

Synthesis of superheavy elements

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Basic directions of researches at FLNR

1. Heavy and superheavy nuclei

- Synthesis and study of properties of superheavy elements
- Chemistry of new elements
- Fusion-fission and multi-nucleon transfer reactions
- Mass-spectrometry and nuclear spectroscopy of SH nuclei

2. Light exotic nuclei

- Properties and structure of light exotic nuclei
- Reactions with exotic nuclei

3. Radiation effects and physical bases of nanotechnology

4. Accelerator technology

OUTLINES

What are the SHE?

What do we know about SHE?

Targets and projectiles.

SHE – factory.

High-current cyclotron DC-280.

New facilities.

Light exotic nuclei.

Applied research.

Conclusions.

Periodic table of the elements Dmitri Mendeleev (1869)

Опыт системы элементов, расположенных по возрастанию атомного веса, с учетом химических свойств.

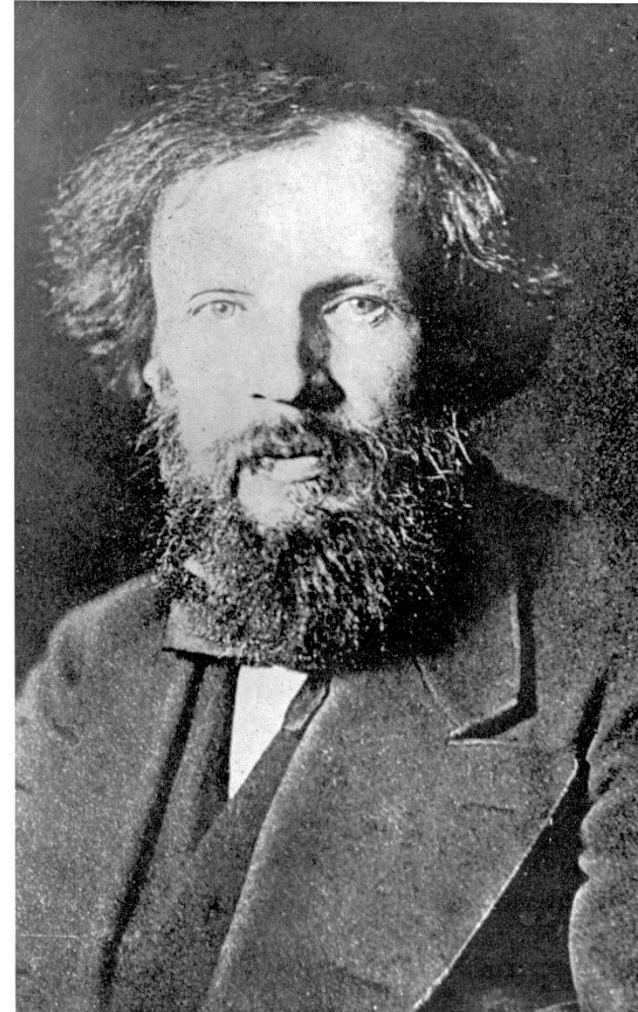
Д. Менделѣевъ.

Н=1. ?=8. ?=22. Cu=63.4. Hg=200.4. Pt=195.1. Au=197.0. Fe=55.8. Ni=58.7. Co=58.9. Mn=54.9. Zn=65.4. Ga=70.3. Ge=72.6. As=74.9. Se=78.6. Br=79.9. Kr=83.8. Rb=85.5. Sr=87.6. Y=88.9. Zr=91.2. Nb=92.9. Mo=95.9. Tc=98.9. Ru=101.1. Rh=102.9. Pd=106.4. Ag=107.9. Cd=112.4. In=114.8. Sn=118.7. Sb=121.8. Te=127.6. I=126.9. Xe=131.3. Ba=137.3. La=138.9. Ce=140.1. Pr=140.9. Nd=144.2. Pm=145.0. Sm=150.4. Eu=151.9. Gd=157.3. Tb=158.9. Dy=162.5. Ho=164.9. Er=167.3. Tm=168.9. Yb=173.0. Lu=175.0. Hf=178.5. Ta=182.0. W=183.8. Re=186.2. Os=190.0. Ir=192.2. Pt=195.1. Au=197.0. Hg=200.4. Tl=204.4. Pb=207.2. Bi=208.9. Po=209.0. At=210.0. Rn=222.0. Ac=227.0. Th=232.0. Pa=231.0. U=238.0. Np=237.0. Pu=244.0. Am=243.0. Cm=247.0. Bk=247.0. Cf=251.0. Es=252.0. Fm=257.0. Md=258.0. No=259.0. Lr=260.0.

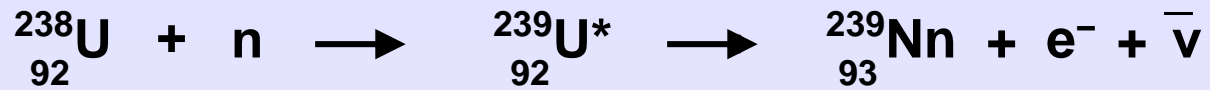
Essai d'une système des éléments d'après leurs poids atomiques et fonctions chimiques par D. Mendeleeff.

18 17/69.

Менделѣевъ

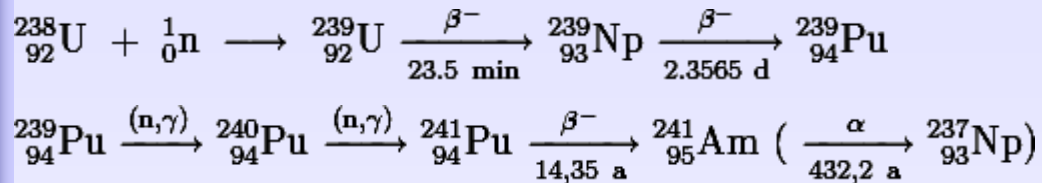


1934: search for transurane

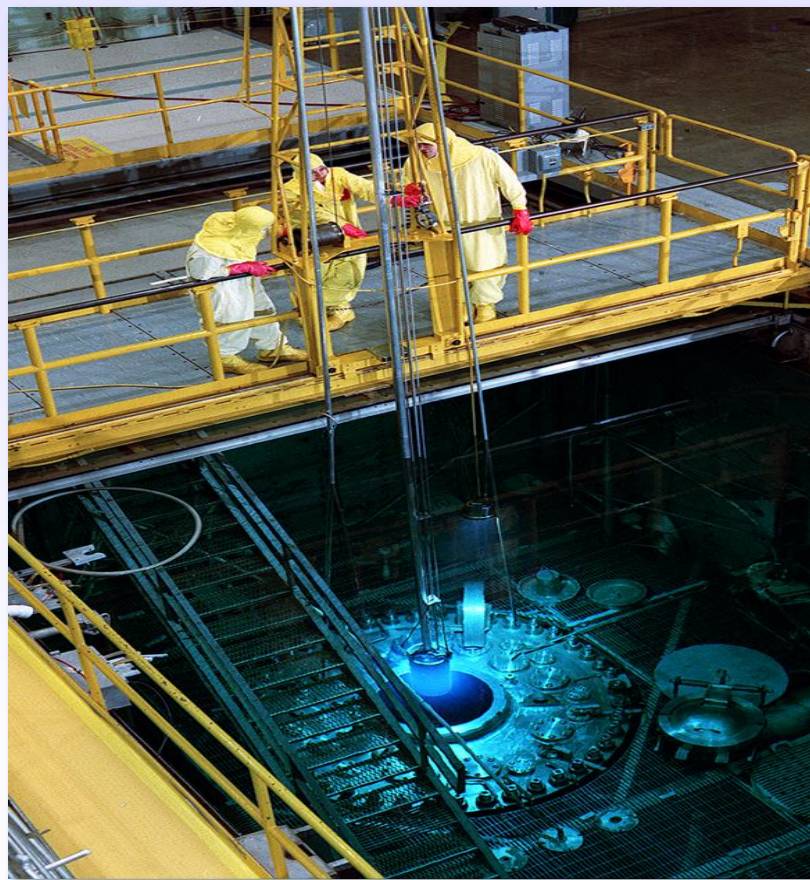


Otto Hahn und Lise Meitner

Irradiation of targets at HFIR reactor (Oak Ridge)



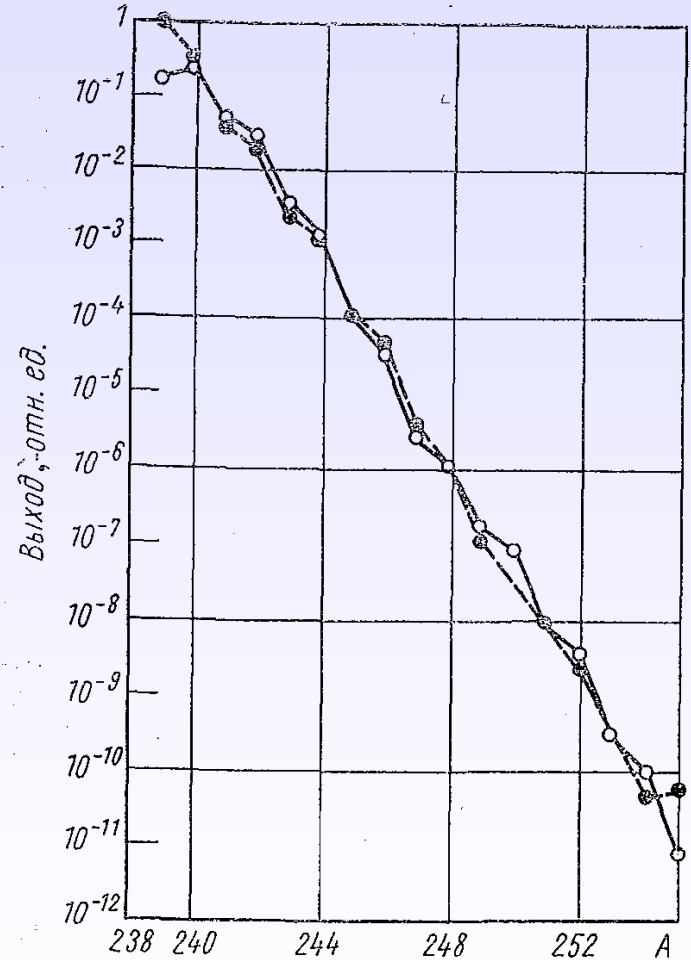
- Irradiation in the HFIR flux trap
 - Thermal-neutron flux of 2.5×10^{15} neutrons/cm²·s
 - 31 target positions (10–13 targets typically irradiated)
 - Produces ~35 mg ²⁵²Cf per target (smaller quantities of Bk, Es, Fm)



