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1 **Tracking the hydro-climatic signal from lake to sediment: a field study from central**  
2 **Turkey**

3

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**Abstract**

24

25 Palaeo-hydrological interpretations of lake sediment proxies can benefit from a robust  
26 understanding of the modern lake environment. In this study, we use Nar Gölü, a non-outlet,  
27 monomictic maar lake in central Turkey, as a field site for a natural experiment using  
28 observations and measurements over a 17-year monitoring period (1997-2014). We compare  
29 lake water and sediment trap data to isotopic, chemical and biotic proxies preserved in its  
30 varved sediments. Nar Gölü underwent a 3 m lake-level fall between 2000 and 2010.

31  $\delta^{18}\text{O}_{\text{lakewater}}$  is correlated with this lake-level fall, responding to the change in water balance.

32 Endogenic carbonate is shown to precipitate in isotopic equilibrium with lake water and there  
33 is a strong relationship between  $\delta^{18}\text{O}_{\text{lakewater}}$  and  $\delta^{18}\text{O}_{\text{carbonate}}$ , which suggests the water balance  
34 signal is accurately recorded in the sediment isotope record. Over the same period,

35 sedimentary diatom assemblages also responded, and conductivity inferred from diatoms  
36 showed a rise. Shifts in carbonate mineralogy and elemental chemistry in the sediment record  
37 through this decade were also recorded. Intra-annual changes in  $\delta^{18}\text{O}_{\text{lakewater}}$  and lake water  
38 chemistry are used to demonstrate the seasonal variability of the system and the influence this  
39 may have on the interpretation of  $\delta^{18}\text{O}_{\text{carbonate}}$ . We use these relationships to help interpret the

40 sedimentary record of changing lake hydrology over the last 1,725 years. Nar Gölü has  
41 provided an opportunity to test critically the chain of connection from present to past, and its  
42 sedimentary record offers an archive of decadal- to centennial-scale hydro-climatic change.

43

44 **Keywords:** Oxygen isotopes, Diatom analysis, Lake sediments, Monitoring, Seasonality,  
45 Carbonates

46

47

48 **Highlights**

49

- 50 • Study of non-outlet, oligosaline, varve-forming lake in a semi-arid region
- 51 • Water balance signal in oxygen isotopes tracked from lake waters to sediments
- 52 • Strong intra- and inter-annual relationships between isotopes and water balance
- 53 • Diatom-inferred conductivity shows a complex response to change in water balance
- 54 • Implications of monitoring data for interpretation of palaeo-records

## 55 1. Introduction

56

57 In order to use lake sediments to reconstruct past climate change reliably, it is vital to  
58 understand the modern hydrology of the study site (e.g. Hollander and McKenzie, 1991; Leng  
59 et al., 1999; Saros, 2009) and to be able to track this signal to the sediments. Lake systems  
60 respond to hydro-climatic variations via a number of linked parameters, including lake  
61 volume, salinity concentrations and the oxygen isotope ( $\delta^{18}\text{O}$ ) composition of waters. Non-  
62 outlet lakes respond particularly dynamically to changes in water balance (Leng and  
63 Marshall, 2004 and references therein); with increased evaporation, water volume decreases,  
64 salts become concentrated and  $\delta^{18}\text{O}_{\text{lakewater}}$  becomes more positive, and vice-versa, although  
65 parameters may be subject to hysteretic effects (Langbein, 1961) as well as other factors such  
66 as saline groundwater inflows.

67 Limnological parameters such as water balance are registered by proxies preserved in  
68 lake sediments, which in turn permit the reconstruction of lake hydrology for pre-  
69 instrumental time periods (Fritz, 2008 and references therein). Past water level fluctuations  
70 can be reconstructed via dated lake marginal depositional facies, such as shoreline terraces  
71 and carbonate platforms (Magny, 2006), and by changes in the species assemblages and life  
72 forms of diatoms and other biological indicators (e.g. Barker et al., 1994). Salinity inferred  
73 from biological indicators, such as diatom and ostracod assemblages, is sometimes quantified  
74 as variability in electrical conductivity (EC) based on transfer function techniques using a  
75 modern training set (e.g. Fritz et al., 2010; Reed et al., 2012). Past salinity levels can also be  
76 reconstructed semi-quantitatively from elemental chemistry ratios such as Ca/Sr and Mg/Ca  
77 (Ito, 2001). In many lakes, the form of carbonate precipitated from lake waters shifts from  
78 low-Mg calcite in dilute lake waters to high-Mg calcite or aragonite in more saline lake  
79 waters (Kelts and Hsü, 1978) and the Ca/Sr ratio can decrease if there is a shift from calcite

80 to aragonite precipitation (Tesoriero and Pankow, 1996). Stable isotopes can also be used as a  
81 palaeo-hydrological proxy: lake water  $\delta^{18}\text{O}$  is recorded in carbonates that precipitate in lake  
82 water;  $\delta^{18}\text{O}_{\text{carbonate}}$  is also modified by temperature and potentially by disequilibrium or  
83 diagenetic effects (Leng and Marshall, 2004 and references therein).

84         Limnological sampling, monitoring and observation can provide fundamental insights  
85 into all of the processes described above, and therefore strengthen the interpretation of lake  
86 sediment records. Monitoring of lake levels leads to an understanding of the sensitivity of a  
87 given lake to hydrological and/or climatic change. Recording biological response to measured  
88 climate or hydrological change improves the interpretation of downcore species changes.  
89 Monitoring data may be especially important when using stable isotopes as a hydro-climatic  
90 proxy because it is not possible to apply modern analogue or transfer function techniques,  
91 substituting time with space, to these records due to their dependence on multiple climatic  
92 and site-specific non-climatic variables (Tian et al., 2011). Monitoring allows the  
93 establishment of the key drivers of  $\delta^{18}\text{O}_{\text{lakewater}}$  in the lake being studied and a better  
94 understanding of how the signal is transferred to carbonates in the sediment record. Such a  
95 monitoring approach can provide a basis for judging which proxies provide the most reliable  
96 register of environmental changes (such as hydro-climate) and why different proxies can  
97 show different trends in the palaeo-limnological record, although the possibility that present  
98 lake states are not good analogues for the past should also be considered.

99         There are logistical and financial barriers to collecting modern data and samples over  
100 multiple years and different seasons for a length of time suitable to ensure robust proxy  
101 interpretation, especially in remote regions. However, in this study, we have been able to  
102 collect a substantial number of samples from Nar Gölü (göl = lake in Turkish), a small,  
103 hydrologically sensitive maar lake in central Turkey, over a period of 17 years (1997-2014).  
104 Although our monitoring and observational data are far from complete, they do allow an

105 assessment to be made of both seasonal variations and multi-year trends. If lake sediments  
106 are sufficiently well resolved in time, it is possible to trace changes measured from lake  
107 waters collected from certain years to the sediments that correspond to that year. Nar Gölü is  
108 particularly useful for such an exercise because the sediment record is annually laminated  
109 (varved). We have therefore been able to correlate, with high precision, monitoring and  
110 instrumental climate data to palaeo-limnological information from the sediment cores over  
111 the same period.

112         The study lake was subject to a progressive water level decrease between 2000 and  
113 2010. We examine how this change in lake water balance was registered by different hydro-  
114 chemical and biological parameters over time, and how they were subsequently incorporated  
115 in the contemporaneous lake sediment record. Some neo-limnological data from Nar Gölü  
116 have been previously published: Jones et al. (2005) compared modelled and measured  $\delta^{18}\text{O}$   
117 results (using water isotope data from 1999-2002) and Woodbridge and Roberts (2010)  
118 examined diatom assemblage data (with contemporary samples taken 2002-2007). Here we  
119 present new water isotope and chemistry data to extend the record up to 2014 and new  
120 sediment isotope and diatom assemblage data to bring the record up to 2010. With this longer  
121 time series of monitoring data, we build on these previous studies and aim to: (1) establish  
122 the general physical, isotopic and geochemical characteristics of the lake, (2) scrutinise intra-  
123 annual trends in lake water chemistry and  $\delta^{18}\text{O}_{\text{lakewater}}$  to understand the seasonal variability of  
124 the system, (3) compare inter-annual variability in lake water chemistry and  $\delta^{18}\text{O}_{\text{lakewater}}$  to  
125 physical and climate variables in order to test the drivers of the record, and (4) compare these  
126 data to isotopic, biological and geochemical proxies from the sediment record. The analysis  
127 of modern limnology and the tracking of signals from the lake water to sediments from the  
128 last decade allow us to assess critically individual palaeo-limnological proxies at Nar Gölü,



129 ultimately to better interpret the long-term sediment record of Holocene hydro-climatic  
130 change (e.g. Jones et al., 2006; Woodbridge and Roberts, 2011; Yiğitbaşıoğlu et al., in press).

131

## 132 **2. Site description**

133

134 Nar Gölü (38°20'24"N, 34°27'23"E; 1,363 m.a.s.l.) is a small (~0.7 km<sup>2</sup>) but relatively deep  
135 (>20 m) maar lake in Cappadocia, central Turkey (Figure 1). It is oligosaline, alkaline and  
136 predominately groundwater-fed, with a residence time of 8-11 years (Jones et al. 2005;  
137 Woodbridge and Roberts, 2010). The crater geology is predominately basalt and ignimbrite  
138 (Gevrek and Kazancı, 2000). Nar Gölü lacks any surface outflow. At its southern edge there  
139 are a series of small inflowing ephemeral stream channels forming an alluvial fan, and the  
140 bathymetric map (Figure 1) shows that this extends into the lake as a fan-delta.

141 The climate of the region is continental Mediterranean (Kutiel and Türkeş, 2005) with  
142 annual precipitation at Niğde, 45 km from Nar Gölü and 1,208 m.a.s.l., averaging 339 mm  
143 from 1935 to 2010. Mean monthly temperatures 1935-2010 varied from an average of +23°C  
144 in July and August to +0.7°C from December to February (see Dean et al., 2013 for more  
145 detailed regional climate data).

146 Although the lake watershed contains no permanent dwellings and only a few  
147 agricultural fields, Nar Gölü has not entirely escaped human impact. Firstly, groundwater  
148 pumping for irrigation in the valley below the lake is likely to have steepened the hydraulic  
149 gradient in recent decades, possibly increasing groundwater outflows from the lake.  
150 Secondly, in 1990 the Turkish Geological Survey (MTA) drilled boreholes near to the lake to  
151 reach artesian geothermal groundwaters (Akbaşlı, 1992). Oral testimony indicates that one of  
152 these drill holes significantly disturbed lake hydrology and ecology (potentially including a  
153 breakdown in lake stratification and a decrease in the population of aquatic macrophytes),

154 probably for several years, for which there is some evidence in lake sediment cores.  
155 Consequently, and given the lake residence time, we restrict our analysis of changing lake  
156 conditions to the period since 1997.

