

Structure and magnetic properties of Mn_2O_3



2009. 09. 30 Lab seminar
Kaist
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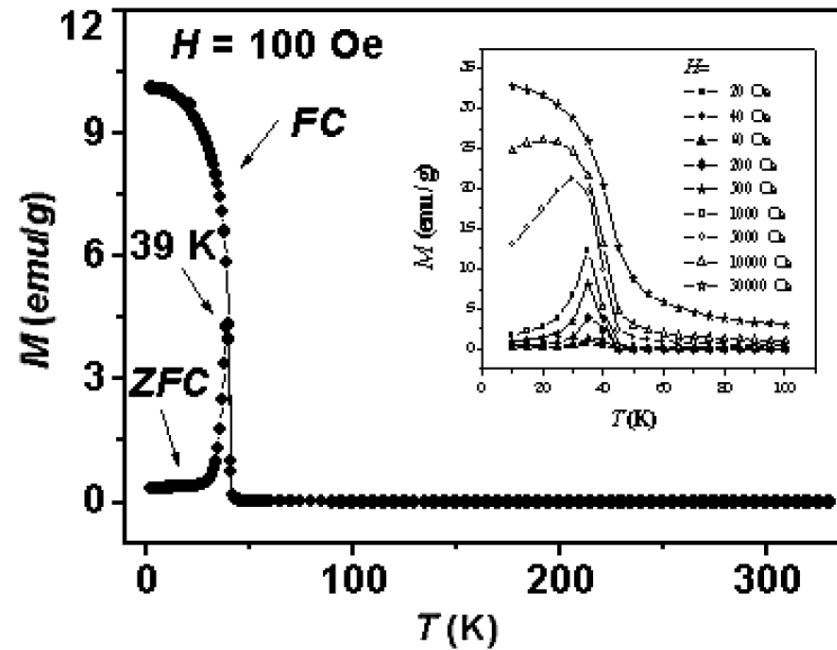
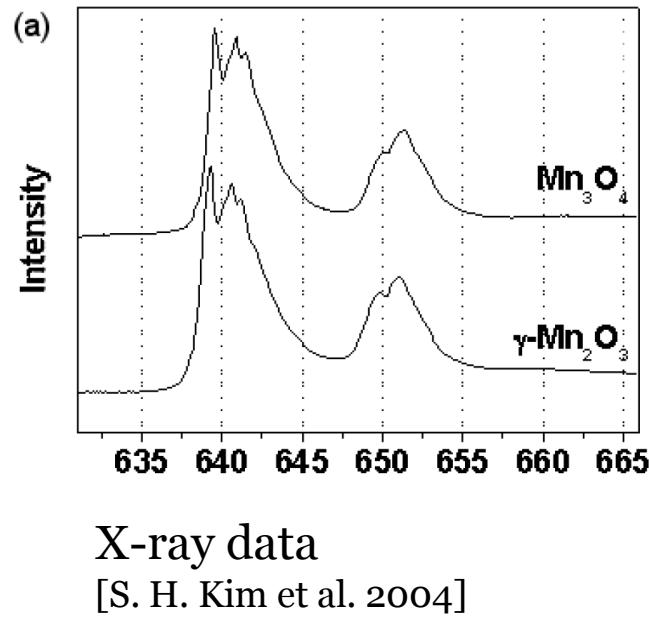
Manganese oxide

- Mn^{2+} : the most stable oxidation state ion
- Mn^{3+} : the second
 - at atmosphere for several days, some parts of Mn_2O_3 sample could be MnO
- Two forms of Mn_2O_3
 - α - Mn_2O_3 : by heating MnO_2 below 800°C
 - γ - Mn_2O_3 : by complex process
- All oxides and hydroxides of manganese form Mn_3O_4 if heated in air to about 1000°C

γ -Mn₂O₃

- No exist in natural – most of the Mn₂O₃ is the α phase
- If heated at 500 °C for 48 hours , or been standing at room temperature for 1 year , $\gamma \rightarrow \alpha$ phase
- Spinel structure
 - but fewer cation sites are occupied than normal spinel (Mn₃O₄)