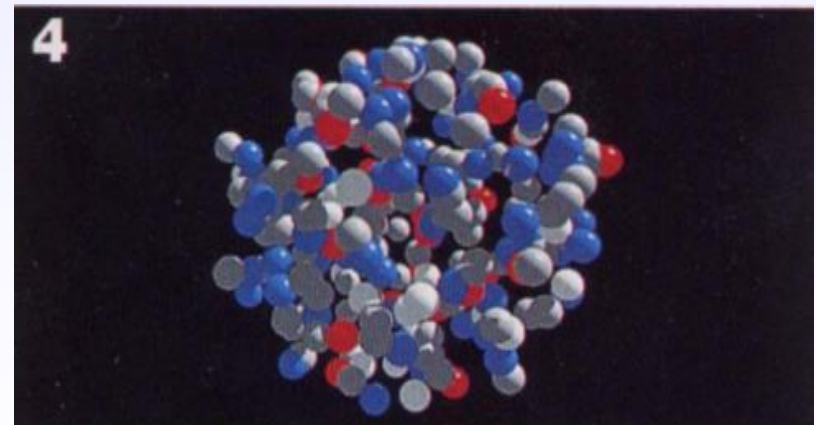
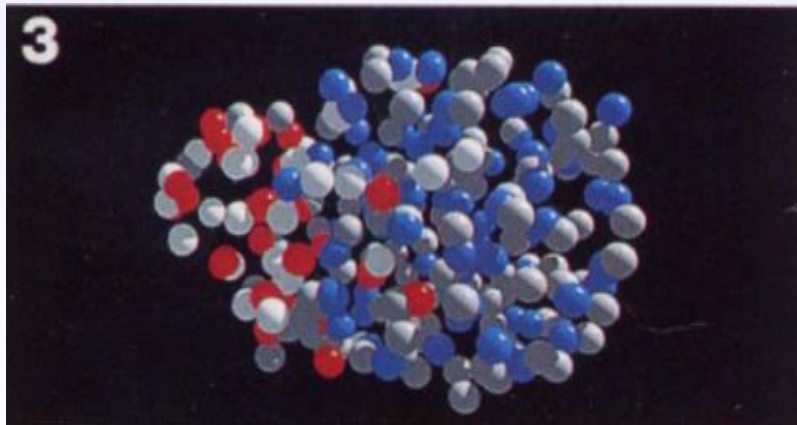
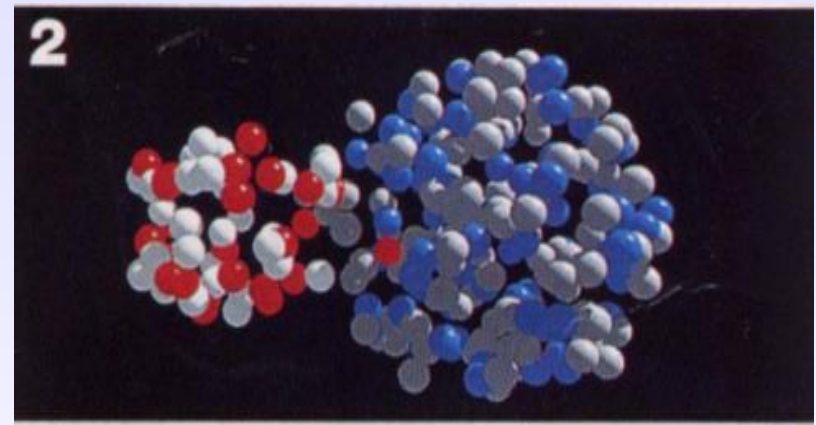
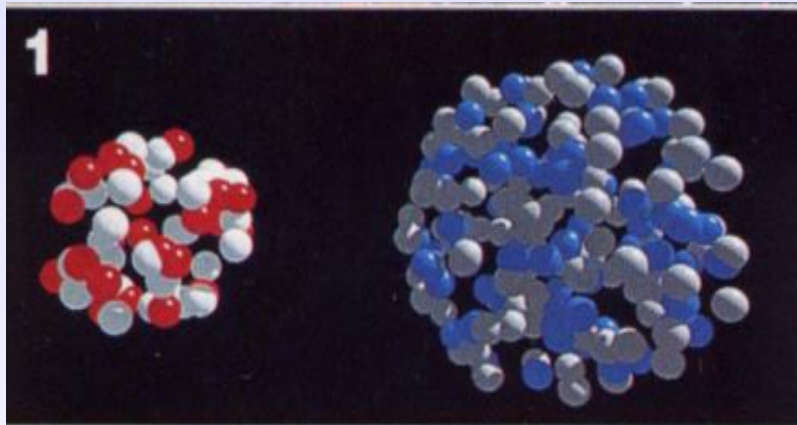
R-process



Explosion $20 \text{ kT} \rightarrow 2 \cdot 10^{24}$ fissions;
 $N_f = \exp\{n-1\} \rightarrow n = 56 \rightarrow T = 336 \text{ ns}$;
last $18 \text{ ns} \rightarrow 90 \%$ of neutrons;
thermonuclear explosion: 50 ns ;
total time: $< 100 \text{ ns}$, total neutron flux: $1.2 \cdot 10^{24} \text{ n} \cdot \text{cm}^{-2}$
But... collection efficiency $10^{-8} \div 10^{-9}$!



Complete Fusion ${}_{92}\text{U} + {}_7\text{N} \rightarrow {}_{99}\text{Es}$
instead of $\text{U} + 14\text{n} \rightarrow 7\beta \rightarrow \text{Es}$



Problems: cross-sections from 100 b \rightarrow 1 pb, amount from 1000 kg \rightarrow 1 mg

Prehistory

- 1966: A. Sobiczewski, F.A. Gareev, B.N. Kalinkin: next “magic numbers” are $Z=114$, $N=184$;
- 1966: V.M. Strutinsky; “shell correction” method;
- 1967: H.B. Meldner: next “magic numbers” are $Z=114$, $N=184$.

Accuracy of predictions:

- Spontaneous fission half-life: $T_{1/2} * 10^{\pm 10} !!$
- α -decay: $T_{1/2} * 10^{\pm 10} !!$

P. Fowler, Tracks of SHE!?



Synthesis of SHE at accelerators

- 1971; Orsay, France; $^{232}\text{Th} + ^{82}\text{Kr} \rightarrow ^{310}126 + 4n$;
 $\sigma_{4n} < 0.5 \text{ mb} !!!$
- 1971; Dubna, SU; $^{208}\text{Pb} + ^{70}\text{Zn} \rightarrow ^{276}112 + 2n$;
 $\sigma_{2n} < 0.1 \text{ mb} !!!$ (1996, GSI, Germany);
- 1971-1975; Dubna, SU; deep inelastic or fission reactions of ^{76}Ge , $^{136}\text{Xe} + ^{238}\text{U}$;
- 1975; Dubna, SU; $^{48}\text{Ca} + \text{actinides}$:
 ^{48}Ca -consumption : 50 – 100 mg/h, 200 US\$/mg;
Total amount: 5 g in SU, 5g in US.

Shell corrections

Proton shell:

114 or/and 126, 120 ?

Neutron shell:

172 or/and 184 ?

