).



Figure 5. SDS-PAGE of Fractions from GV Degrading Enzyme Purification Lanes 1 to 3 contain purified enzyme, 40-65% ammonium sulfate ppt, and crude extract, respectively. M; molecular weight standards.

### 3.4. Enzyme Characterization.

### 3.4.1. pH Profile.

The most active fraction (partially purified enzyme) from the first Q Sepharose column was used to determine the optimal assays conditions for the enzyme. The profile of the enzyme activity versus pH (Figure 6) was obtained using 3 mM substrate at 25 °C in 50 mM bis-tris-propane buffer.



Figure 6. pH Profile of GVH

## 3.4.2. <u>Temperature Profile</u>.

1.

Under the same condition used for the pH profile, a temperature profile was determined for the GVH enzyme at the optimal pH of 8.5 (Figure 7).



Figure 7. Temperature Profile of GVH

### 4. CONCLUSIONS

First report of enzymatic degradation of GV -demonstrated presence of GVH in several strains.

The A. undina GVH has been purified to apparent homogeneity (as determined by SDS-PAGE) following ammonium sulfate precipitation and two successive Q Sepharose chromatography steps.

The GVH enzyme has fairly narrow pH and temperature profiles with optima at approximately pH 8.5 and 35 °C, respectively.

Blank

÷.

Э

.

### LITERATURE CITED

1. Mulbry, W.W., Karns, J.S., Kearney, P.C., Nelson, J.O., McDaniel, C.S., and Wild, J.R., "Identification of a Plasmid-Borne Parathion Hydrolase Gene from *Flavobacterium* sp. by Southern Hybridization with *opd* from *Pseudomonas diminuta.,*" <u>Appl. Environ.</u> <u>Microbiol.</u> Vol. 51, pp 926-930 (1986).

2. McDaniel, C.S., Harper, L.L., and Wild, J.R., "Cloning and Sequencing of a Plasmid-Borne Gene (*opd*) Encoding a Phosphotriesterase," <u>J. Bacteriol.</u> Vol. 170, pp 2306-2311 (1988).

3. Dave, K.I., Miller, C.E., and Wild, J.R., "Characterization of Organophosphorus Hydrolases and the Genetic Manipulation of the Phosphotriesterase from *Pseudomonas diminuta.*," <u>Chem-Biol. Interactions</u> Vol. 87, pp 55-68 (1993).

4. Rowland, S.S., Speedie, M.K., and Pogell, B.M., "Purification and Characterization of a Secreted Phosphotriesterase (Parathion Hydrolase) from *Streptomyces lividans*," <u>Appl. Environ. Microbiol.</u> Vol. 57, pp 440-444 (1991).

5. Dumas, D.P., Caldwell, S., Wild, J., and Raushel, F., "Purification and Properties of the Phosphotriesterase from *Pseudomonas diminuta.*," <u>J. Biol. Chem.</u> Vol. 264, pp 19659-19665 (1989).

6. Dumas, D.P., Durst, H.D., Landis, W.G., Raushel, F.M., and Wild, J.R., "Inactivation of Organophosphorus Nerve Agents by the Phosphotriesterase from *Pseudomonas diminuta.*," <u>Arch. Biochem. Biophys.</u> Vol. 277, pp 155-159 (1990).

7. DeFrank, J.J., Beaudry, W.T., Cheng, T-c., Harvey, S.P., Stroup, A.N. and Szafraniec, L.L., "Screening of Halophilic Bacteria and *Alteromonas* Species for Organophosphorus Hydrolyzing Enzyme Activity," <u>Chem. Biol. Interactions</u> Vol. 87, pp 414-448 (1993).

8. DeFrank, J.J., Cheng, T-c., Kolakowski, J.E. and Harvey, S.P., "Advances in the Biodegradation of Chemical Warfare Agents and Related Materials," <u>Journal of Cellular</u> <u>Biochemistry Supplement</u> Vol 0 (21A):41 (1995).

9. Serdar, C.M., Murdock, D.C., and Rohde, M., "Parathion Hydrolase Gene from *Pseudomonas diminuta* MG: Subcloning, Complete Nucleotide Sequence, and Expresssion of the Mature Portion of the Enzyme in *Escherichia coli.,*" <u>Bio/Technology</u> Vol. 7, pp 1151-1155 (1989).

10. Serdar, C.M., and Gibson, D.T., "Enzymatic Hydrolysis of Organophosphates: Cloning and Expression of a Parathion Hydrolase Gene from *Pseudomonas diminuta.,*" <u>Bio/Technology</u> Vol. 3, pp 567-571 (1985). 11. Cheng, T-c., Harvey, S.P., and Chen, G.L., "Cloning and Expression of a Gene Encoding a Bacterial Enzyme for Decontamination of Organophosphorus Nerve Agents and Nucleotide Sequence of the Enzyme.," <u>Appl. Environ. Microbiol.</u> Vol. 62, pp 1636-1641 (1996).

12. Ashani, Y., Shapira, S, Levy, D., and Wolfe, A.D., "Butyrylcholinesterase and Acetylcholinesterase Prophylaxis against Soman Poisoning in Mice," <u>Biochem.</u> <u>Pharmacol.</u> Vol. 41, pp 37-41 (1991).

13. Kolakowski, J.E., DeFrank, J.J., Harvey, S.P., Szafraniec, L.L., Beaudry, W.T., Lai, K., and Wild, J.R., "Enzymatic Hydrolysis of the Chemical Warfare Agent VX and its Neurotoxic Analogues by Organophosphorus Hydrolase," <u>Biocatalysis and Biotransformation</u> Vol. 15, pp 297-312 (1997).

14. Lai, K., Dave, K.I., Wild, J.R., Szafraniec, L.L., Beaudry, W.T., and Harvey, S.P., "Enzymatic Decontamination of Organophosphorus Chemical Agents by Genetic and Biochemical Manipulation of Organophosphorus Hydrolase," in <u>Proceedings of the</u> <u>1994 ERDEC Scientific Conference on Chemical and Biological Defense Research</u>, 15-18 November 1994, ERDEC SP-036, U.S. Army Edgewood Research, Development and Engineering Center, Aberdeen Proving Ground, MD, May 1996, UNCLASSIFIED Report (AD A313 080).

15. ACTA MEDICA 39:27, 1996.

16. Petrikovics, I., Cheng, T-c, Papahadjopoulos, D., Hong, K., Yin, R., DeFrank, J. J., Jaing, J., Song, Z. H., McGuinn, W. D., Sylvester, D., Pei, L., Madec, J., Tamulinas, C., Jaszberenyi, J. C., Barcza, T., and Way, J. L., "Long Circulating Liposomes Encapsulating Organophosphorus Acid Anhydrolase in Diisopropylfluorophosphate," <u>Antagonism Toxicological Sciences</u> Vol. 57, pp 16-21 (2000).

17. Perrella, F.W., "EZ-FIT: A Practical Curve-Fitting Program for the Analysis of Enzyme Kinetic Data on IBM-PC Compatible Computers," <u>Analytical Biochemistry</u> Vol. 174, pp 437-447 (1988).

DEPARTMENT OF THE ARMY CDR USASBCCOM ATTN AMSSB SCI C 5183 BLACKHAWK ROAD APG MD 21010-5424

# **OFFICIAL BUSINESS**

# STANDARD B