157

158 Figure 1

159

### 160 **3. Materials and methods**

161

#### 162 *3.1 Fieldwork*

163

164 Water samples were collected from the lake during 22 field visits between 1997 and 2014.

165 When conditions permitted, depth profiles were taken from the deepest part of the lake  
166 through the water column using a Van Dorn bottle (Van Dorn, 1956) or a Glew corer (Glew  
167 et al., 2001) with temperature, pH and EC measured at the time on a Myron ® meter.

168 Maximum lake depths were estimated using a Garmin Fish Finder ® and a weighted tape and  
169 checked against water level stage readings at the lake edge when possible. Bathymetry was  
170 measured using a Boomer system coupled with a high precision GPS, based on 53 transect  
171 lines north-south and east-west (Smith, 2010), in order to identify a suitable coring site.

172 Samples were taken for isotope and major ion analysis in the UK. Surface water samples  
173 were taken in bottles initially washed three times in the sample, at 0.5 m depth to remove any  
174 direct effects of exchange with the atmosphere. Where it was not possible to go out on the  
175 lake, surface samples were taken from the same spot on the edge of the lake. Edge samples  
176 were also taken by members of the local community between February and June 2012, as  
177 well as a photo diary that allowed us to establish when snowmelt occurred that year (SI  
178 Figure 1). Spring waters from the catchment (Figure 1) were also regularly sampled.

179 Simple sediment traps, consisting of cylindrical plastic tubes under funnels, were  
180 attached at a variety of depths onto ropes that were secured with an anchor on the lake bed  
181 and a float on the surface and replaced every year. Since 2010, Tinytag ® temperature  
182 loggers have been attached to the sediment trap lines at a number of depths through the water  
183 column. These provide temperature measurements at 20-minute intervals throughout the year.

184 A 44 cm long sediment core, which covers all but the last few years of the period of  
185 lake water monitoring, was taken in 2010 (NAR10) using a weighted stationary piston corer,  
186 another having been taken with a Glew corer (36 cm) in 2006 (NAR06). Longer cores  
187 spanning 1,720 years were taken in 2001/2 (NAR01/02).

188

### 189 3.2 Laboratory analyses

190

191 Water samples were analysed for  $\delta^{18}\text{O}$  and  $\delta\text{D}$  on a VG Isoprime mass spectrometer and a  
192 EuroPyrOH analyser. Isotopic ratios are given as ‰ deviations from VSMOW, and analytical  
193 reproducibility was 0.05‰ for  $\delta^{18}\text{O}$  and 2‰ for  $\delta\text{D}$ . Major ion concentrations were measured  
194 on water samples as soon as possible after returning from the field on a Metrohm ion  
195 chromatography system. Data were converted from milligrams/litre to milliequivalents/litre  
196 ( $\text{meqL}^{-1}$ ) (Hem, 1970).

197 Carbonates from sediment traps and core sediments were analysed for  $\delta^{18}\text{O}$  using an  
198 offline extraction technique and a VG Optima mass spectrometer and data are given as ‰  
199 deviations from VPDB, with an analytical reproducibility of 0.1‰. Carbonate mineralogy  
200 was investigated by X-ray diffraction. The scanning range used was  $5\text{-}65^\circ 2\theta$  and the scan  
201 rate was  $2^\circ 2\theta$  per minute with a step size of 0.05. The TRACES program by Diffraction  
202 Technology was used to identify which minerals were present. Where two or more minerals  
203 were present, the proportions of each were determined by calculating the area under the peaks

204 and the percentage of aragonite compared to calcite was estimated from experimentally  
 205 calibrated conversion curves (Hardy and Tucker, 1988).

206 X-ray fluorescence (XRF) analysis of elemental sediment chemistry was carried out  
 207 on split half cores by a field portable XRF spectrometer, which produces one single  
 208 dispersive energy spectra for each 3 mm sampling point on the core surface, with data in  
 209 parts per million.

210 Diatom samples were prepared using standard methods adapted from Battarbee et al.  
 211 (2001), described in detail in Woodbridge and Roberts (2010).

212

### 213 3.3 Numerical analyses

214

215 To model aragonite precipitation dynamics in Nar Gölü, the palaeo-temperature equation of  
 216 Kim et al. (2007) is used:

217

$$218 T = (17.88 * 1000) / (1000 * \ln((1000 + \delta^{18}\text{O}_{\text{aragonite}}) / (1000 + \delta^{18}\text{O}_{\text{lakewater}})) + 30.77) - 273.15 \quad (1)$$

219

220 where  $\delta^{18}\text{O}_{\text{aragonite}}$  and  $\delta^{18}\text{O}_{\text{lakewater}}$  are expressed against VSMOW and T in °C.

221

222 To model calcite precipitation dynamics, the palaeo-temperature equation of Hays and  
 223 Grossman (1991) is used:

224

$$225 T = 15.7 - 4.36 * (\delta^{18}\text{O}_{\text{calcite}} - \delta^{18}\text{O}_{\text{lakewater}}) + 0.12 * (\delta^{18}\text{O}_{\text{calcite}} - \delta^{18}\text{O}_{\text{lakewater}})^2 \quad (2)$$

226

227 where  $\delta^{18}\text{O}_{\text{calcite}}$  is expressed against VPDB,  $\delta^{18}\text{O}_{\text{lakewater}}$  against VSMOW and T in °C.

228

229 Diatom data have been used to infer EC using a combined salinity training set  
230 (comprising data from East Africa, North Africa and Spain) provided by the European  
231 Diatom Database (EDDI) (Juggins, 2014). Training sets and models were selected based on  
232 the percentage of fossil sample species represented in the modern data set, the number of sites  
233 in which these species are present and the model performance ( $r$  and RMSEP), and the  
234 models were run using C2 software (Juggins, 2003). The combined salinity EDDI modern  
235 training set was identified as possessing the highest number of matching analogue diatom  
236 species in the Nar Gölü fossil assemblage (74.4%; species not in the training set include  
237 *Clupeoparvus anatolicus*, a species endemic to Nar Gölü; Woodbridge et al., 2010). Weighted  
238 averaging with inverse deshrinking was identified as the model with highest predictive ability  
239 ( $r=0.85$ ) and lowest prediction errors (RMSEP = 0.47). Detrended Correspondence Analysis  
240 (DCA) was also applied to the diatom percentage data because the length of the axis was >2  
241 units (Lepš and Šmilauer, 2003).

242 Monthly instrumental meteorological data from a nearby station at Niğde (155 m  
243 altitudinal difference, 45 km from Nar Gölü) have been used to create a hydro-climatic index  
244 of moisture availability (precipitation/evaporation; P/E). Because of the 8-10 year residence  
245 time of the lake water (Jones et al., 2005), we calculated a cumulative weighted 8-year P/E  
246 index.

247

## 248 **4. Results**

249

### 250 *4.1 Basic limnological and sedimentological information*

251

252 EC and major ion data show that the lake is oligosaline, with a mean conductivity value over  
253 the past 15 years of  $3,270 \mu\text{Scm}^{-1}$  (Tables 1 and 2) and  $\delta^{18}\text{O}_{\text{lakewater}}$  values that are higher than

254 freshwater  $\delta^{18}\text{O}_{\text{spring}}$  values (Figure 2), indicating that the lake waters are evaporated relative  
255 to spring waters. A former lake high-stand is evident from carbonate-encrusted rocks and  
256 strandline deposits, surveyed at 5 m above the 2010 water level, or 2 m above the lake  
257 elevation in 2000, and provides physical evidence of the tendency of the lake level to  
258 fluctuate. The sediments of Nar Gölü comprise alternating organic and carbonate layers  
259 (varves; Ojala et al., 2012), with an organic and carbonate couplet shown to represent one  
260 year of sedimentation based on analysis of sediment traps, thin sections and independent  
261 dating of the sediment cores by  $^{210}\text{Pb}$  and  $^{137}\text{Cs}$  (Jones et al., 2005; Woodbridge and Roberts,  
262 2010).

263

264 Figure 2

265

266 Tables 1 and 2

267

#### 268 4.2 *Intra-annual variability*

269

270 Figure 3 shows the intra-annual variability in water chemistry from samples taken  
271 between June 2011 and July 2012. Within the data available,  $\delta^{18}\text{O}_{\text{lakewater}}$  values peak at –  
272 0.13‰ in mid-September 2011 before falling to –1.76‰ in mid-March 2012 and then  
273 increasing to –0.39‰ in mid-July 2012. EC values also peak in mid-September 2011 at 3,540  
274  $\mu\text{Scm}^{-1}$ , before decreasing to 2,190  $\mu\text{Scm}^{-1}$  in late February 2012 (when there was heavy  
275 snowfall and partial lake icing) and increasing again to 3,500  $\mu\text{Scm}^{-1}$  by July 2012. pH values  
276 decreased from 8.1 in June 2011 to 7.3 in February 2012 before increasing to 8.0 by June  
277 2012. Magnesium concentrations decreased from 9.3  $\text{meqL}^{-1}$  in September 2011 to a  
278 minimum of 3.2  $\text{meqL}^{-1}$  in late February 2012 and then increased to 16.5  $\text{meqL}^{-1}$  by July

279 2012, whereas calcium concentrations showed the opposite trend, shifting from 2.2 meqL<sup>-1</sup> in  
280 June 2011 to 4.0 meqL<sup>-1</sup> in late February 2012 to 1.2 meqL<sup>-1</sup> in July 2012.

281 Because the lake is monomictic, depth profiles, as well as surface samples, were  
282 taken. In the summer, the waters of Nar Gölü are thermally and isotopically stratified, with  
283 warmer and isotopically more positive waters in the epilimnion, followed by a shift at ~7 m  
284 to colder and isotopically more negative values in the hypolimnion (Figure 4). The degree of  
285 stratification becomes more pronounced from the spring to summer. While no depth profiles  
286 were taken during the autumn or winter at Nar Gölü, temperature loggers show that the lake  
287 is thermally mixed between November and March, with the same temperatures at 5 m and 21  
288 m during the winter and then diverging in the spring (Figure 3 for 2011-12, but also observed  
289 for other years; Eastwood et al., unpublished data).