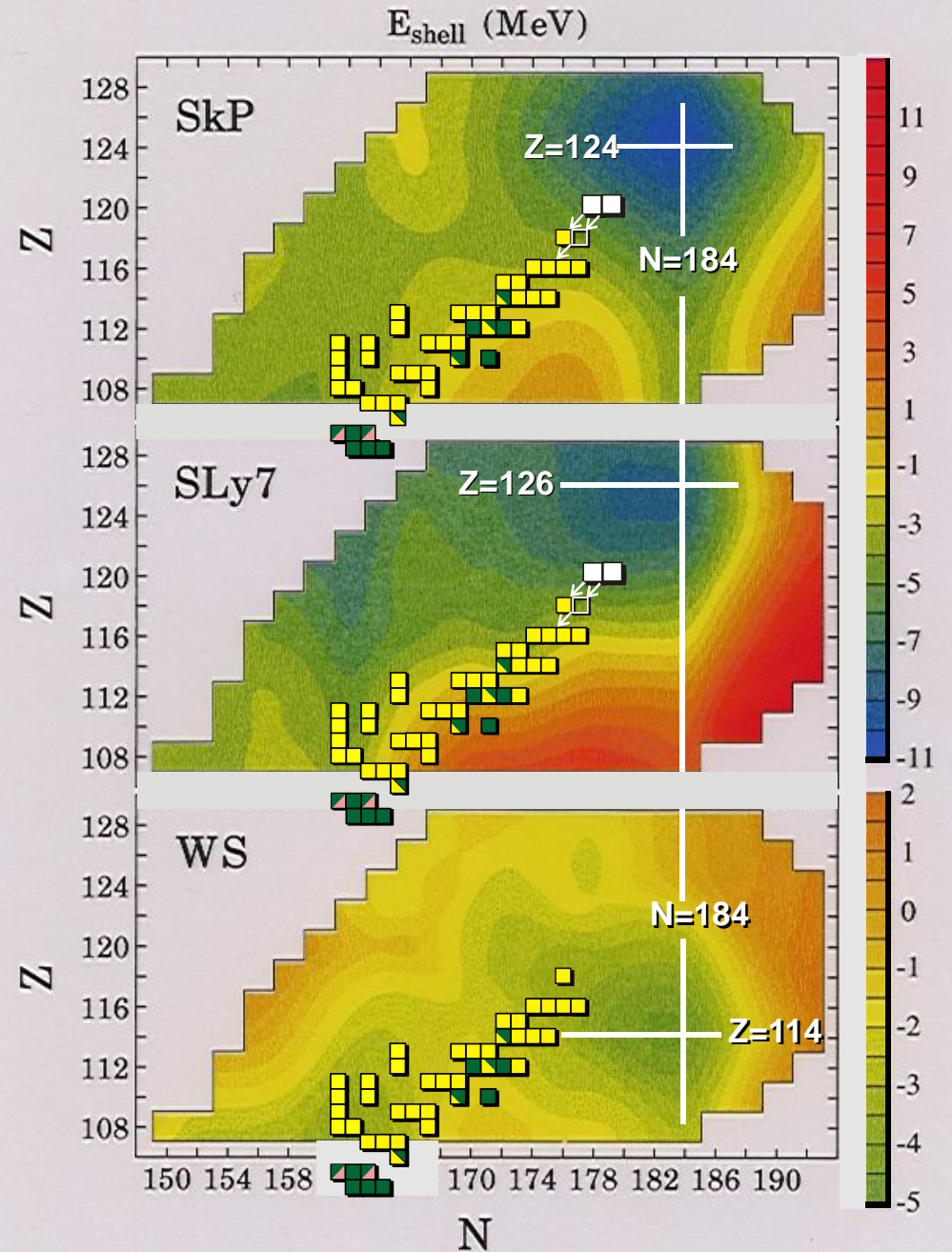
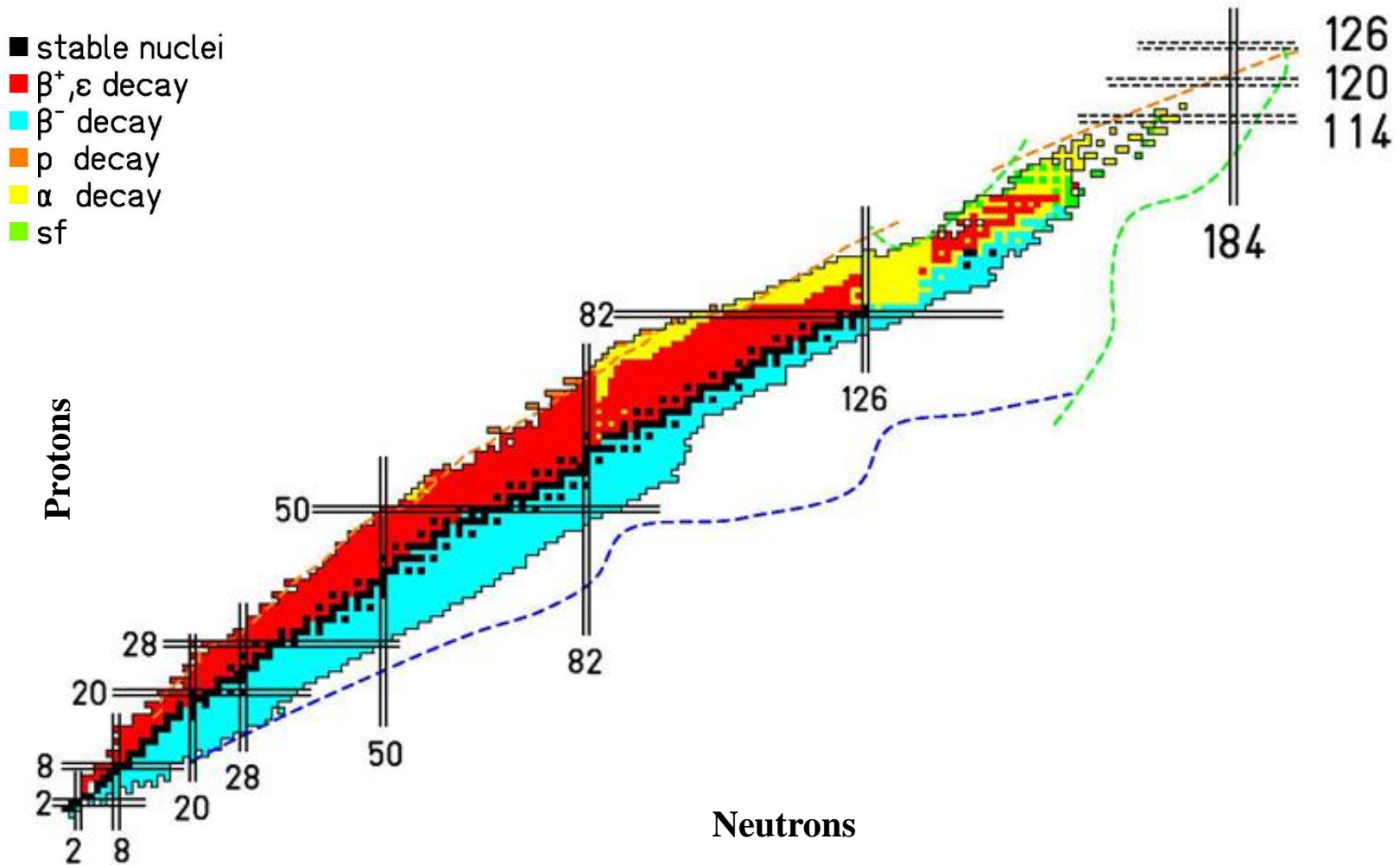
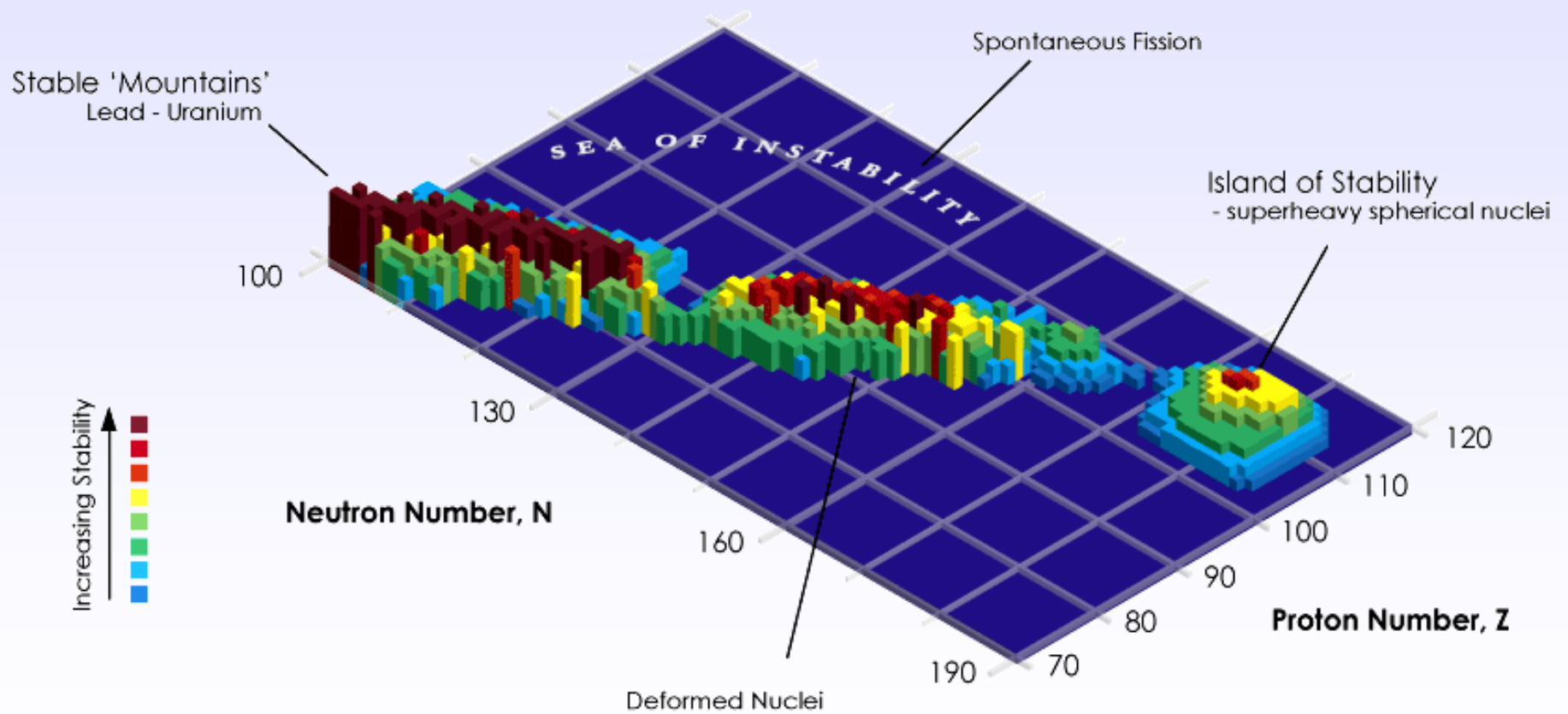


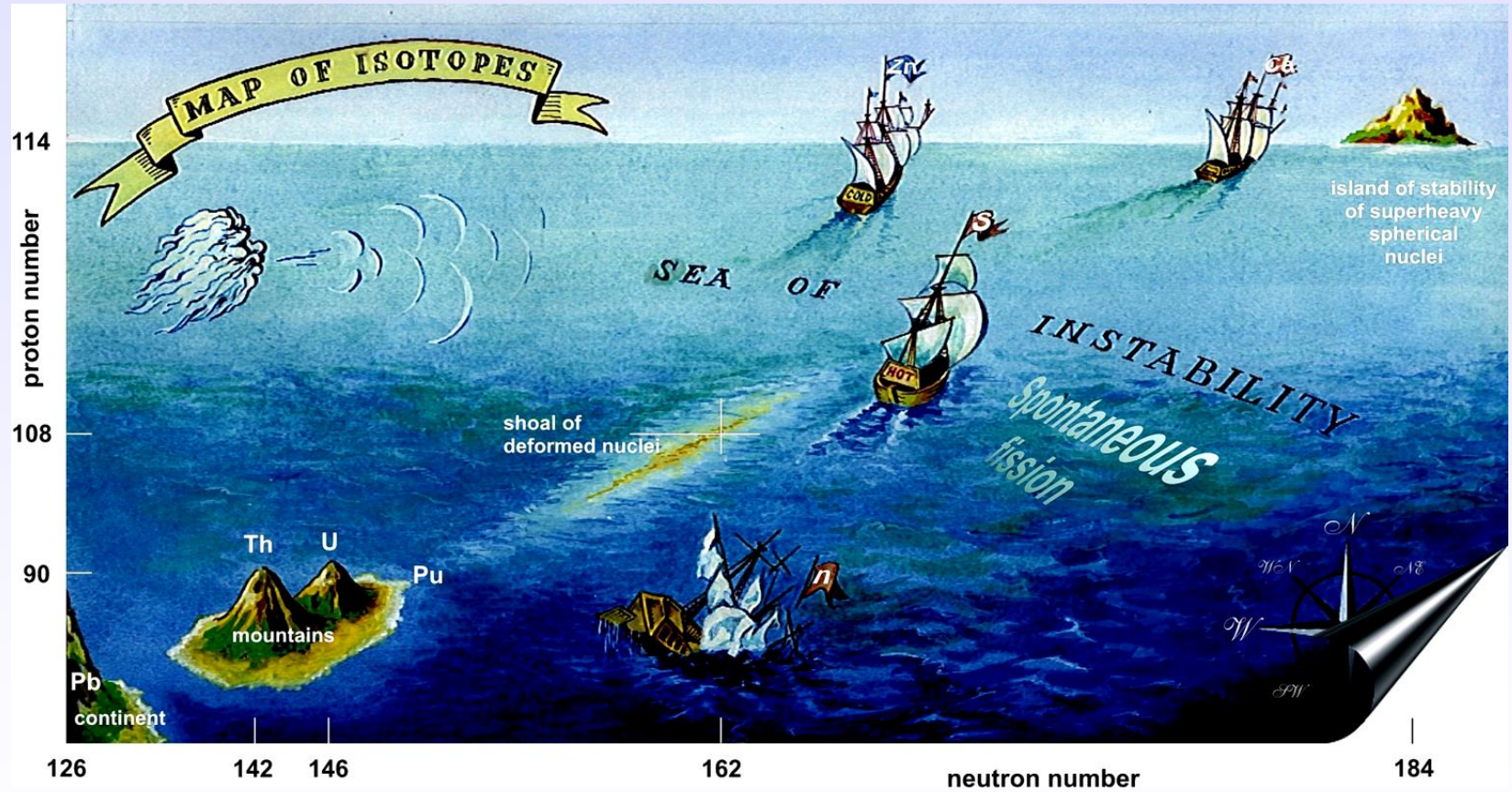
Chart of the Nuclides (decay modes)