γ -Mn₂O₃ nanoparticles



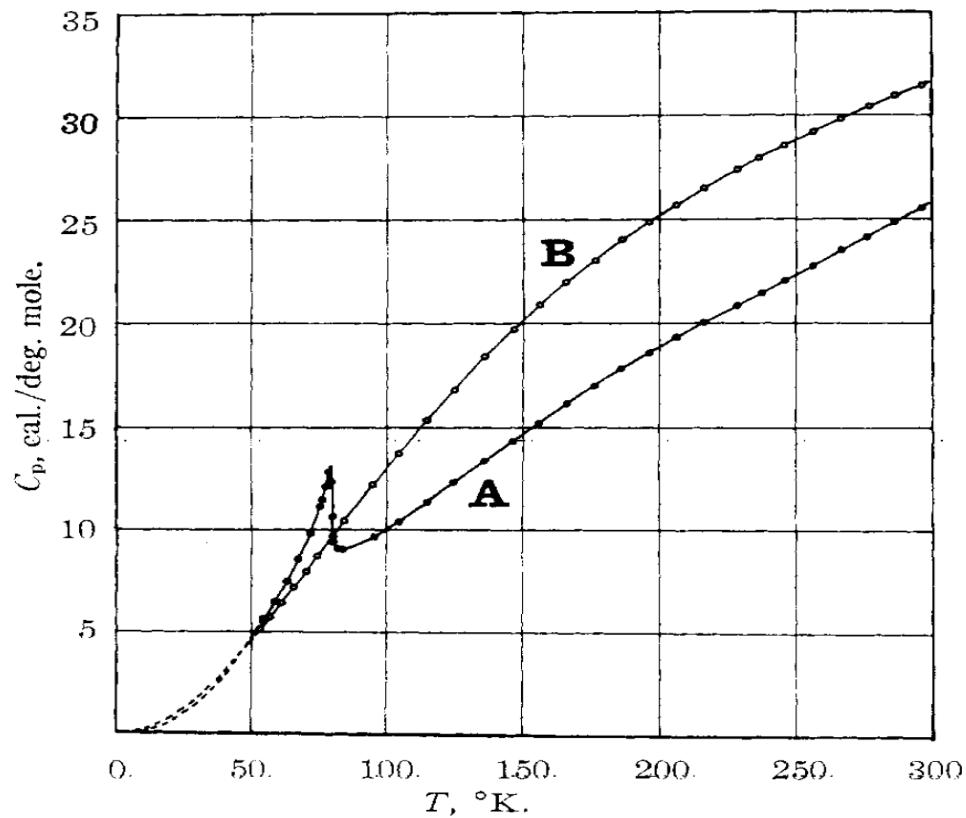
- zero field cooled(ZFC) and field cooled(FC) magnetization curves
 - Phase transition from ferrimagnetic to paramagnetic at 39K
- ZFC magnetization curves at other applied fields
 - magnetization increases with increasing H
 - transition temperature shift

$\alpha\text{-Mn}_2\text{O}_3$

- Crystal unit cell : 32 Mn³⁺ ions and 48 O²⁻ ions
- Structural phase transition
 - Above 302K : cubic bixbyite [S.Geller 1970]
 - Below 302K : orthorombic bixbyite [S.Geller 1970]
(308K [R.W. Grant, S. Geller 1968])