290

291 Figure 3

292

293 Figure 4

294

#### 295 4.3 *Inter-annual trends*

296

297 When considering inter-annual trends, samples collected from the same time of year  
298 over multiple years are used to remove possible issues caused by the significant intra-annual  
299 variability in the system presented in section 4.2. July is the month for which most data are  
300 available. Samples from the lake centre are considered most representative of overall lake  
301 conditions, because shallow water edge samples may be more affected by evaporation,  
302 particularly in summer months. Nonetheless, the difference between centre and edge  
303  $\delta^{18}\text{O}_{\text{lakewater}}$  samples is only  $\pm 0.3\text{‰}$  ( $1\sigma$ ,  $n=4$ ) in years where both were taken, which is small

304 considering the size of the inter-annual isotopic shifts seen in the record. Therefore, edge  
305 samples from 2000 and 2005 have been combined with centre samples from other years to  
306 provide a more complete record. As Figure 5 shows,  $\delta^{18}\text{O}_{\text{lakewater}}$  values increased from –  
307 3.20‰ in July 2000 to –0.24‰ in July 2010. Over this period, the lake level fell by  
308 approximately 3 m and lake water volume shrank by ~20%. Measured July surface EC values  
309 increased from 3,300  $\mu\text{Scm}^{-1}$  in 2001 to 3,500  $\mu\text{Scm}^{-1}$  in 2012 (Figure 5), while lake surface  
310 pH from the same month rose from ~7.5 in 2001 to >8 in 2008, before declining to 7.8 by  
311 2012 (Table 1).

312 The  $\delta^{18}\text{O}_{\text{lakewater}}$  increase for the period 2000-2010 was matched by an increase in  
313 sediment core  $\delta^{18}\text{O}_{\text{carbonate}}$  values from –3.7‰ to –0.5‰. There is a close relationship between  
314 sediment trap and core  $\delta^{18}\text{O}_{\text{carbonate}}$  values from the same years, with both showing an increase  
315 over the period of study (Figure 5). Sediment trap samples collected from different depths in  
316 the same years (2002 and 2004) have  $\delta^{18}\text{O}_{\text{carbonate}}$  values that are the same within analytical  
317 reproducibility. There are small differences between sediment trap and core  $\delta^{18}\text{O}_{\text{carbonate}}$   
318 values from 2003, 2004 and 2005 but the trends are the same in both data sets. Between 2006  
319 and 2007 there was a start of a trend towards a reduction in the Ca/Sr ratio and a shift from  
320 calcite to aragonite in lake sediment carbonates (Figure 5).

321

322 Figure 5

323

324 EC inferred from sedimentary diatom assemblages (diatom-inferred electrical  
325 conductivity; DI-EC) underestimates modern measured lake EC (Figure 6) (reasons for this  
326 will be proposed in sections 5.2 and 5.3, partly related to the fact *C. anatolicus* is not  
327 included in the modern training set), but DI-EC trends do broadly match those from  
328 measurements taken during the monitoring period. A change in sedimentary diatom



329 assemblages began earlier than the DI-EC increase, with *C. anaticus* and *Synedra acus*  
330 replacing *Nitzschia paleacea* as the dominant taxa after 2001 (Figure 6).

331

332 Figure 6

333

334 The closest meteorological station to Nar Gölü with a long-term data set is at Niğde.  
335 Annual precipitation in this area was at or below the long-term average (339 mm) from 1997  
336 to 2008, with the exception of 2001. In addition, the 1990s saw a significant ( $>3^{\circ}\text{C}$ ) rise in  
337 average summer temperatures (Turkish State Meteorological Service, pers. comm).

338

## 339 5. Discussion

340

### 341 5.1 Intra-annual variability at Nar Gölü

342

343 The seasonal variability in surface water  $\delta^{18}\text{O}$  and conductivity shown in Figure 3 can be  
344 explained by two main factors. Firstly, the water in the lake as a whole has lower  $\delta^{18}\text{O}$  in the  
345 autumn, winter and spring, as these are the main seasons for rainfall and snowfall, input of  
346 which will lower  $\delta^{18}\text{O}_{\text{lakewater}}$  (Dean et al., 2013). Although not quantified, observational data  
347 show that lake levels were visibly higher in the spring than during the following summers.  
348  $\delta^{18}\text{O}_{\text{lakewater}}$  in 2012 was lowest in mid-March and the photo diary (SI Figure 1) shows this  
349 was the time in that year of snowmelt from the catchment. Rainfall is also greatest in the  
350 spring (Kutiel and Türkeş, 2005). Secondly, stratification of lake waters in the summer leads  
351 to more positive  $\delta^{18}\text{O}$  values in surface waters than at depth because the hypolimnion is  
352 unaffected by evaporative processes. Comparison of the depth profiles from April, June, July  
353 and September (Figure 4) show that the isocline becomes more enhanced as the year

354 progresses, with a 1.00‰ difference between surface and bottom water  $\delta^{18}\text{O}$  values in  
355 September 2011 compared to a 0.75‰ difference in July 2010, 0.24‰ in June 2011 and  
356 0.23‰ in April 2013.

357         Given the seasonal variability in  $\delta^{18}\text{O}_{\text{lakewater}}$ , we need to establish the timing of  
358 carbonate precipitation to allow for proper interpretation of the palaeo-record. Carbonate  
359 precipitation in surface waters is demonstrated by the observation that sediment traps at 5 m  
360 depth are encrusted in carbonate when changed each year, whereas deeper ones are not.  
361 Variability in  $\delta^{18}\text{O}_{\text{carbonate}}$  with depth in one of the sediment traps suggests carbonate  
362 precipitation under changing temperatures and/or  $\delta^{18}\text{O}_{\text{lakewater}}$  (Figure 7), i.e. that carbonate  
363 precipitation occurs at different times of the year. However,  $\delta^{18}\text{O}_{\text{carbonate}}$  measured in the  
364 sediment record from a whole-year varve will be weighted towards the time of maximum  
365 precipitation. Observations suggest this occurs between May and early July. Firstly, in July,  
366 calcium values at the surface are lower than at depth, suggesting draw-down of calcium  
367 carbonate from the surface waters (Reimer et al., 2009), whereas in April 2013 calcium  
368 concentration was still higher in surface waters than at depth, suggesting this draw-down had  
369 yet to occur (Table 2). Secondly, analysis of the stratigraphy of Nar Gölü sediment traps  
370 collected in July shows carbonate deposited on top of organic matter, while sediment traps  
371 collected in April show organic matter on top of carbonate (Figure 8), suggesting that the  
372 carbonate for that year had yet to precipitate.

373

374 Figure 7

375

376 Figure 8

377

378           Additionally, it is possible to run Eqs. 1 and 2 using various  $\delta^{18}\text{O}_{\text{lakewater}}$  and  
379 temperature scenarios, and then to compare the calculated equilibrium  $\delta^{18}\text{O}_{\text{carbonate}}$  values  
380 from these equations to measured  $\delta^{18}\text{O}_{\text{carbonate}}$  from the sediment core. By seeing where the  
381 calculated values match the measured values, it is possible to investigate better the timing of  
382 carbonate precipitation. Before doing this, equilibrium precipitation and a lack of diagenetic  
383 effects need to be demonstrated. It is not unknown for carbonate to precipitate out of  
384 equilibrium with lake waters (Fronval et al., 1995; Teranes et al., 1999). During the July 2012  
385 field season, carbonate in the form of aragonite was seen precipitating from the waters in a  
386 ‘white-out’ event (as seen in other lakes; Romero-Viana et al., 2008; Sondi and Juracic, 2010;  
387 Viehberg et al., 2012) around the edges of the lake (SI Figure 2). Comparison of the  
388  $\delta^{18}\text{O}_{\text{carbonate}}$  value from this aragonite precipitate ( $-1.3\text{‰}$ ) to the  $\delta^{18}\text{O}_{\text{carbonate}}$  value predicted  
389 using Eq. 1 ( $-1.8\text{‰}$ , using the  $\delta^{18}\text{O}_{\text{lakewater}}$  ( $-0.39\text{‰}$ ) and temperature ( $+25.6^\circ\text{C}$ ) values  
390 measured from a water sample taken at the same time), show that it formed in equilibrium  
391 within analytical and equation error. Diagenesis may alter the carbonate isotope signal  
392 between precipitation and deposition (Teranes and Bernasconi, 2000). However, at Nar Gölü,  
393 there are only small differences between the  $\delta^{18}\text{O}_{\text{carbonate}}$  values of trap and core sediments  
394 from the same year (Figure 5) and the inter-annual trends are very similar, which suggests  
395 minimal alteration of the  $\delta^{18}\text{O}_{\text{carbonate}}$  signal during and after deposition.

396           Based on the observations already outlined, we assume that most carbonate is  
397 precipitated sometime after April but before the end of July and in surface waters. Therefore,  
398 we use likely surface water temperature and  $\delta^{18}\text{O}_{\text{lakewater}}$  values from May to July to calculate  
399 potential  $\delta^{18}\text{O}_{\text{carbonate}}$  values. Temperatures vary from year to year, but temperature loggers  
400 suggest temperatures change from  $\sim+12.5^\circ\text{C}$  in the beginning of May to  $\sim+17.5^\circ\text{C}$  in mid-  
401 June to  $\sim+20.0^\circ\text{C}$  in the beginning of July to  $\sim+22.5^\circ\text{C}$  in mid-July (Figure 3). Consequently,  
402 temperatures ranging from  $+12.5^\circ\text{C}$  to  $+22.5^\circ\text{C}$  and  $\delta^{18}\text{O}_{\text{lakewater}}$  at  $0.2\text{‰}$  intervals from the

403 measured July values for individual years are used. Varves from 2001-2006 were composed  
404 of calcite, whereas varves from 2007-2010 were >75% aragonite, so Eqs. 2 and 1 were used  
405 respectively. In the majority of years, at  $\sim +20^{\circ}\text{C}$  and a  $\delta^{18}\text{O}_{\text{lakewater}}$  value from July, or 0.2‰  
406 lower, the measured  $\delta^{18}\text{O}_{\text{carbonate}}$  values match the  $\delta^{18}\text{O}_{\text{carbonate}}$  predicted from the equations  
407 (Figure 9). These temperature and  $\delta^{18}\text{O}_{\text{lakewater}}$  values are both representative of conditions  
408 around the end of June and the beginning of July, suggesting carbonate precipitation peaks at  
409 this time and that  $\delta^{18}\text{O}_{\text{carbonate}}$  in the sediment record should reflect  $\delta^{18}\text{O}_{\text{lakewater}}$  at these times.