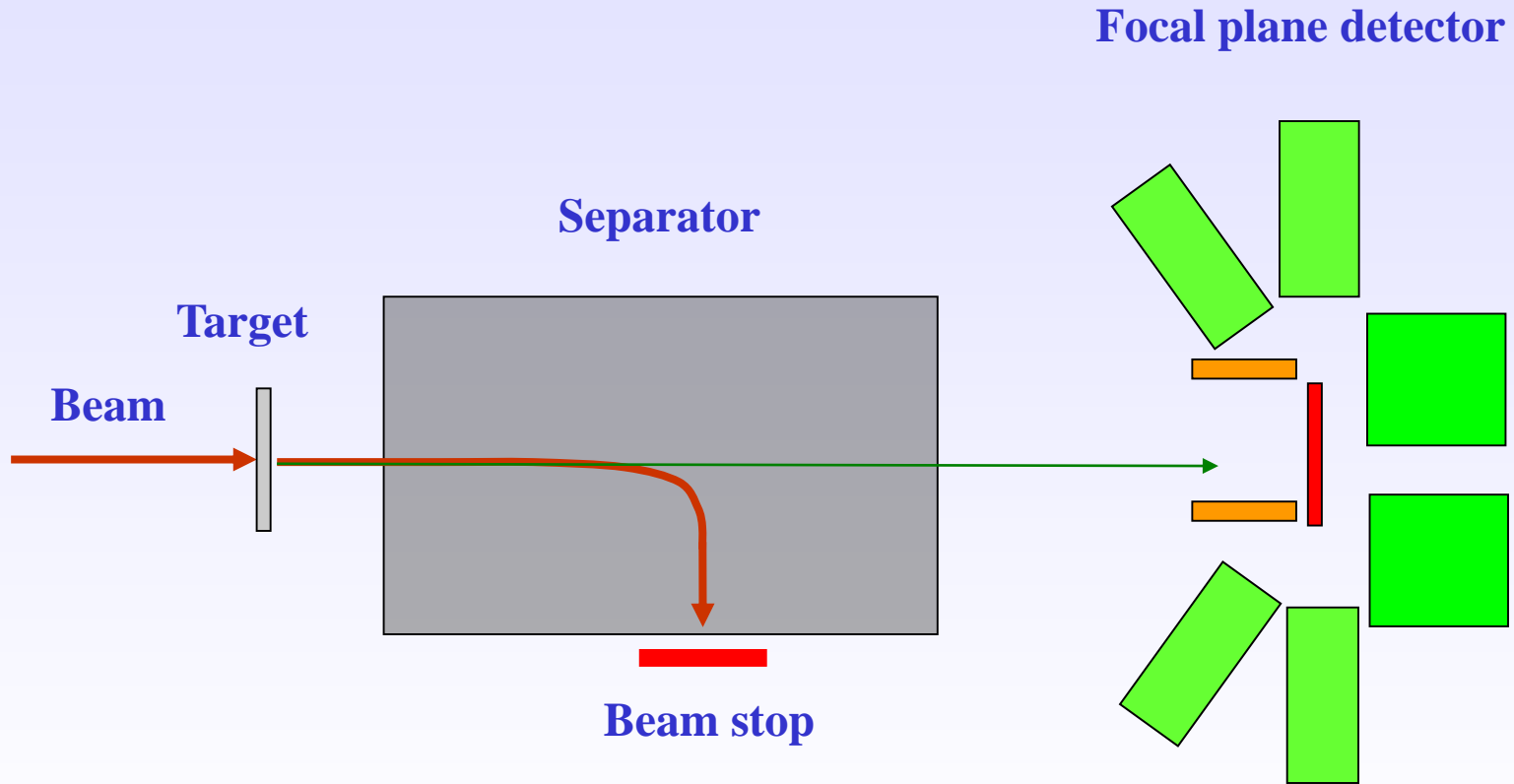
Theoretical predictions of position of SHE



Artistic Chart of the Nuclides



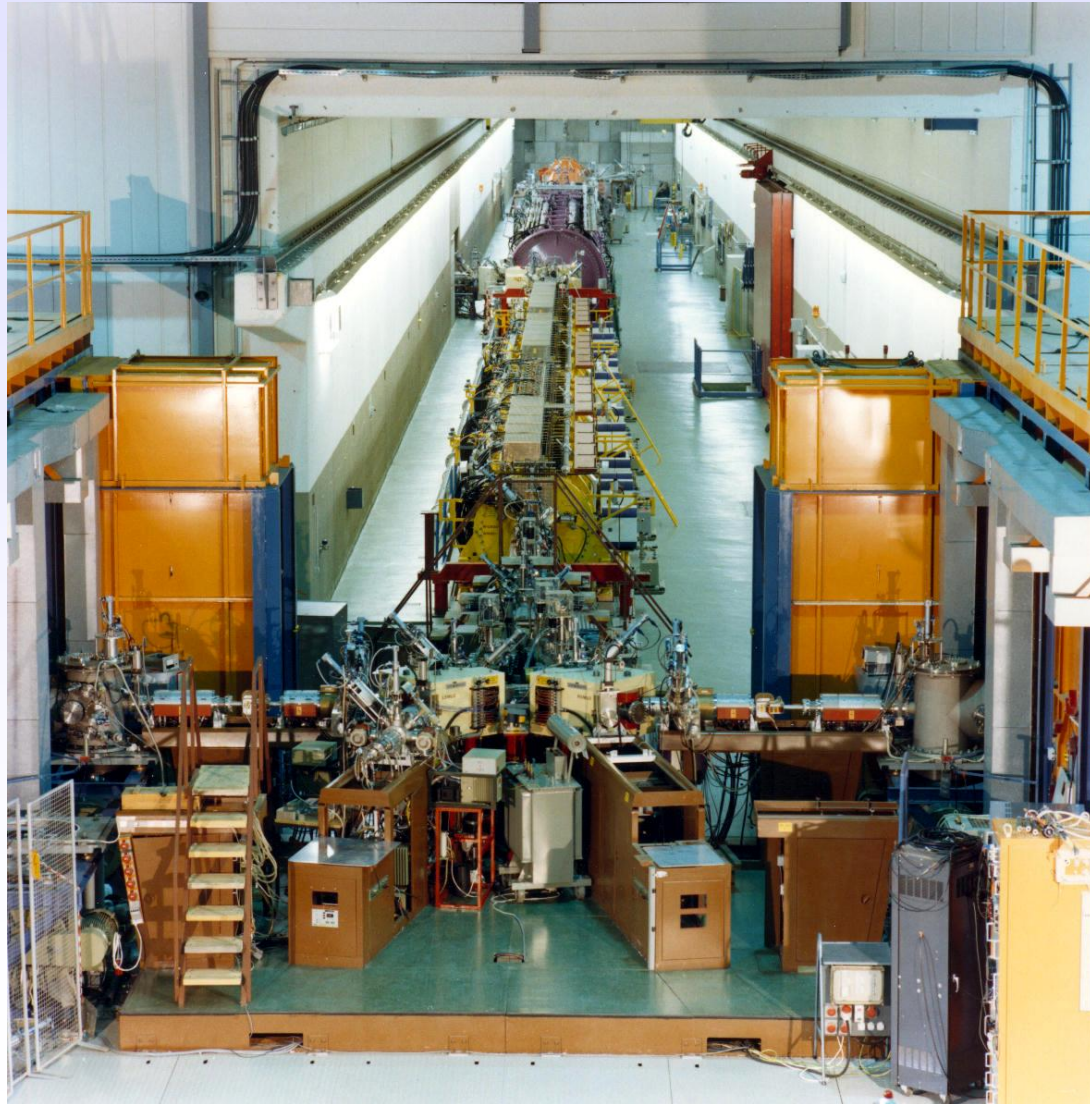
How to synthesize?