Bixbyite - manganese iron oxide mineral with formula (Mn, Fe)₂O₃

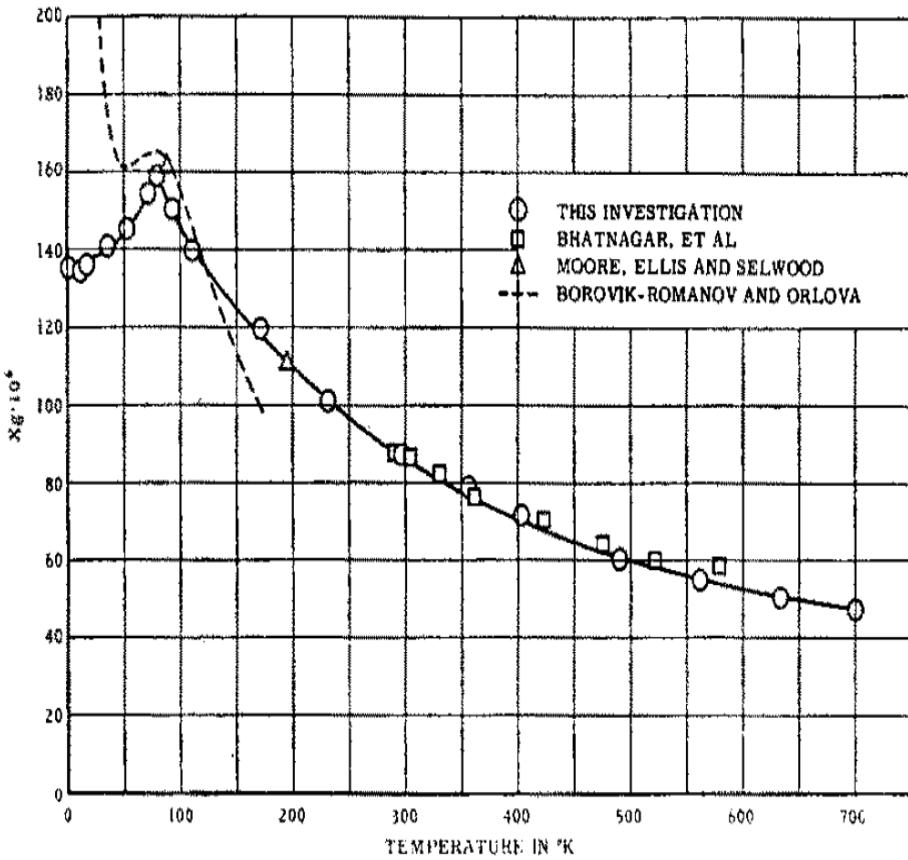
α -Mn₂O₃



Low temperature heat capacity
[E. G. King 1954]

- Peak at 79.4°K

$\alpha\text{-Mn}_2\text{O}_3$

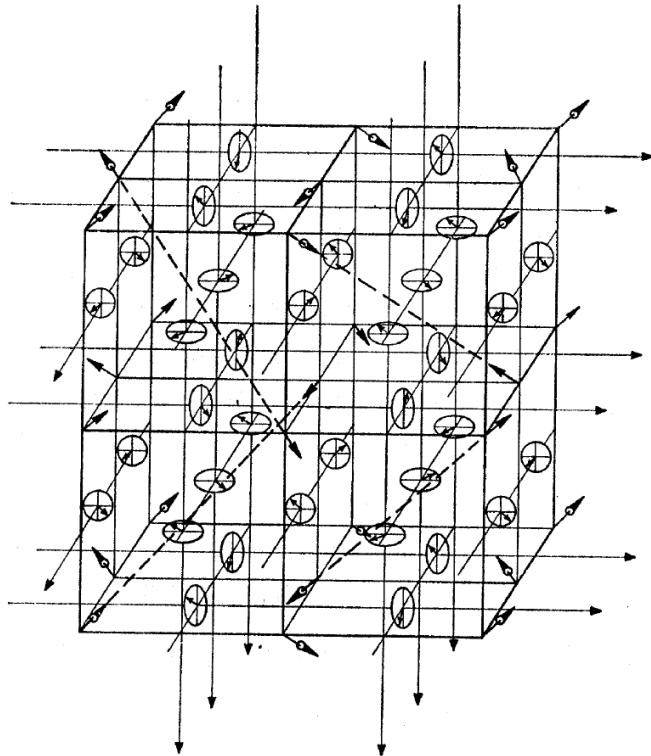


Magnetic susceptibility

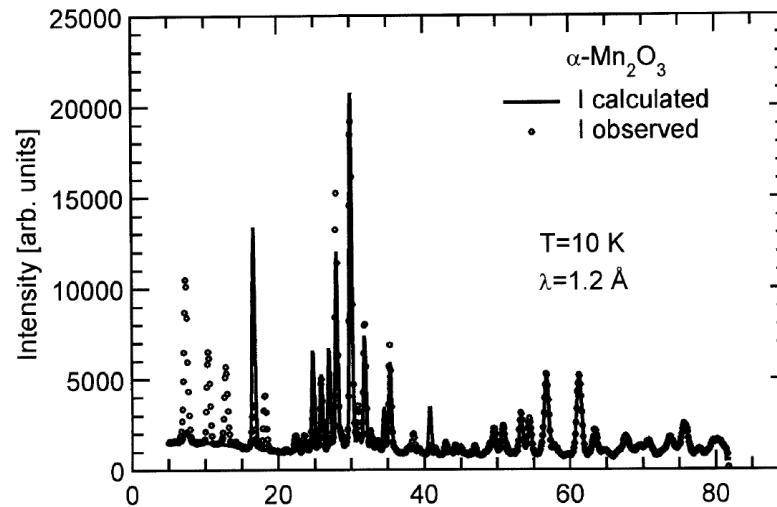
[R. G. Meisenheimer and D. L. Cook 1958]