410

411 Figure 9

412

## 413 5.2 *Inter-annual trends at Nar Gölü*

414

415 Nar Gölü experienced a period of falling lake levels between 2000 and 2010. It is possible  
416 that this was partly caused by depletion of regional groundwater levels and steepening of the  
417 hydraulic gradient north of the lake watershed. The lake may also be recovering from  
418 groundwater disturbance due to the drilling of the borehole in 1990. Additionally, climate  
419 changes will have had a significant control on water balance through this period. Based on  
420 the climate data given in section 4.3, the combination of less precipitation and hotter  
421 summers 1997-2008 would have reduced direct precipitation and shallow groundwater inflow  
422 and increased water losses through evaporation from the lake surface. The cumulative  
423 weighted 8-year P/E index from Niğde reached maximum values in 1997, decreasing to a  
424 minimum in 2005-2008 (Figure 5) (matched by the lake level decrease of  $\sim 3$  m) and then rose  
425 again in 2009 and 2010 (at which time the lake level stabilised). Whatever the precise causes  
426 of the observed lake-level fall (climate and/or pumping of groundwater), the results show that  
427 this is reflected in the monitoring and sedimentary record. There are close parallels with

428 monitoring studies of lakes Mogan and Eymir on the edge of Ankara (Özen et al., 2010).  
429 Although these two lakes have been impacted by nutrient pollution and other human actions,  
430 they also showed a very clear hydrological response to the same drought conditions recorded  
431 at Nar Gölü, from 2004 to 2007, demonstrating a region-wide hydrological response of lake  
432 ecosystems to climatic forcing. Our monitoring shows that decreasing water levels of Nar  
433 Gölü between 2000 and 2010 were associated with an increase in  $\delta^{18}\text{O}_{\text{lakewater}}$  of  $\sim 3\text{‰}$  and in  
434 lake surface water EC of  $\sim 600 \mu\text{Scm}^{-1}$  (although more EC measurements in the early 2000s  
435 would have been required to clarify that there was indeed a period of low EC at this time).

436 Changes in  $\delta^{18}\text{O}_{\text{lakewater}}$  are generally driven by changes in  $\delta^{18}\text{O}_{\text{precipitation}}$  and/or  
437 modification by evaporation within-lake (Leng and Marshall, 2004 and references therein).  
438 Here,  $\delta^{18}\text{O}_{\text{spring}}$  values are seen to represent local precipitation since they plot on the meteoric  
439 water line (Figure 2) and have remained more or less stable over the study period (Table 1),  
440 indicating that changes in  $\delta^{18}\text{O}_{\text{precipitation}}$  could not be driving the increase in  $\delta^{18}\text{O}_{\text{lakewater}}$ .  
441 Rather, the strong relationship between  $\delta^{18}\text{O}_{\text{lakewater}}$  and lake depth (Figure 5) adds weight to  
442 the suggestion that  $\delta^{18}\text{O}_{\text{lakewater}}$  trends are driven by changing water balance (e.g. Jones et al.,  
443 2005).

444 To observe how this signal has been transferred to the palaeo-limnological record,  
445 isotopic, geochemical and biological proxies have been analysed for individual lake varves  
446 from short sediment cores. There is a good match between changes in hydro-climate, lake  
447 depth and the  $\delta^{18}\text{O}_{\text{carbonate}}$  record (Figure 5). Equilibrium precipitation and a lack of diagenetic  
448 effects have already been demonstrated (section 5.1). Assuming there is always equilibrium  
449 precipitation and diagenesis never significantly alters the isotope signal, two factors should  
450 control  $\delta^{18}\text{O}_{\text{carbonate}}$ :  $\delta^{18}\text{O}_{\text{lakewater}}$  and the temperature-dependent carbonate-water fractionation  
451 effect. The very strong, positive relationship between  $\delta^{18}\text{O}_{\text{lakewater}}$  and  $\delta^{18}\text{O}_{\text{carbonate}}$  ( $r=+0.99$ ,  
452  $n=8$ ,  $p<0.005$ ) and the weighting of carbonate precipitation to the summer months indicates

453 that  $\delta^{18}\text{O}_{\text{lakewater}}$  (as we have shown, itself driven by water balance) is the key driver of  
454  $\delta^{18}\text{O}_{\text{carbonate}}$ .

455         There is evidence of an increase in the summer Mg/Ca ratio (Table 2), caused by  
456 concentration of magnesium due to evaporation and loss of calcium by precipitation of  
457 calcium carbonate (Kelts and Talbot, 1990). There was also a shift in the sediment core from  
458 calcite precipitation 1997-2006 to mostly aragonite precipitation 2007-2010. Shifts from  
459 calcite to aragonite precipitation may be associated with an increase in the Mg/Ca ratio of  
460 lake water (Müller et al., 1972; Kelts and Hsü, 1978; Ito, 2001), which favours the  
461 precipitation of aragonite over calcite (Berner, 1975; De Choudens-Sanchez and Gonzalez,  
462 2009). At Nar Gölü, the recent switch from calcite to aragonite precipitation and the decrease  
463 in the Ca/Sr ratio (Figure 5) (Tesoriero and Pankow, 1996) supports the interpretation of  
464 these proxies as indicative of a negative hydrological trend. Of note, in comparison to the  
465  $\delta^{18}\text{O}_{\text{carbonate}}$  trends, there is a threshold response from calcite to aragonite.

466         Comparison of measured EC with DI-EC shows similar overall trends, but there is an  
467 offset in absolute values (Figure 6). The intra-annual data provide a partial explanation as to  
468 why DI-EC is lower than measured EC in Nar Gölü. Whereas the EC measurements shown  
469 on Figure 6 were taken in July, much of the diatom growth is believed to occur earlier in the  
470 year, when EC is substantially lower (Figure 3). The availability of measured EC data  
471 unfortunately do not allow us to observe the actual nature of the inferred shift in conductivity  
472 post 2006, in terms of timing and rate. Some individual diatom species change earlier than  
473 the shift in the DI-EC record, albeit at a gradual rate, for example *C. anatolicus*. The  
474 observed trends in diatom assemblages may indicate a response to controls other than  
475 conductivity and/or salinity, for example changing lake habitat availability (Barker et al.,  
476 1994), and care must be taken when using such biological indicators as a proxy of mean  
477 annual conductivity (Juggins, 2013). An increase in marginal environments as lake-levels

478 fall, for example, may explain increases in periphytic taxa such as *A. minutissimum*  
479 (Woodbridge and Roberts, 2011) and/or a change in seasonal mixing regime during the  
480 period of lake-level decrease.

481 In summary, there is a correspondence through the 2000s between measured  
482 hydrological parameters on the one hand, and lake chemistry and hydro-biology  
483 reconstructed from sedimentary proxy data on the other, although parameters show different  
484 responses in terms of type (threshold vs. linear) and sensitivity to change. The P/E ratio was  
485 highest (most positive water balance) in 1997, with a marked decline after 2003, ending in  
486 2008. The lake-level decline ended in 2010.  $\delta^{18}\text{O}_{\text{lakewater}}$  and EC show a rise through the  
487 2000s with the former stabilising after 2008 in a similar way to the P/E trend (Figure 5). The  
488 shift to higher  $\delta^{18}\text{O}_{\text{carbonate}}$  from 2000 also starts to slow after 2008. In contrast, carbonate  
489 mineralogy and Ca/Sr data show threshold responses and DI-EC shows a less clear trend than  
490  $\delta^{18}\text{O}_{\text{carbonate}}$ , although changes are more linear when looking at the abundance of individual  
491 diatom species.

492

### 493 5.3 *Implications for the interpretation of palaeo-records*

494

495 Monitoring work as described here is primarily carried out to improve interpretations of long-  
496 term palaeo-hydrological records, such as those previously published from this site (Jones et  
497 al., 2006; Woodbridge and Roberts, 2011; Dean et al., 2013). In the case of Nar Gölü, the  
498 magnitude of the variability in  $\delta^{18}\text{O}_{\text{carbonate}}$  and DI-EC recorded through the monitoring period  
499 covers much of the variability seen in  $\delta^{18}\text{O}_{\text{carbonate}}$  and DI-EC over the last 1,720 years (shown  
500 by the shaded boxes on Figure 10).

501

502 Figure 10

503  
504 Viewing the monitored changes within the longer-term context of the palaeo-record  
505 highlights a number of points of interest. There is a relative lack of response in the DI-EC  
506 record compared to  $\delta^{18}\text{O}_{\text{carbonate}}$  data (Jones et al., 2006) through most of the record. In  
507 contrast, changes in diatom assemblages, reflected by the diatom zonation (derived by  
508 stratigraphically-constrained cluster analysis; Woodbridge and Roberts, 2011), do correlate  
509 well with shifts in  $\delta^{18}\text{O}$  (Figure 10), with *Achnantheidium minutissimum* increasing at  
510 AD1400, showing the same relationship as observed through the monitoring period. This  
511 shift in diatom species does not significantly alter the DI-EC reconstruction, potentially  
512 because *A. minutissimum* and other non-planktonic species respond to habitat availability as  
513 well as to EC in this system. The strength of the DI-EC reconstruction is also reduced by the  
514 lack of environmental knowledge about *Clipeoparvus anatolicus*, a dominant species in the  
515 assemblage, but a newly described species from Nar Gölü (Woodbridge et al., 2010), which  
516 is not included in the modern training set. The ordination of diatom taxa via DCA provides a  
517 summary representation of species changes at Nar Gölü. The DCA axis 1 score does,  
518 however, show a pattern similar to the DI-EC, only showing significant changes around  
519 AD500 as *Cyclotella meneghiniana* and *Staurosira construens* var *venter* are replaced by an  
520 assemblage dominated by *N. paleacea* (Woodbridge and Roberts, 2011). DCA axis 2 records  
521 change around AD1400 as *A. minutissimum* and *Synedra acus* become more dominant in the  
522 record (Figure 10).

523         The lake monitoring described here, in conjunction with a multiproxy record of past  
524 hydro-climatic change, substantially reduces the possibility of interpretive errors of the  
525 palaeo-record. Our monitoring data, and the discussion of the DI-EC here and in Woodbridge  
526 and Roberts (2011), suggest that care is needed when using the DI-EC reconstruction in terms  
527 of absolute values of conductivity change. By superimposing the range of variability in



528 different proxies during the period of lake monitoring with that shown in the palaeo-record, it  
529 is also possible to identify which periods in the past potentially lack a modern analogue. The  
530 similarity between the shifts in the monitoring period and at AD1400 now allows partial  
531 quantification of this change. Although there is no direct analogue of the changes at AD500,  
532 the record points to a lake-level increase, associated with a shift in the diatom assemblage,  
533 with a magnitude that was larger than changes in the reverse direction ~AD1400 (Figure 10)  
534 and that observed in recent times.