FLNR U400 cyclotron



UNILAC (GSI)



Reactions of Synthesis



Projectiles ${}^{48}\text{Ca}$ produced by
Heavy Ion Accelerator U400;

Energy: 235-250 MeV
($v \approx 0.1 c$);

Intensity: 1.0-1.5 pμA
($n \times 10^{12} \div 10^{13}$ 1/s);

Consumption: 0.5-0.8 mg/h

Beam dose: $(0.3-3.0) \cdot 10^{19}$

Prices per 1 mg

${}^{197}\text{Au} \approx 0.045$ US\$

$\text{natU}_3\text{O}_8 \approx 0.03$ US\$

${}^{239}\text{Pu} \approx 4$ US\$

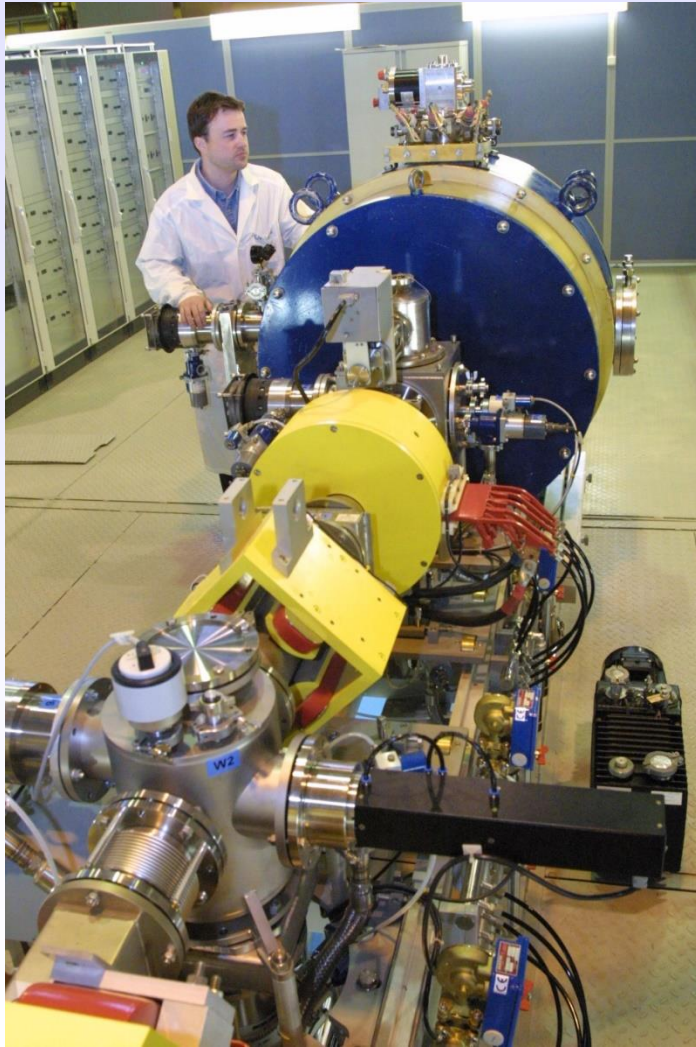
${}^{48}\text{Ca} \approx 80$ US\$

${}^{249}\text{Cf} \approx 60,000$ US\$

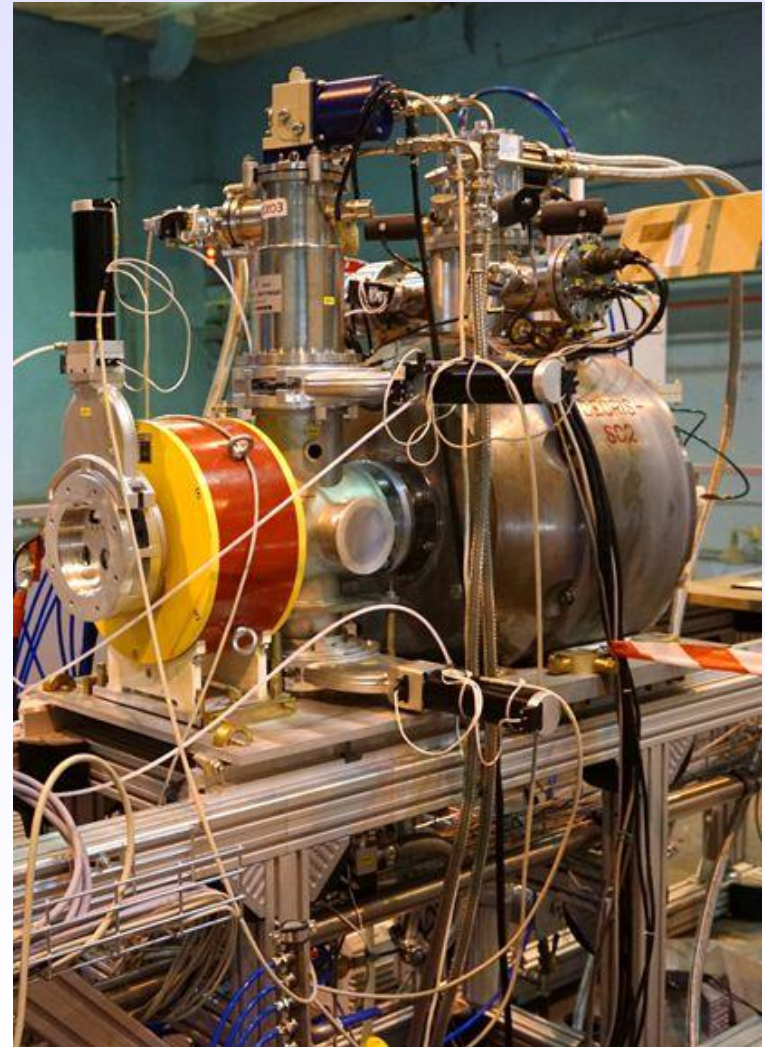
Target materials	Producer	Isotope enrichment (%)
${}^{232}\text{Th}$	----	100
${}^{233}\text{U}$	RFNC	---
${}^{238}\text{U}$	----	99.3
${}^{237}\text{Np}$	IAR	99.3
${}^{239}\text{Pu}$	RFNC	---
${}^{240}\text{Pu}$	IAR/ORNL	99.98
${}^{242}\text{Pu}$	RFNC/ORNL	99.98
${}^{244}\text{Pu}$	ORNL	98.6
${}^{243}\text{Am}$	IAR / ORNL	99.9
${}^{245}\text{Cm}$	IAR	98.7
${}^{248}\text{Cm}$	IAR / ORNL	97.4
${}^{249}\text{Bk}$	ORNL	≥ 95
${}^{249}\text{Cf}$	IAR/ORNL	97.3
${}^{249,250,251}\text{Cf}$	ORNL	(50+14+36)%