- 4 sets of magnetic susceptibility data
- Antiferromagnetic ordering at about 80K

$\alpha\text{-Mn}_2\text{O}_3$

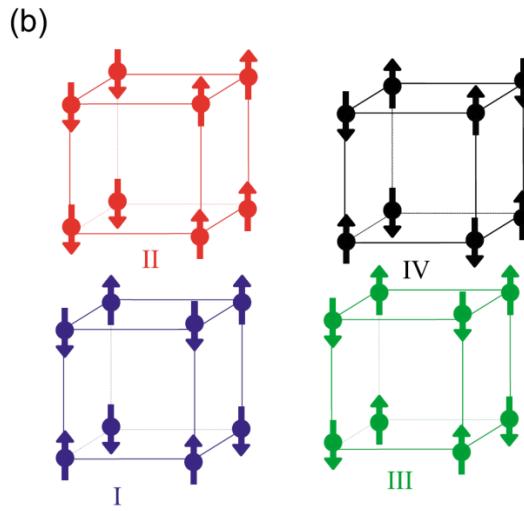
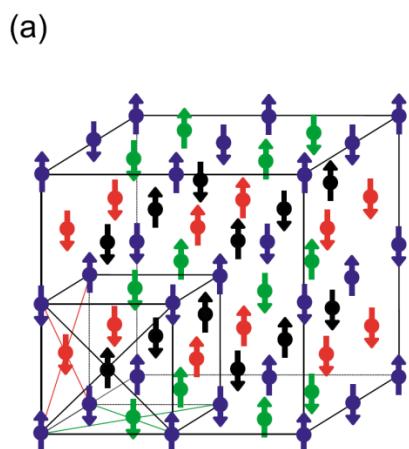


Noncollinear magnetic structure [Grant et al. 1968 and Geller et al. 1971]

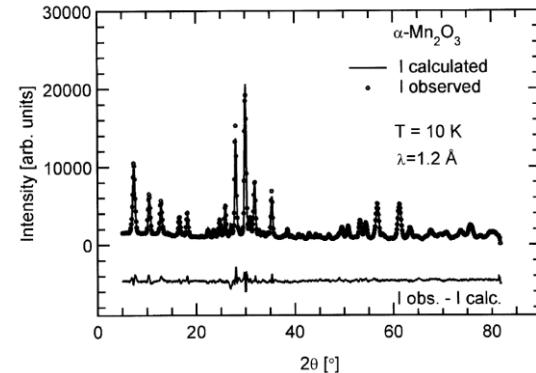


Neutron diffraction pattern at 10K
[M. Regulski and R. Schneider et al. 2002]

$\alpha\text{-Mn}_2\text{O}_3$



Magnetic structure
[Cable et al. 1957
And M. Regulski, R. Schneider et al
2002]



- Collinear model : all spins are parallel to the z-axis (z-axis is arbitrary)
- 4 sublattice : I (C type arrangement), II (A type), III (E type), IV (G type)
- Magnetic moments for each sublattice : I ($3.8\sim4.0\mu_B$), II($3.8\sim4.0\mu_B$), III($3.5\sim3.9\mu_B$), IV ($3.3\sim3.5\mu_B$)

References

- J. Cable, M. Wilkinson, E. Woolan, W. Koehler, ORNL-2302, Phys. Prog. Rep. (1957) 43
- E.G. King. *J. Am. Chem. Soc.* 76 (1954), p. 3289
- R.W. Grant, S. Geller, J.A. Cape and G.P. Espinosa. *Phys. Rev.* 175 (1968), p. 686
- S. Geller and G.P. Espinosa. *Phys. Rev. B* 1 (1970), p. 3763
- S. Geller. *Acta Crystallogr.* 27 (1971), p. 821
- M. Regulski, R. Przenios o, I. Sosnowska and D. Hohlwein, BENSC Experimental Reports 1988. *HMI-B* 559 (1999), p. 41
- Greenwood, Norman N. ; Earnshaw, A. (1997), chemistry of the elements(2nd ed.), Oxford: Butterworth-Heinemann, ISBN 0-75-6-3365-4
- Ferrimagnetism in γ -Manganese Sesquioxide ($\gamma\text{-Mn}_2\text{O}_3$) Nanoparticles, Kim S. H, Choi B. J, Lee G. H., Oh S. J., Kim B., Choi H. C., Park J, Chang Y., Journal of the Korean Physical Society, 46, 4, (2005), 941~944