535

## 536 **6. Conclusions**

537

538 Using the example of Nar Gölü, we have highlighted how monitoring data can be used to test  
539 assumptions and to help produce more robust interpretations of the sediment record, although  
540 our findings could be tested further by a larger dataset based on multiple annual  
541 measurements. Due to the varved nature of the sediments, it has been possible to compare  
542  $\delta^{18}\text{O}$  from core sediments to  $\delta^{18}\text{O}$  from trap sediments to  $\delta^{18}\text{O}$  from water samples from  
543 specific years. While Nar Gölü is a non-outlet lake in a semi-arid region and therefore  
544  $\delta^{18}\text{O}_{\text{lakewater}}$  is likely to reflect water balance, monitoring is still vital to test this and to assess  
545 the response rate and magnitude of the different palaeo-hydrological proxies. The strong  
546 relationship between  $\delta^{18}\text{O}_{\text{carbonate}}$ ,  $\delta^{18}\text{O}_{\text{lakewater}}$  and changes in lake depth, and the apparent  
547 equilibrium precipitation of the carbonate, indicate that  $\delta^{18}\text{O}_{\text{carbonate}}$  at Nar Gölü is likely to  
548 provide a reliable indicator of regional hydro-climatic change over longer time periods.  
549 Based on modern response times,  $\delta^{18}\text{O}$  can offer a hydro-climatic signal of decadal-scale  
550 resolution at this lake. Other palaeo-hydrological proxies, including DI-EC and carbonate  
551 mineralogy, exhibit more complex or less easily quantified responses to changes in water  
552 balance, with a less linear response between climate change and proxy records. However,

553 these proxies offer complementary data, which provide a cross-check when conducting  
554 palaeo-hydrological reconstructions.

555 In the ‘natural laboratory’ that is offered by Nar Gölü, conditions make it possible to  
556 critically test the chain of connection from present to past, and from the lake waters to the  
557 palaeo-record. Our analyses link together the timescales of monitoring and observation on the  
558 one hand, with those of palaeo-hydrological reconstruction on the other. The conclusions  
559 drawn from this study are site-specific, and in other lakes other proxies may exhibit the  
560 clearest relationship to hydro-climate. Nonetheless, our analysis does provide a critical test of  
561 causal relationships that are often assumed, rather than demonstrated, to be the case.

562

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564

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722 **Tables**

723

724 **Table 1**  $\delta^{18}\text{O}$  from lake surface waters and the upper spring in the catchment, and EC and pH

725 values from surface lake waters

	$\delta^{18}\text{O}_{\text{lakewater}}$ surface centre ‰ VSMOW	$\delta^{18}\text{O}_{\text{lakewater}}$ surface edge ‰ VSMOW	$\delta^{18}\text{O}$ upper spring ‰ VSMOW	EC $\mu\text{Scm}^{-1}$	pH
Mar. 1997		-3.20			
Aug. 1999		-2.95		2500	7.4
July 2000		-3.22			
July 2001	-2.64		-10.55	3300	7.9
Mar. 2002		-3.14	-10.63		
July 2002	-2.42		-10.70		
July 2003	-2.50		-10.59		
May 2004	-2.73				
July 2005		-1.88			
Sep. 2006	-0.87	-1.67	-10.56	3390	7.8
July 2008	-0.57		-10.60	3380	8.3
May 2009	-1.17	-1.46		3430	8.5
July 2009	-0.56		-10.63	3370	8.2
July 2010	-0.24		-10.65	3430	8.5
June 2011	-0.81		-10.55	3390	8.2
Sep. 2011	-0.19	-0.13	-10.63	3540	8.1
Feb. 2012		-1.25		2190	7.3
June 2012		-0.75			
July 2012	-0.34	-0.39	-10.74	3500	7.8
April 2013	-0.90		-10.57	3720	7.7
April 2014	-1.10		-10.61	3333	

726

727 **Table 2** Major ion data from surface lake water samples

	Concentration meqL <sup>-1</sup>						
	SO <sub>4</sub> <sup>-2</sup>	Cl <sup>-</sup>	Na <sup>+</sup>	K <sup>+</sup>	Mg <sup>2+</sup>	Ca <sup>+2</sup>	Mg/Ca
Aug. 1999	3.2	27.4	16.5	3.7	8.5	3.0	2.8
July 2009	3.6	20.1	14.6	3.8	10.1	2.1	4.8
July 2010	3.8	22.7	16.2	3.6	15.4	1.0	15.4
June 2011	2.9	20.2	13.7	4.0	8.8	2.2	4.0
Sep. 2011	4.4	22.4	19.0	3.8	9.4	3.2	2.9
Feb. 2012	3.0	20.2	4.1	0.0	3.2	4.0	0.8
July 2012	4.1	23.9	16.9	3.8	16.5	1.2	13.8
April 2013	3.6	20.2	19.8	3.6	7.1	3.6	2.0

728

729 **Figure captions**

730

731 **Figure 1** Nar Gölü catchment, shaded grey, with bathymetric map showing the alluvial fan in  
732 the southern part of the lake and the variability in depth. **Figure 2**  $\delta D$ - $\delta^{18}O$  plot, with data  
733 from the Ankara GNIP station 1964-2009 (IAEA/WMO 2014) defining the meteoric water  
734 line. Spring waters plot on the meteoric water line, whereas lake waters plot on a local  
735 evaporation line.

736 **Figure 3** Intra-annual variability in  $\delta^{18}O$ , EC, pH and magnesium and calcium concentrations  
737 from water samples taken from the lake edge between June 2011 and July 2012, and data  
738 from temperature loggers at 5 m and 21 m depth from the same time period (the convergence  
739 of the lines in November signifies the thermal mixing of the lake and the divergence in March  
740 the stratification of the lake).

741 **Figure 4** Depth profiles of isotope and geochemical variables from different times of the year  
742 (although note profiles were not all taken in the same year), showing the changes in thermo-,  
743 chemo- and iso-clines from spring to summer.

744 **Figure 5**  $\delta^{18}O_{\text{lakewater}}$  (from July surface water samples), measured EC (from July surface  
745 water samples),  $\delta^{18}O_{\text{carbonate}}$  from NAR10 core and sediment traps, XRF-derived Ca/Sr ratio  
746 (see section 3.2 for details) and % aragonite vs. calcite from NAR10 core, plotted with  
747 changes in maximum lake depth and 8-year cumulative weighted P/E ratio from Niğde (data  
748 provided by the Turkish Meteorological Service).

749 **Figure 6** Major diatom taxa and DI-EC in NAR06 (Woodbridge and Roberts, 2010) and  
750 NAR10 cores (new data), and measured EC (from July surface water samples).

751 **Figure 7** Intra-annual variability in  $\delta^{18}O_{\text{carbonate}}$  as recorded in a sediment trap in the lake at 5  
752 m depth between summer 2001 and summer 2002.

753 **Figure 8** Sediment trap deployed in April 2013 and collected in April 2014, showing the  
754 seasonality of sedimentation in Nar Gölü.

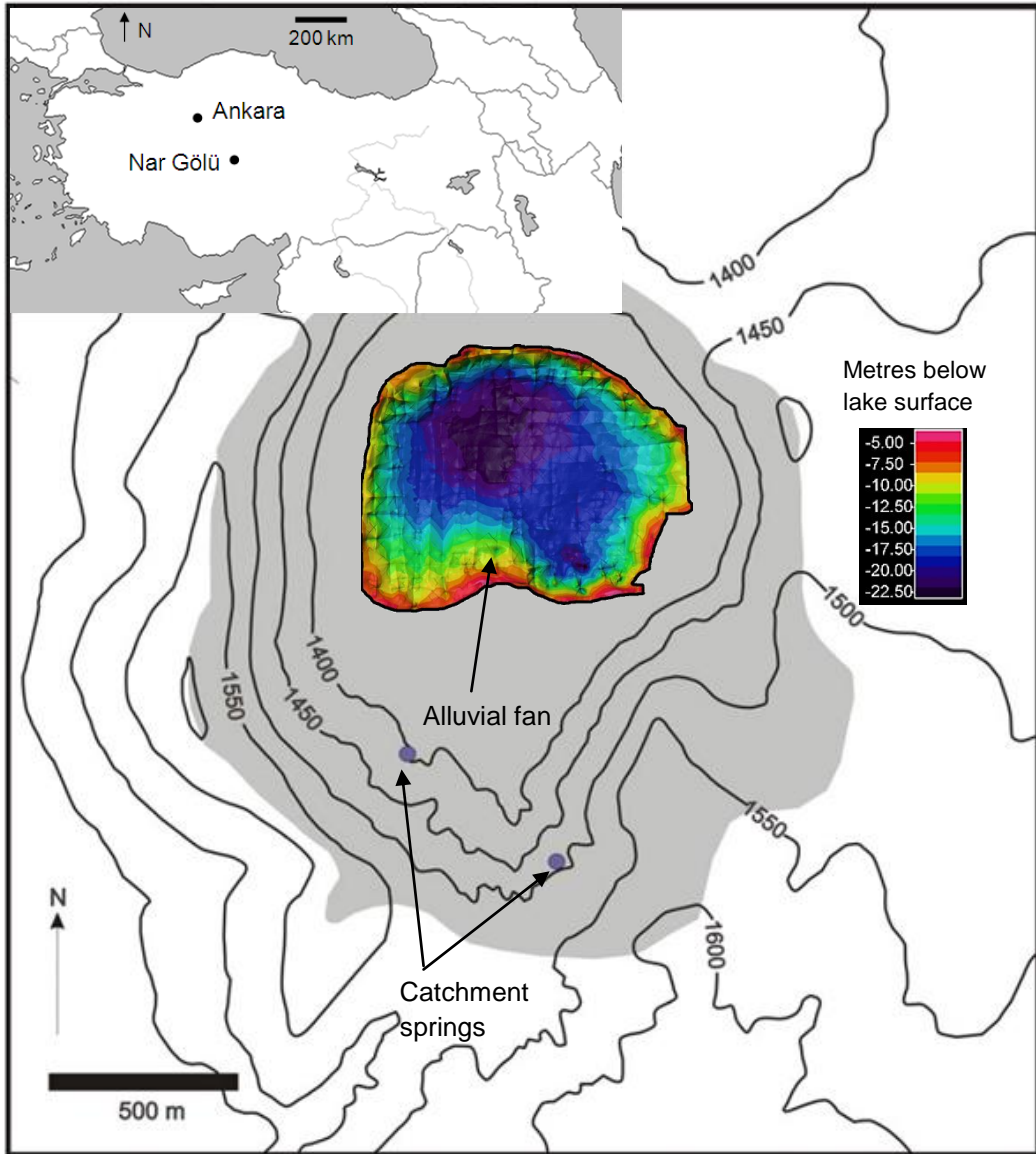
755 **Figure 9** Predicted  $\delta^{18}\text{O}_{\text{carbonate}}$  values from Eqs. 1 and 2 compared to measured  $\delta^{18}\text{O}_{\text{carbonate}}$   
756 from NAR10 core, using a variety of lake surface temperature and  $\delta^{18}\text{O}_{\text{lakewater}}$  values that  
757 represent conditions in the lake from July back to May.