Superconducting 18 GHz ECR ion sources

DECRIS-SC1

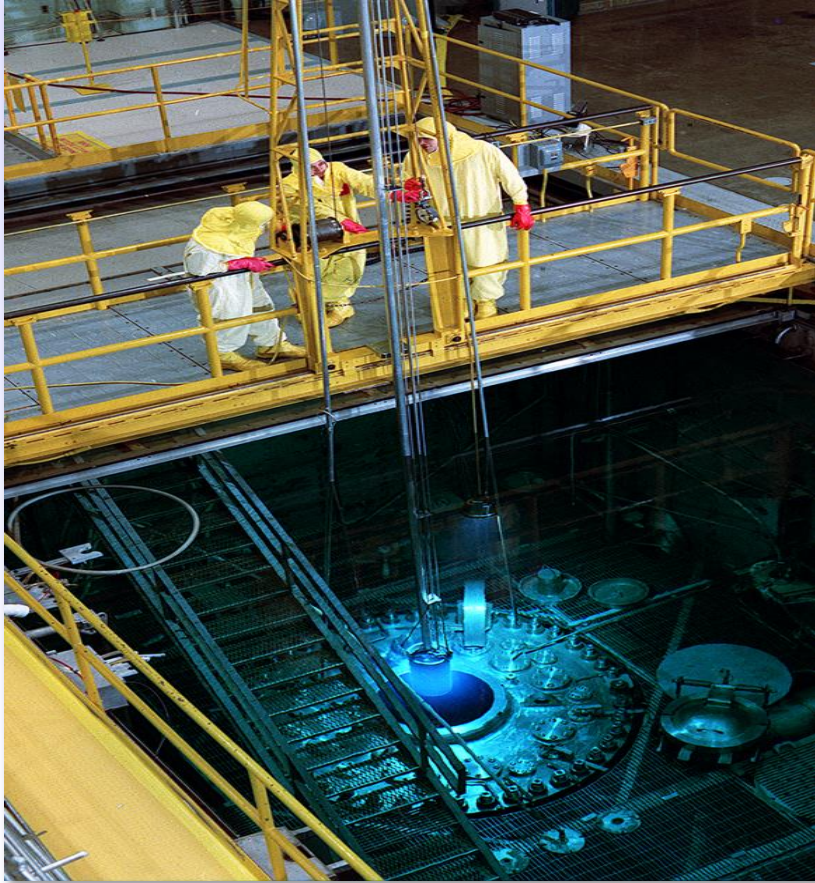


DECRIS-SC2

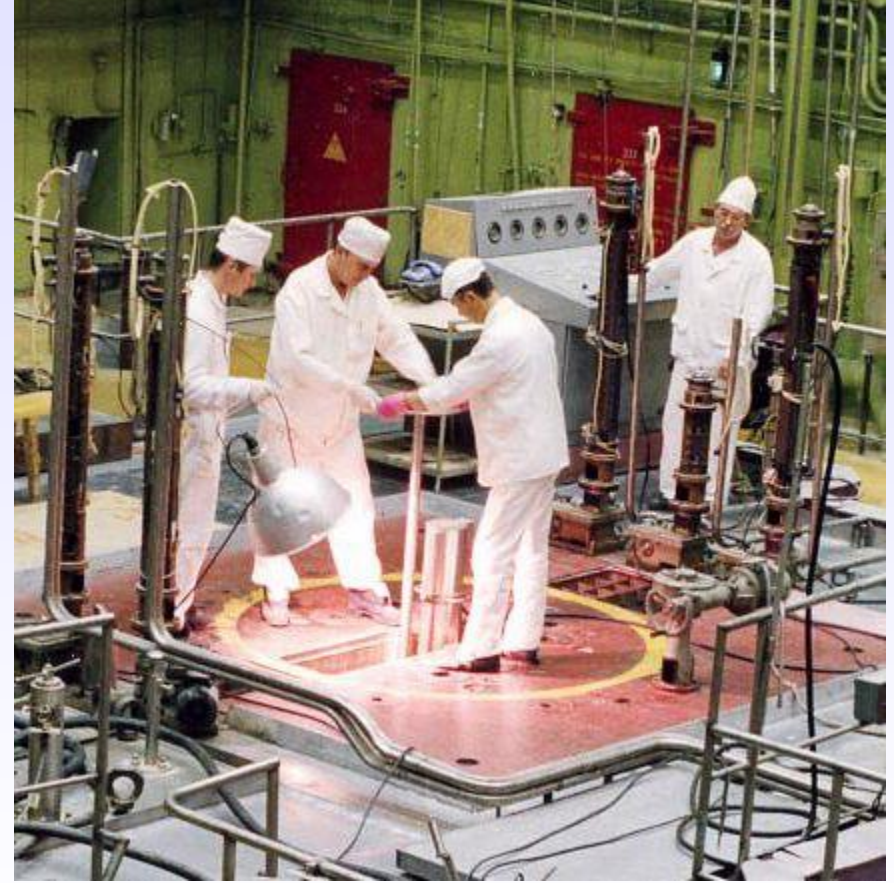


Isotope reactors

HFIR, ORNL, Oak Ridge, USA, 85 MW



CM-3, IAR, Dimitrovgrad, RF, 100 MW



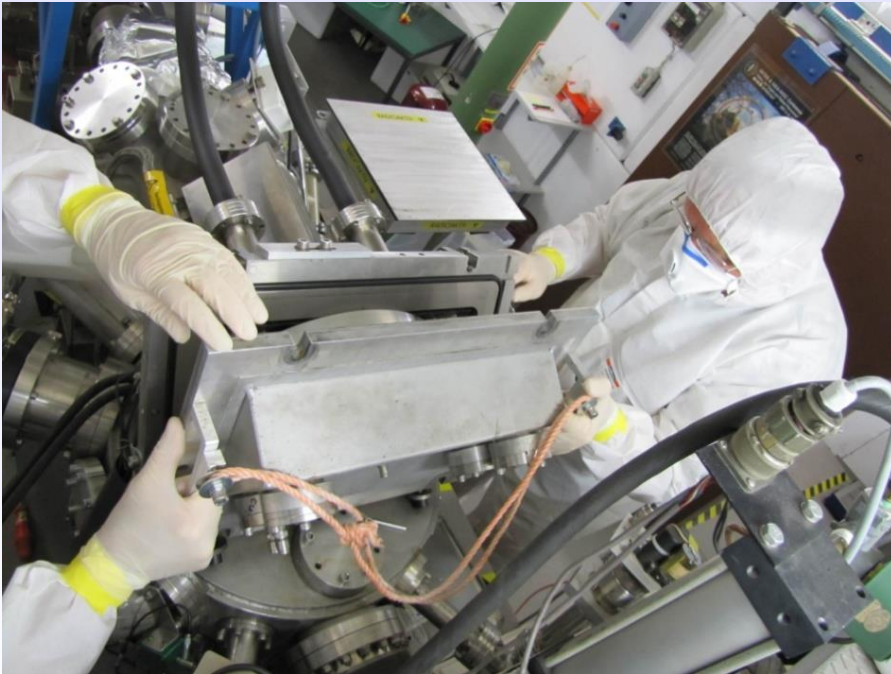
Isotope separator is necessary !