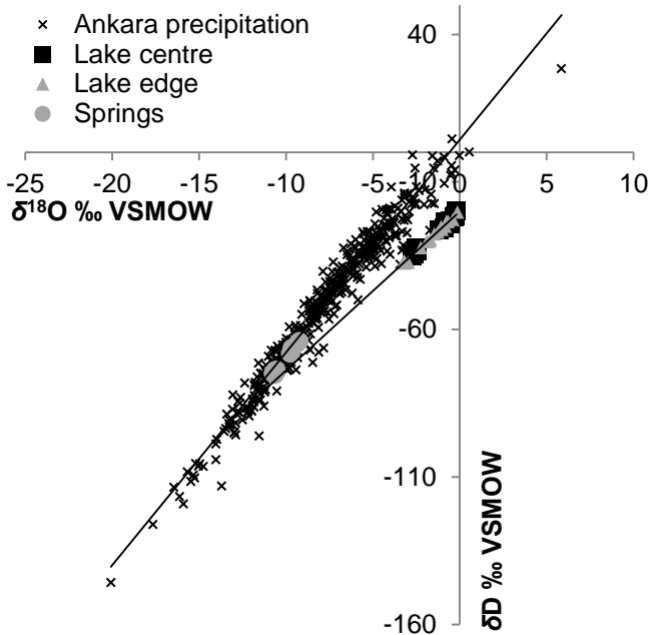
758 **Figure 10** 1,720-year records of diatom species (Woodbridge and Roberts, 2011) and  
759  $\delta^{18}\text{O}_{\text{carbonate}}$  (Jones et al., 2006) from the NAR01/02 cores. The variability in DI-EC and  
760  $\delta^{18}\text{O}_{\text{carbonate}}$  seen during the monitoring period from the NAR10 core are shown by the shaded  
761 boxes. Diatom zones from Woodbridge and Roberts (2011) are shown.

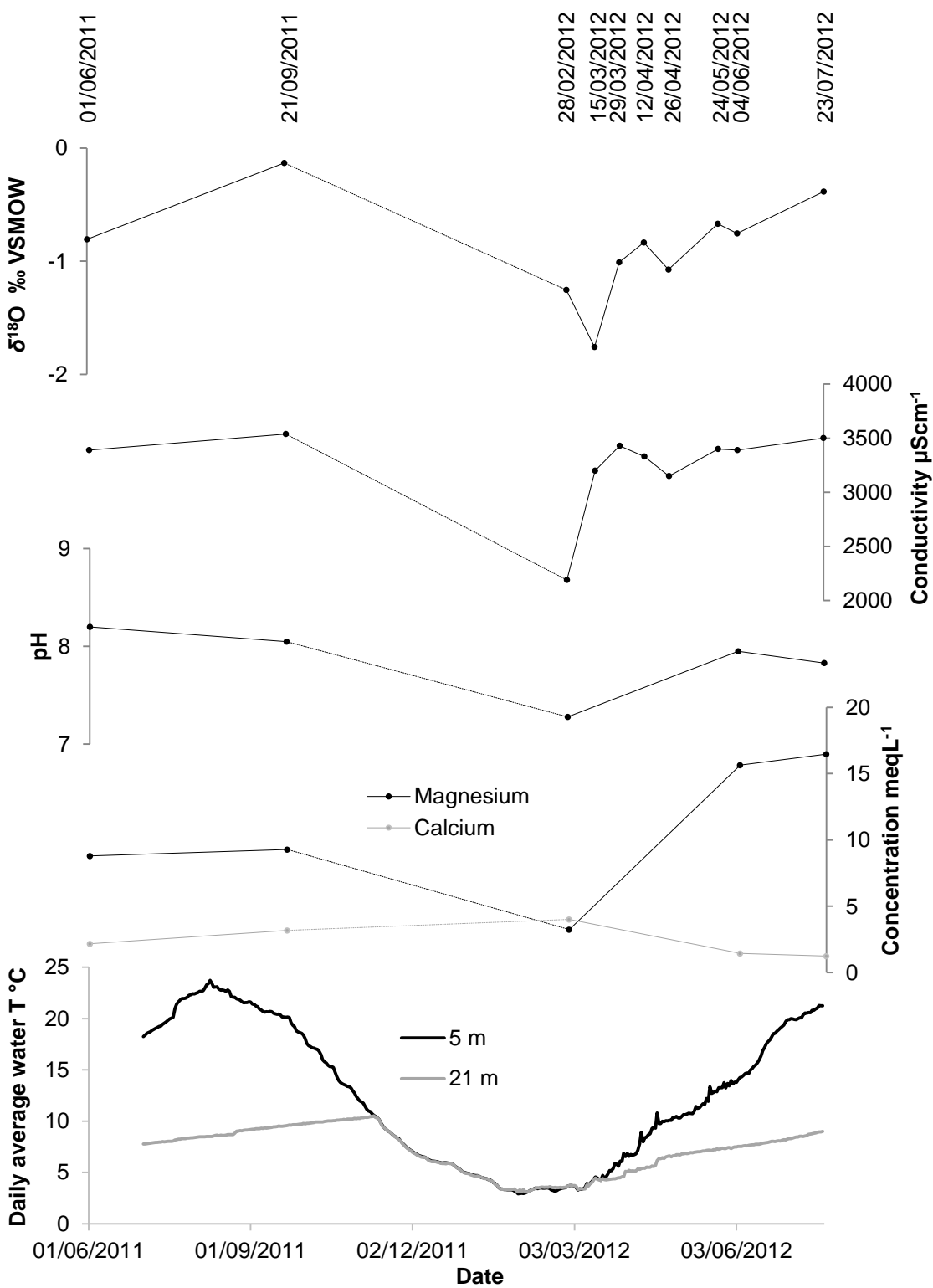
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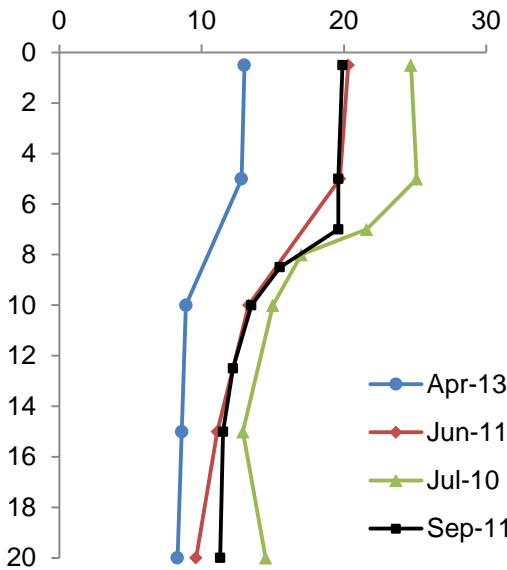
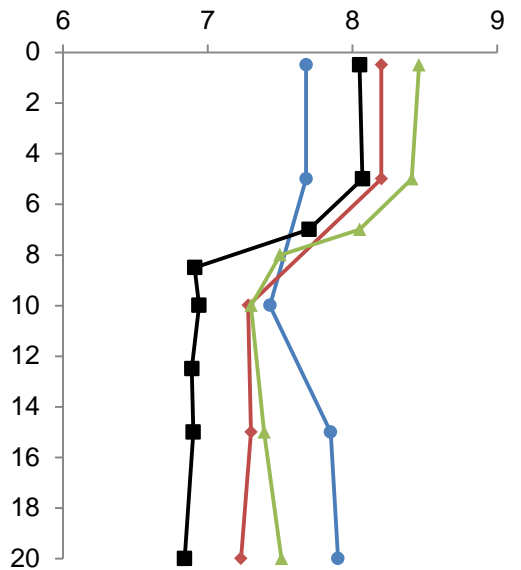
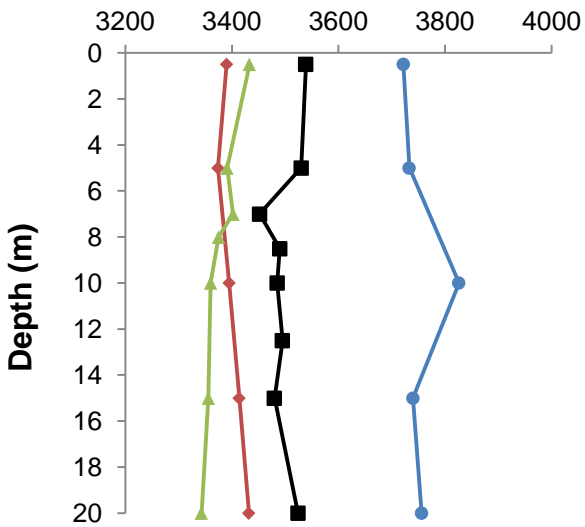
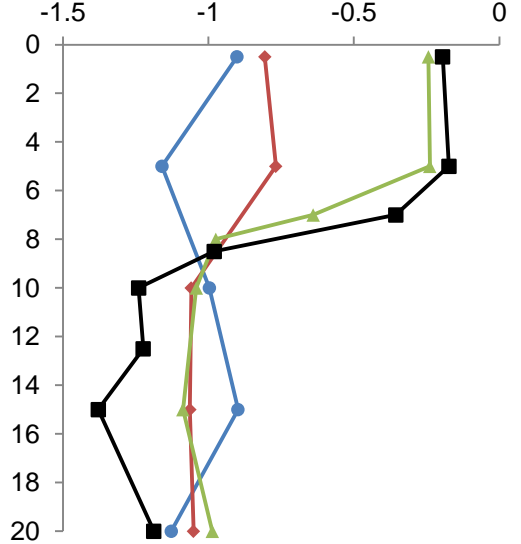
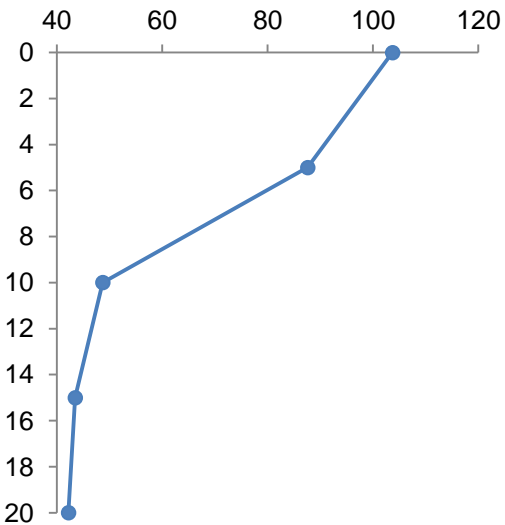
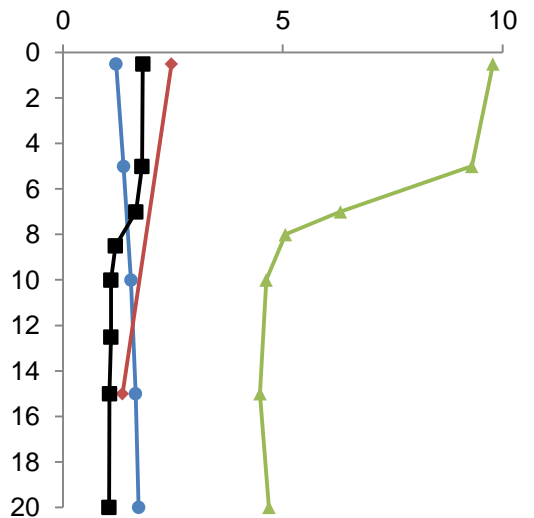
763 **SI Figure 1** Photographs of Nar Gölü between March and April 2012.

764 **SI Figure 2** ‘White-out’ around the edges of Nar Gölü in July 2012, and inset SEM image  
765 identifying this as aragonite.

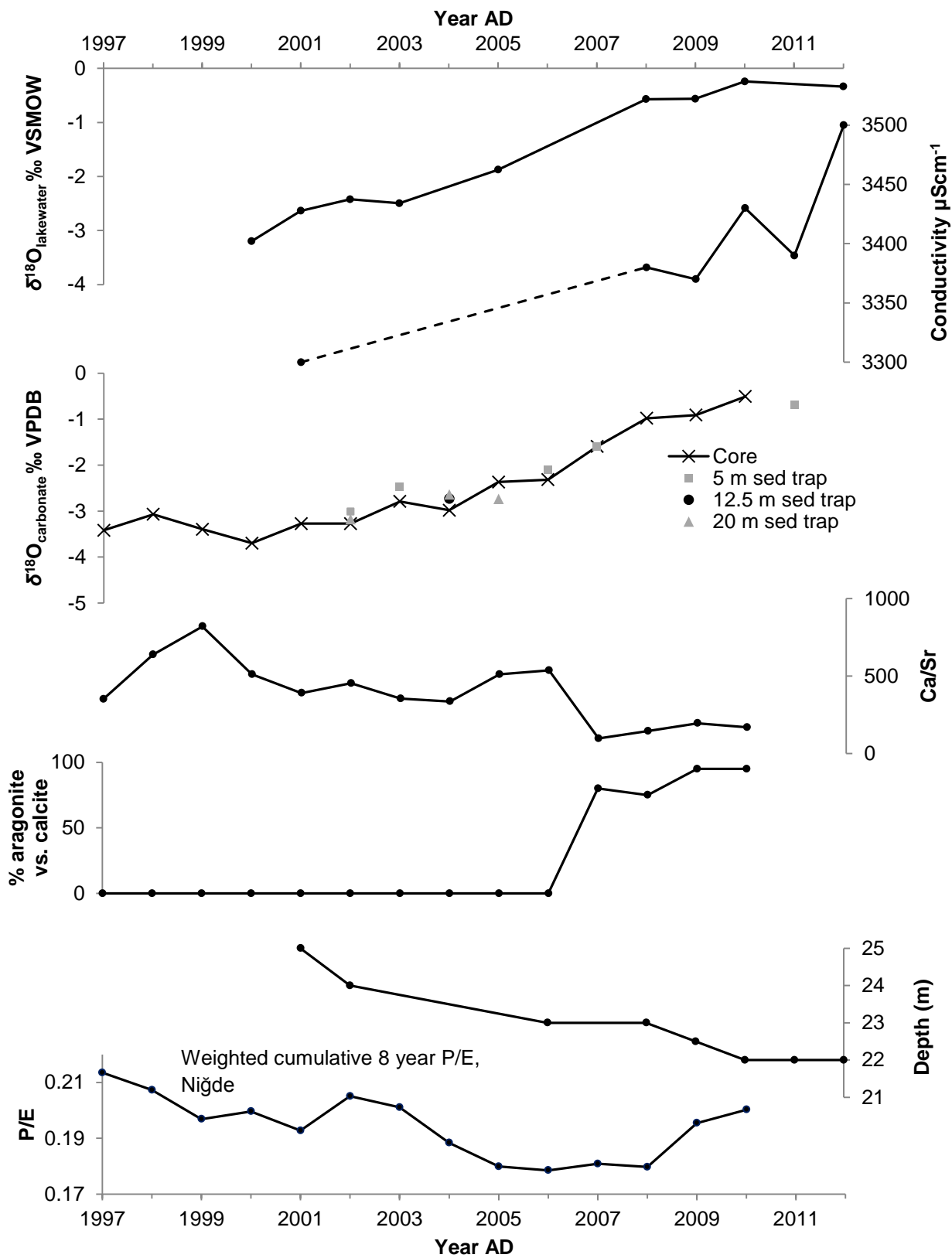


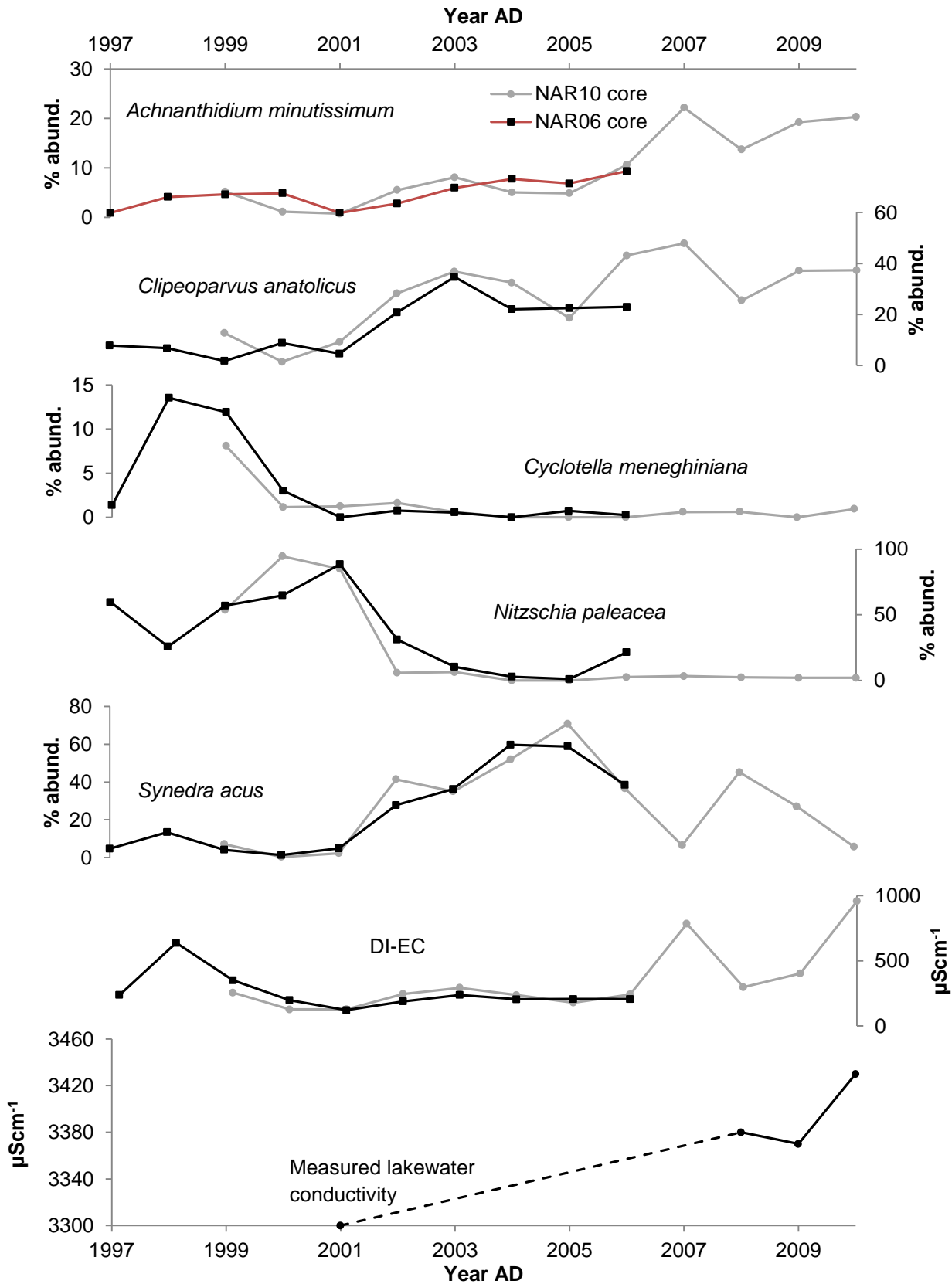




**Temperature °C****pH****Conductivity  $\mu\text{Scm}^{-1}$**  **$\delta^{18}\text{O}$  ‰ VSMOW****Dissolved oxygen %****Mg/Ca meqL<sup>-1</sup>**