22 mg of ^{249}Bk , \approx 1 M\$, 1 year at HIFR ORNL



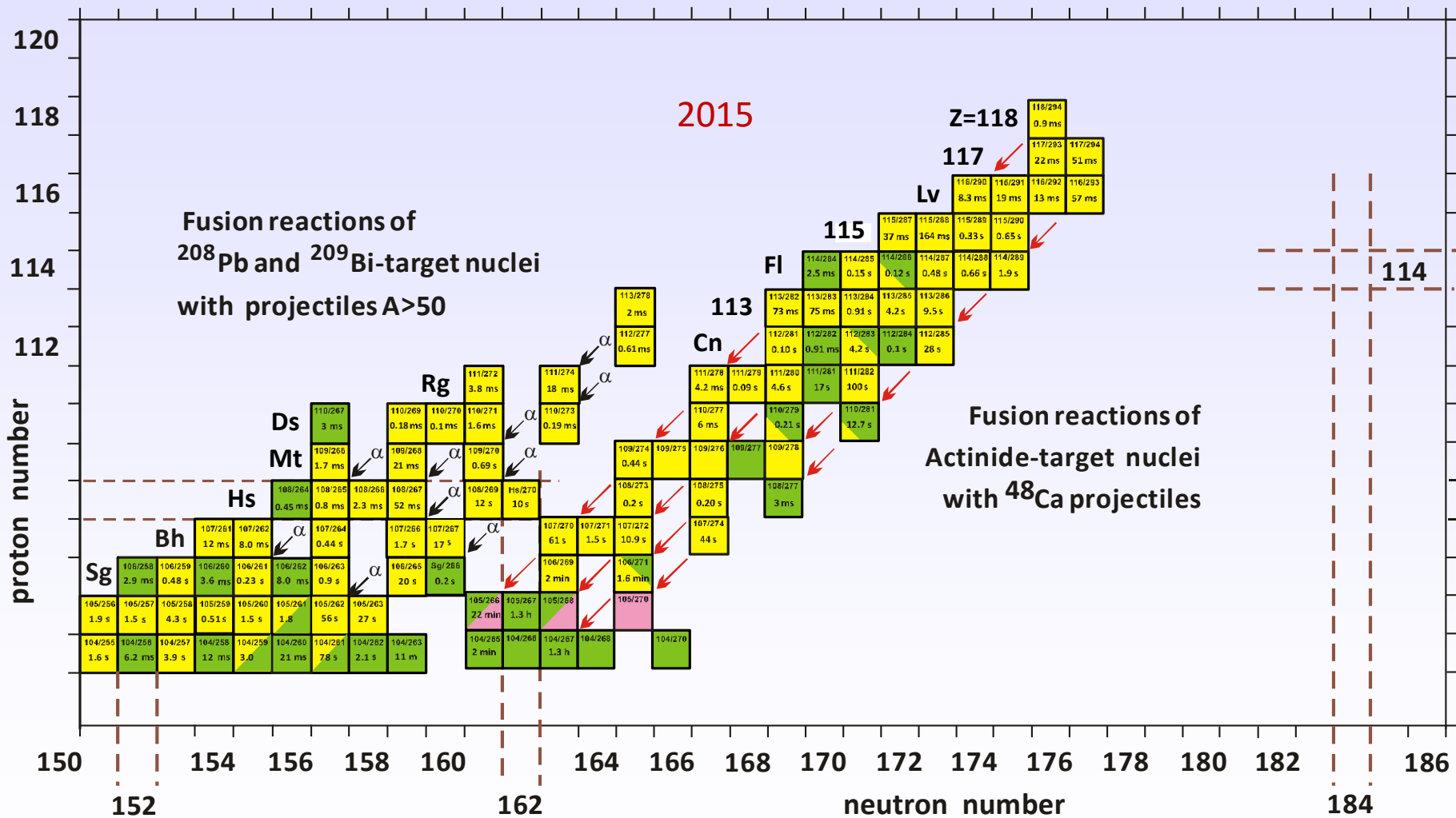
$\text{Bk}(\text{NO}_3)_3$ Product

Targets – radiation safety



Dubna Gas Filled Recoil Separator



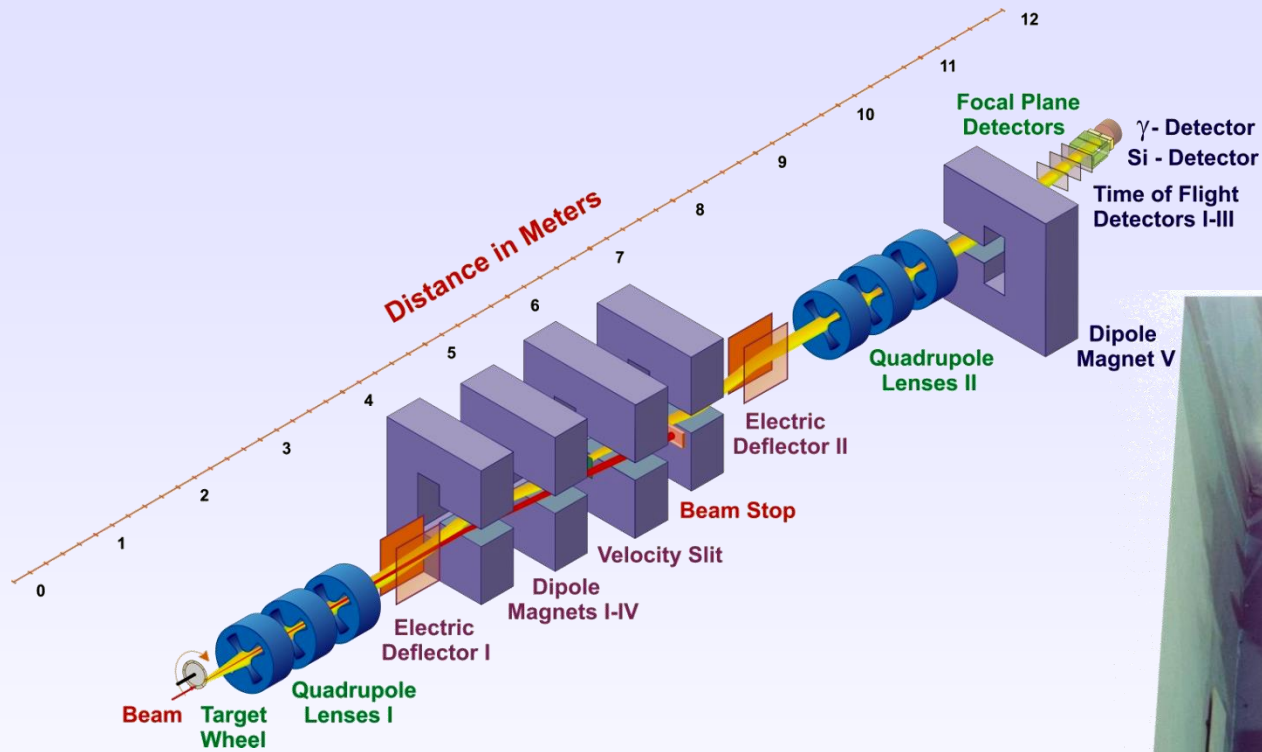


Confirmations

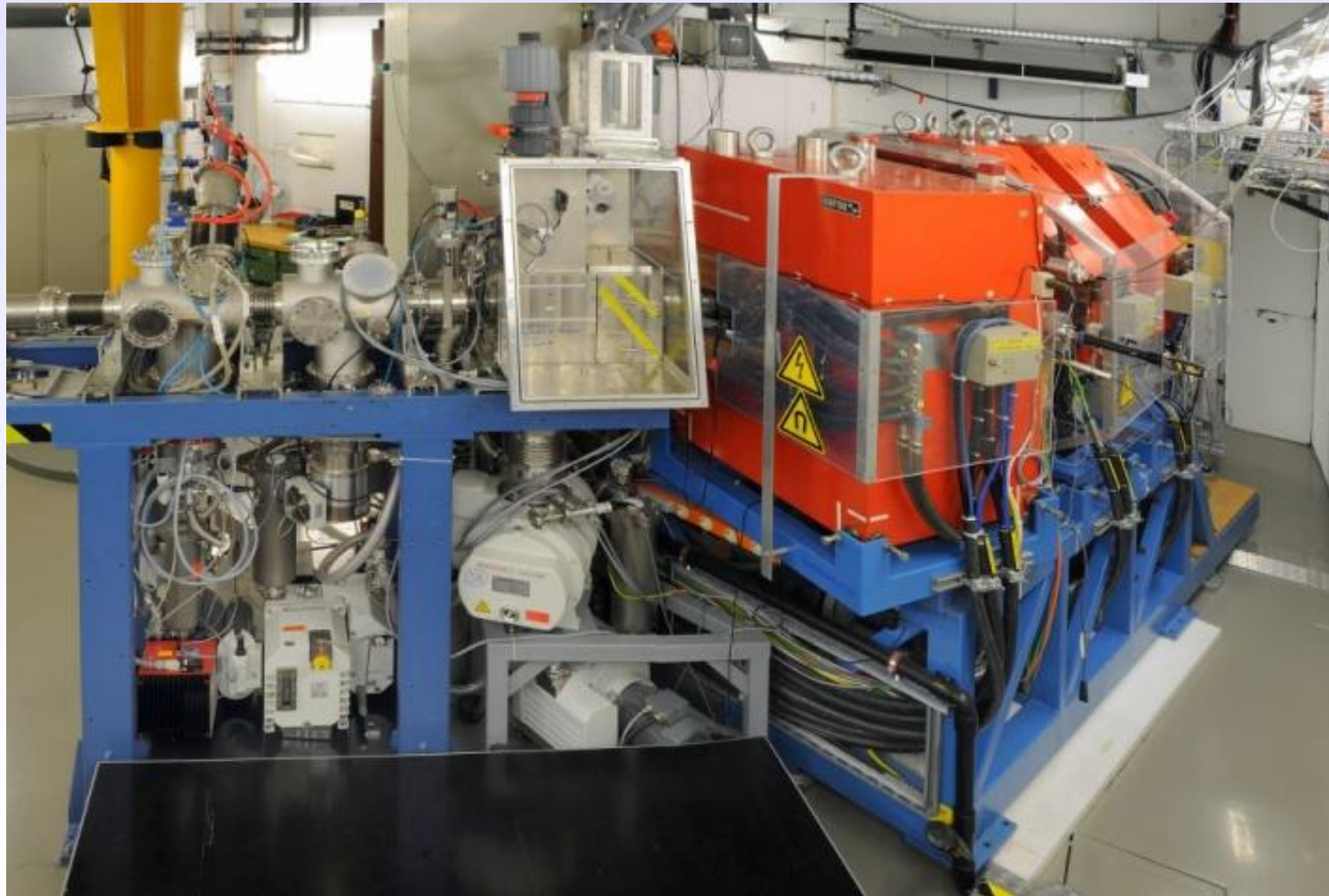
(2007-2014)

A/Z	Setup	Laboratory	Publications
²⁸³112	SHIP	GSI Darmstadt	Eur. Phys. J. A 32, 251 (2007)
²⁸³112	COLD	PSI-FLNR (JINR)	NATURE 447, 72 (2007)
^{286, 287}114	BGS	LRNL (Berkeley)	P.R. Lett. 103, 132502 (2009)
^{288, 289}114	TASCA	GSI – Mainz	P.R. Lett. 104, 252701 (2010)
^{292, 293}116	SHIP	GSI Darmstadt	Eur. Phys. J. A 48: 62 (2012)
^{287, 288}115	TASCA	GSI – Mainz	P.R. Lett. 111, 112502 (2013)
²⁹⁴117	TASCA	GSI-Mainz	P.R. Lett. 112, 172501 (2014)

The Velocity Filter «SHIP»



TransActiniden Separator und Chemie Apparatur (TASCA) (GSI, Darmstadt)



The Festive Naming Ceremony of the new chemical elements **114 - Flerovium** and **116 - Livermorium** took place on 24 October 2012 in Moscow



IUPAC President Prof.
Kazuyuki Tatsumi



**William Goldstein (Livermore),
Yuri Oganessian (JINR) and
James Roberto (Oak Ridge)**

30th December 2015:

Discovery of new elements with atomic numbers **115, 117,**
and **118** is approved!

Priority for elements 115 and 117 is assigned to:

JINR (Dubna) - LLNL (California, USA) - ORNL (Oak
Ridge, USA) collaboration.

Priority for element 118 is assigned to:

JINR (Dubna) - LLNL (California, USA) collaboration.



I U P A C

International Union of Pure
and Applied Chemistry

The 7th period of the periodic table of elements is now complete!

113 (Nh) (Nihonium)	114 Fl Flerovium	115 (Mc) (Moscovium)	116 Lv Livermorium	117 (Ts) (Tenessine)	118 (Og) (Oganesson)
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**All the elements were synthesized for the first time at the U400 cyclotron of
the Flerov Laboratory of Nuclear Reactions of JINR.**

D.I. Mendeleev's Periodic Table of the Elements (2015)

1																	18
IA																	VIII A
Водород H 1,00794 Hydrogen	2											13	14	15	16	17	Гелий He 4,0026 Helium
Литий Li 6,941 Lithium	Бериллий Be 9,01218 Beryllium											Бор B 10,811 Boron	Углерод C 12,011 Carbon	Азот N 14,0067 Nitrogen	Кислород O 15,9994 Oxygen	Фтор F 18,9984 Fluorine	Неон Ne 20,1797 Neon
Натрий Na 22,989768 Sodium	Магний Mg 24,3050 Magnesium	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
Калий K 39,0983 Potassium	Кальций Ca 40,078 Calcium	III B	IV B	V B	VI B	VII B	VIII B	VIII B	VIII B	IB	IIB	Алюминий Al 26,981539 Aluminum	Кремний Si 28,0855 Silicon	Фосфор P 30,97376 Phosphorus	Сера S 32,066 Sulfur	Хлор Cl 35,4527 Chlorine	Аргон Ar 39,948 Argon
Рубидий Rb 85,4678 Rubidium	Стронций Sr 87,62 Strontium	Иттрий Y 88,90585 Yttrium	Цирконий Zr 91,224 Zirconium	Никобий Nb 92,90638 Niobium	Молибден Mo 95,94 Molybdenum	Технеций Tc [98] Technetium	Рутений Ru 101,07 Ruthenium	Родий Rh 102,90550 Rhodium	Палладий Pd 106,42 Palladium	Серебро Ag 107,8682 Silver	Кадмий Cd 112,411 Cadmium	Галлий Ga 69,723 Gallium	Германий Ge 72,61 Germanium	Мышьяк As 74,92159 Arsenic	Селен Se 78,96 Selenium	Бром Br 79,904 Bromine	Криптон Kr 83,80 Krypton
Цезий Cs 132,90543 Cesium	Барий Ba 137,327 Barium	Лантан La 138,9055 Lanthanum	Гафний Hf 178,49 Hafnium	Тантал Ta 180,9479 Tantalum	Вольфрам W 183,84 Tungsten	Рений Re 186,207 Rhenium	Осмий Os 190,23 Osmium	Иридий Ir 192,22 Iridium	Платина Pt 195,08 Platinum	Золото Au 196,96654 Gold	Ртуть Hg 200,59 Mercury	Таллий Tl 204,3833 Thallium	Свинец Pb 207,2 Lead	Висмут Bi 208,98037 Bismuth	Полоний Po [209] Polonium	Астат At [210] Astatine	Радон Rn [222] Radon
Франций Fr [223] Francium	Радий Ra 226,025 Radium	Актиний Ac [227] Actinium	Резерфордий Rf [261] Rutherfordium	Дубний Db [262] Dubnium	Сибборгий Sg [266] Seaborgium	Бергий Bh [262] Bohrium	Хассий Hs [269] Hassium	Мейтнерий Mt [268] Meitnerium	Дармштадтий Ds [269] Darmstadtium	Рентгений Rg [272] Roentgenium	Коперниций Cn [277] Copernicium	113	Флеровий Fl [287] Flerovium	115	Ливерморий Lv [289] Livermorium	117	118

Лантаноиды Lanthanides

Церий Ce 140,115 Cerium	Прометий Pr 140,90765 Promethium	Неодим Nd 144,24 Neodymium	Прометий Pm [145] Promethium	Самарий Sm 150,36 Samarium	Европий Eu 151,965 Europium	Гадолиний Gd 157,25 Gadolinium	Тербий Tb 158,92534 Terbium	Диспрозий Dy 162,50 Dysprosium	Гольмий Ho 164,93032 Holmium	Эрбий Er 167,26 Erbium	Тулий Tm 168,93421 Thulium	Иттербий Yb 173,04 Ytterbium	Лютеций Lu 174,967 Lutetium
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Водород H 1,00794 Hydrogen
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Актиноиды Actinides