$\delta^{18}\text{O}_{\text{carbonate}} \text{‰ VPDB}$

-4.0

-3.0

-2.0

0

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10

15

20

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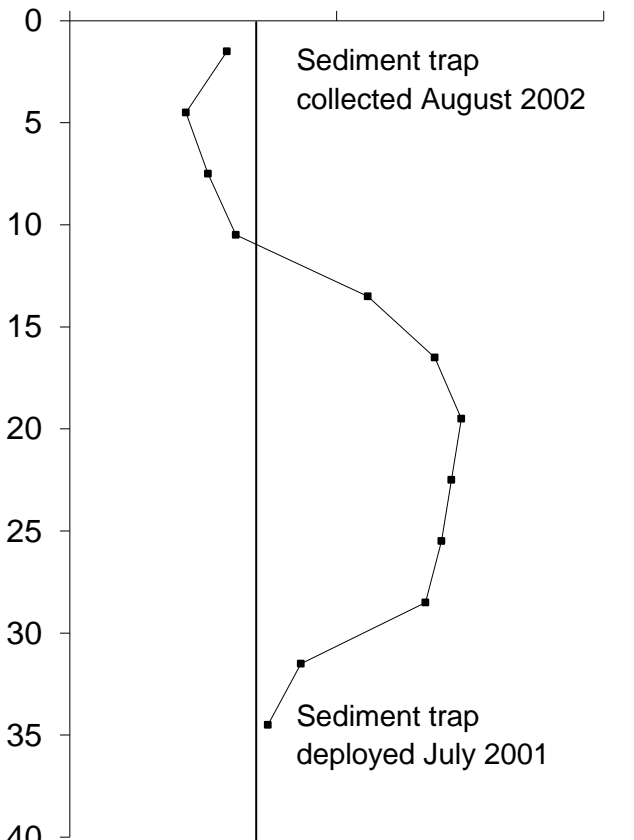
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Depth cm

Sediment trap  
collected August 2002

Sediment trap  
deployed July 2001

$\delta^{18}\text{O}_{\text{carbonate}}$  value of  
2002 core sediment  
(-3.3‰)



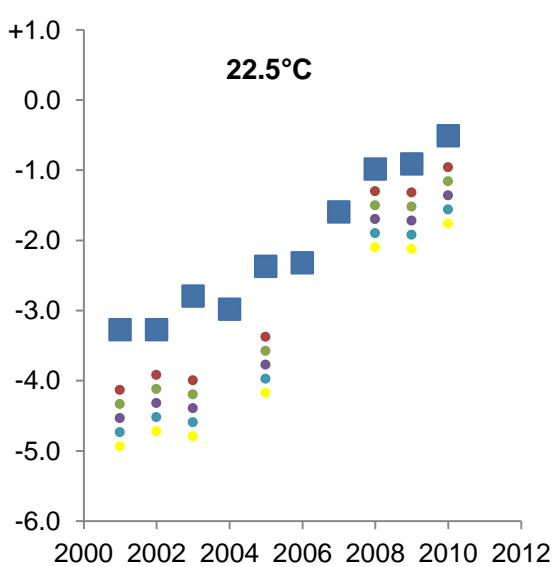
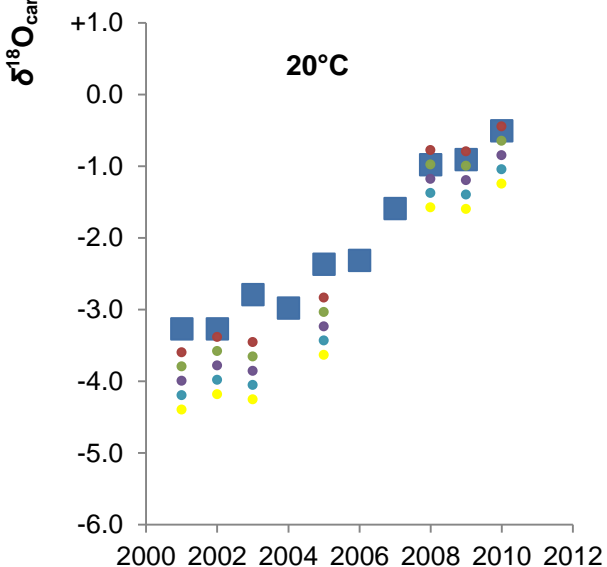
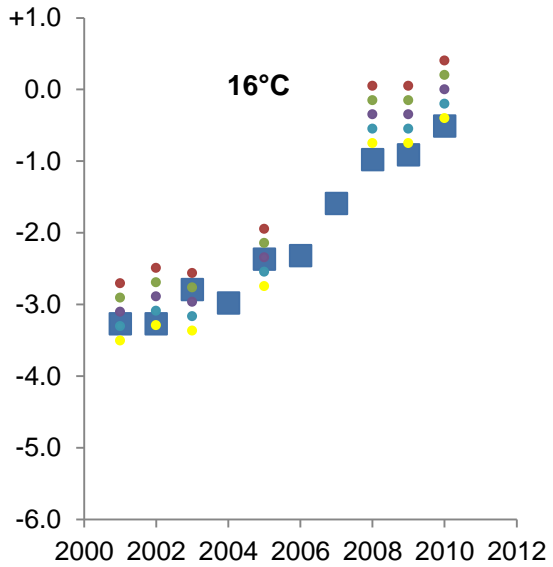
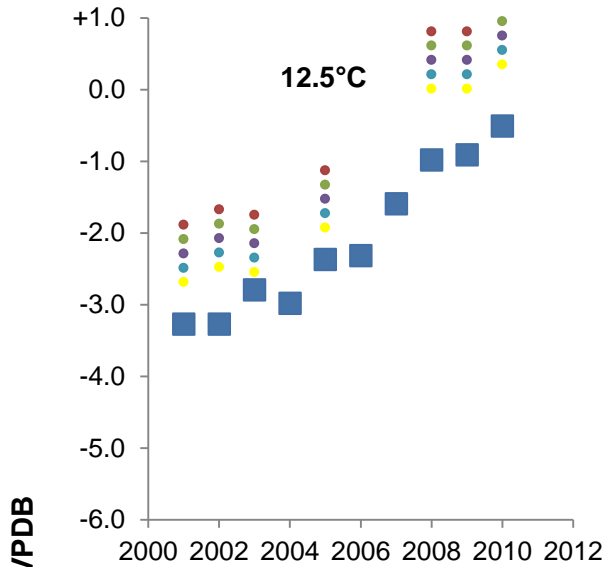
Organic material  
deposited prior to  
retrieval in April 2014

Carbonate presumed  
to have been  
deposited in summer  
2013

Organic material  
deposited after  
deployment in April  
2013



- Measured
- Predicted with July  $\delta^{18}\text{O}_{\text{lw}} -0.2\text{‰}$
- Predicted with July  $\delta^{18}\text{O}_{\text{lw}} -0.4\text{‰}$
- Predicted with July  $\delta^{18}\text{O}_{\text{lw}} -0.6\text{‰}$
- Predicted with July  $\delta^{18}\text{O}_{\text{lw}} -0.8\text{‰}$



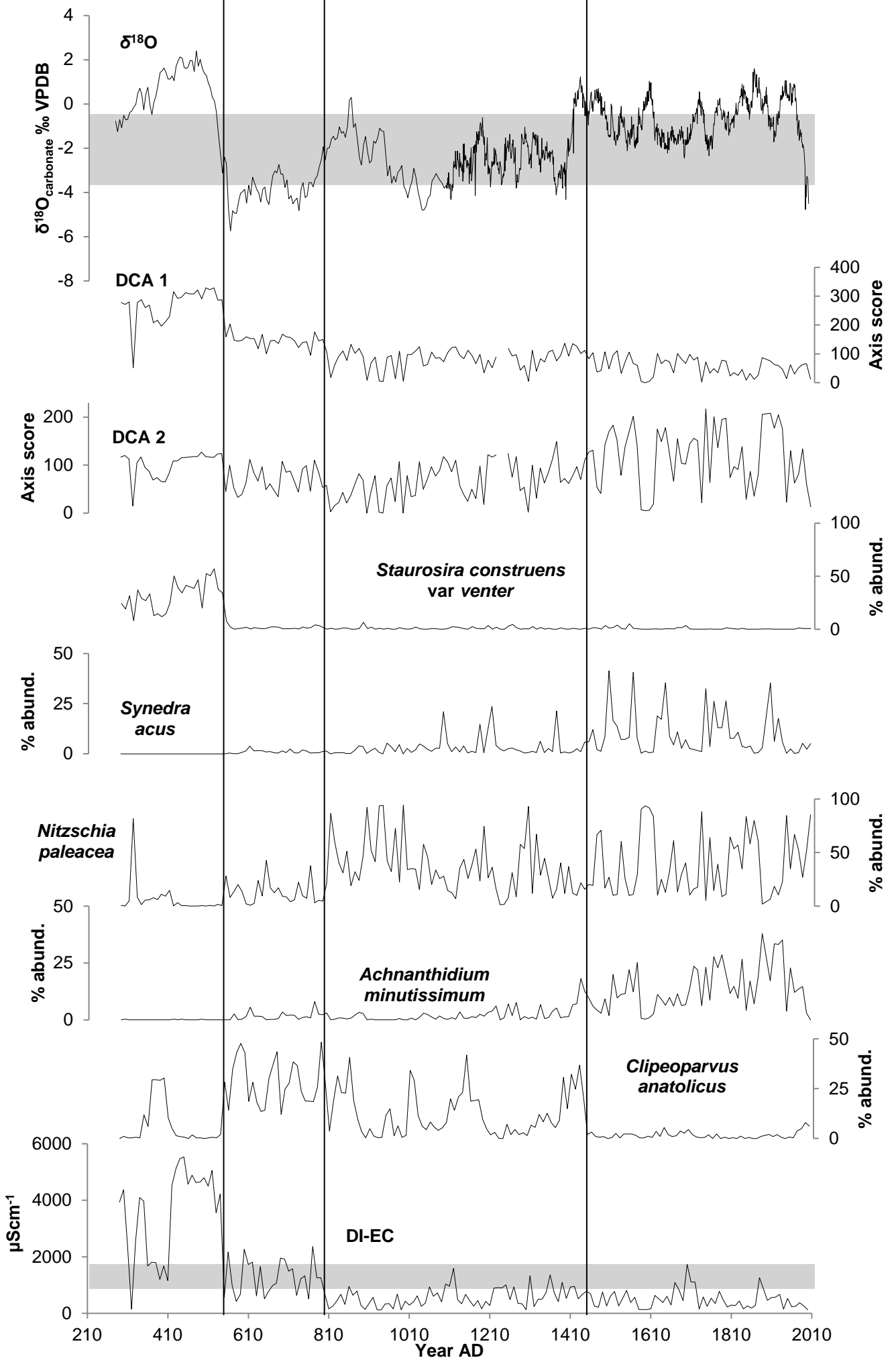
Year AD

ND1

ND2

ND3

ND4

Nar Gölü  
diatom zones

5 March 2012



19 March 2012



11 April 2012





100.0 μm