Торий Th 232,0381 Thorium	Протактиний Pa 231,03688 Protactinium	Уран U 238,0289 Uranium	Нептуний Np [237] Neptunium	Плутоний Pu [244] Plutonium	Америций Am [243] Americium	Кюрий Cm [247] Curium	Беркелий Bk [247] Berkelium	Калифорний Cf [251] Californium	Эйнштейний Es [252] Einsteinium	Фермий Fm [257] Fermium	Менделевий Md [258] Mendelevium	Нобелий No [259] Nobelium	Лоуренсий Lr [262] Lawrencium
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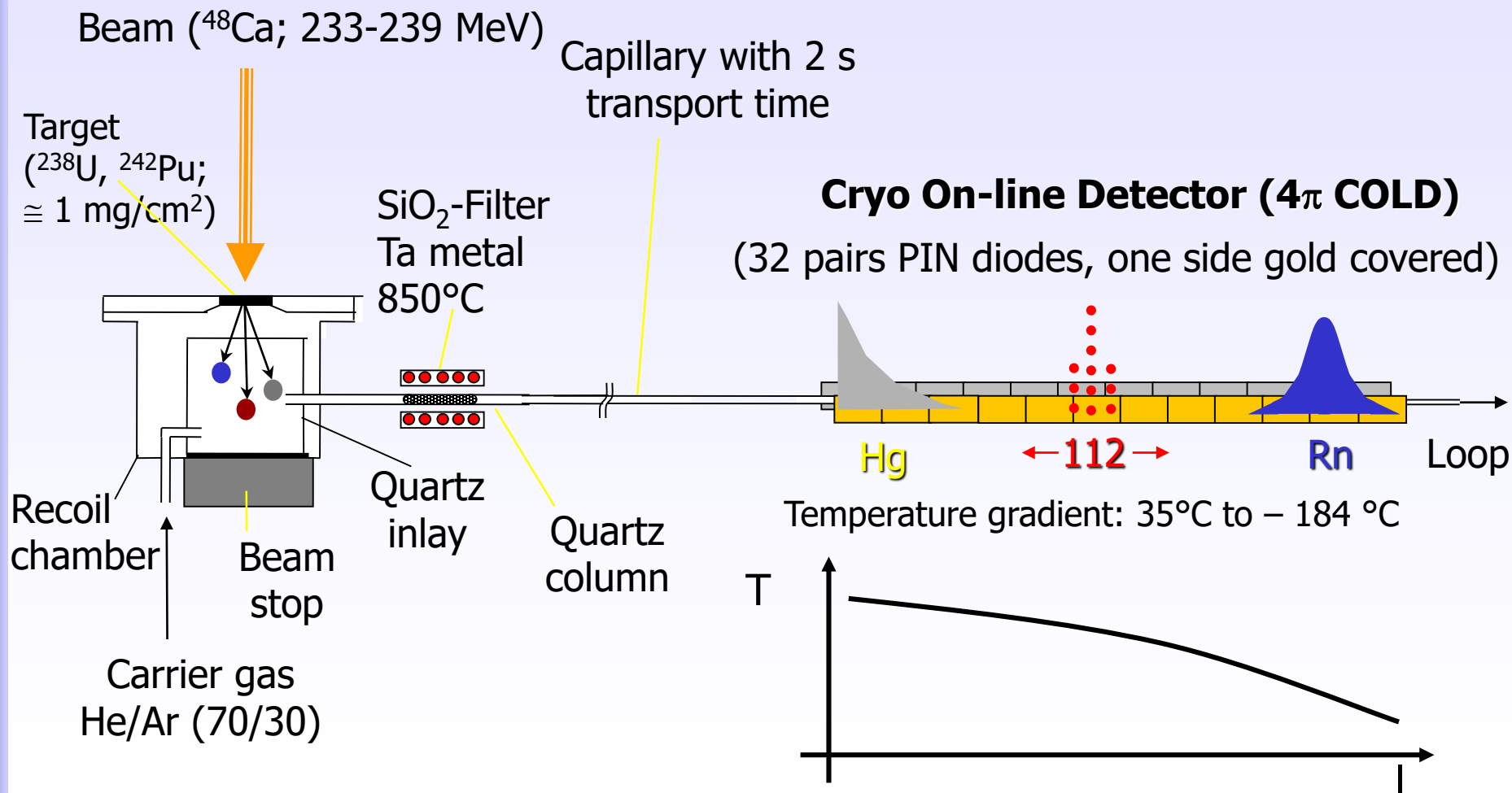
H - символ
1,00794 - атомный номер
1s² - электронная конфигурация
13,59844 - 1-й потенциал ионизации, эВ
0,0899 - плотность кг/м³
-259,34 - температура плавления, °C
-252,87 - температура кипения, °C

■ s-элементы ■ d-элементы
■ p-элементы ■ f-элементы

GAS PHASE CHEMISTRY WITH ELEMENTS 112 AND 114

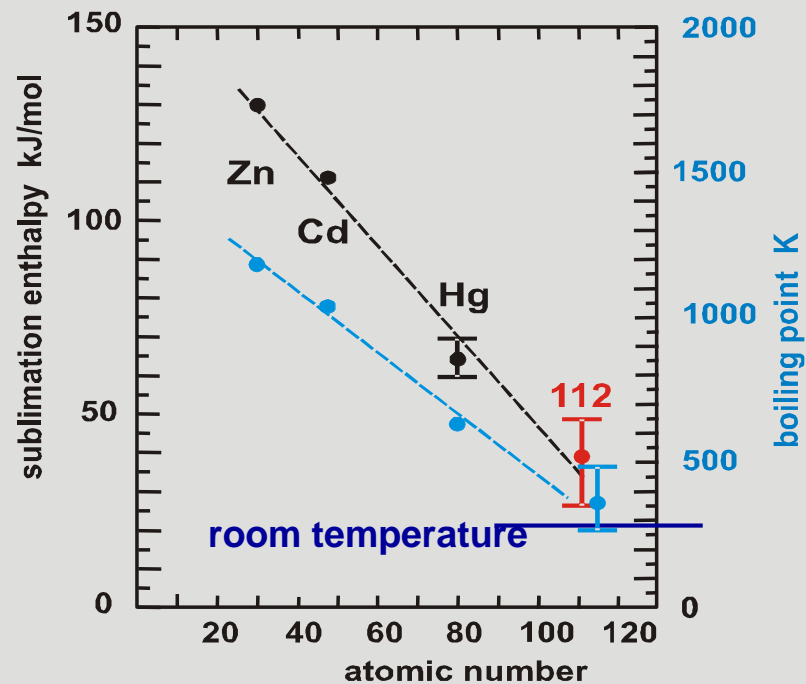
- **Are elements 112 and 114 volatile metals?**
- **How do relativistic effects influence the chemistry of E112 and of E114?**

Химия элементов 112 & 114



Trend of sublimation enthalpy within group 12

Element 112 is a noble metal – like Hg



As predicted by Bernd Eichler, 1974 (!)

That we have learnt:

- **SHE can be synthesized;**
- **Chemistry of SHE can be studied;**
- **We have only 12,000 hours beam time / year;**
- **We need new facilities;**
- **We have not enough experimental space;**
- **We can not accelerate ions heavier than Xe;**
- **Radiation safety requirements are strong;**

What to do further?

Complete fusion reactions:

- Higher beam current;
- More target material: 15 mg → 150 mg;

Heaviest target: $^{249}\text{Cf} \rightarrow Z_{\text{max}} = 118$



- Heavier projectiles (^{50}Ti , ^{54}Cr , ^{58}Fe , ^{64}Ni ...)
- Heavier targets: ^{250}Cm , ^{251}Cf ;
- Exotic targets: ^{254}Es , ^{257}Fm -???
- Symmetric reactions:
 $^{136}\text{Xe} + ^{136}\text{Xe}$, $^{136}\text{Xe} + ^{150}\text{Nd}$, $^{150}\text{Nd} + ^{150}\text{Nd}$;
- Reactions with RI.

DC280-cyclotron – stand-alone SHE-factory



- Synthesis and study of properties of superheavy elements.
- Search for new reactions for SHE-synthesis.
- Chemistry of new elements.

DC280 (expected) E=4÷8 MeV/A		
Ion	Ion energy [MeV/A]	Output intensity
${}^7\text{Li}$	4	1×10^{14}
${}^{18}\text{O}$	8	1×10^{14}
${}^{40}\text{Ar}$	5	6×10^{13}
${}^{48}\text{Ca}$	5	$0,6-1,2 \times 10^{14}$
${}^{54}\text{Cr}$	5	2×10^{13}
${}^{58}\text{Fe}$	5	1×10^{13}
${}^{124}\text{Sn}$	5	2×10^{12}
${}^{136}\text{Xe}$	5	1×10^{14}
${}^{238}\text{U}$	7	5×10^{10}

Specialized high-current cyclotron DC280

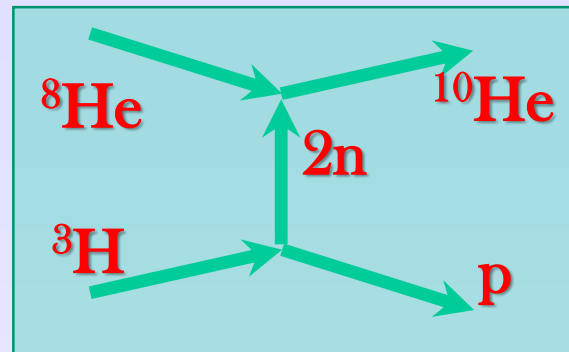
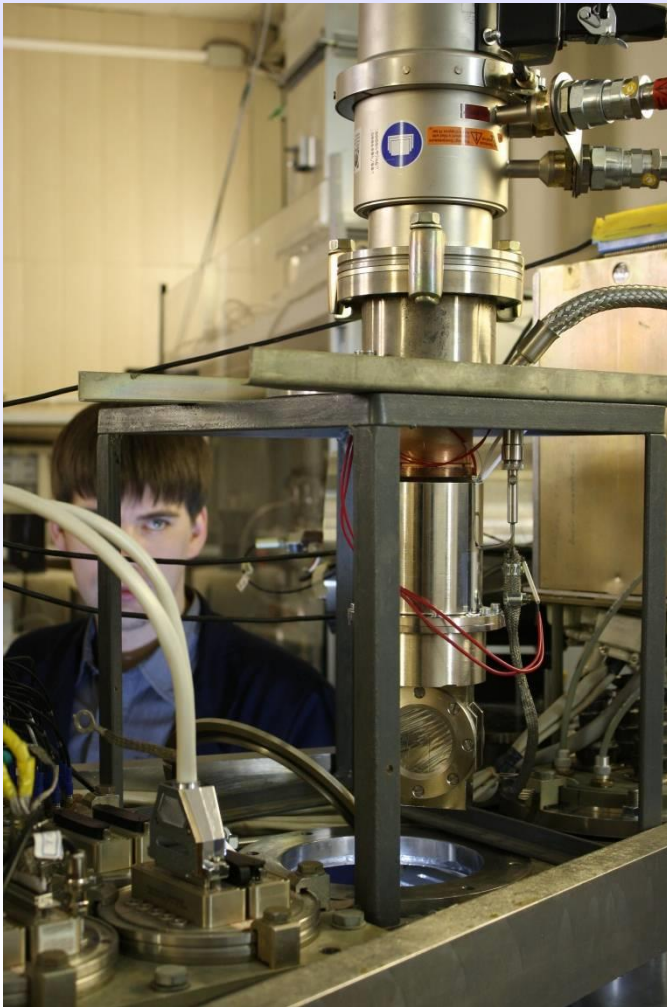


SHE-Factory (June, 2016)



<http://inflnr.jinr.ru/dc280.html>

^{10}He : 2n-transfer



Fragment-separator ACCULINNA-2: assembling and testing



Radiation-physical and radioisotope investigations

detailed study of effects induced by heavy ions in matter aimed at applications of beams of accelerated ions in nanotechnology,

investigation of radiation resistance of materials under the influence of multi-charged ions,

testing of microelectronic circuits for space technology,

development of next-generation of functional track membranes,

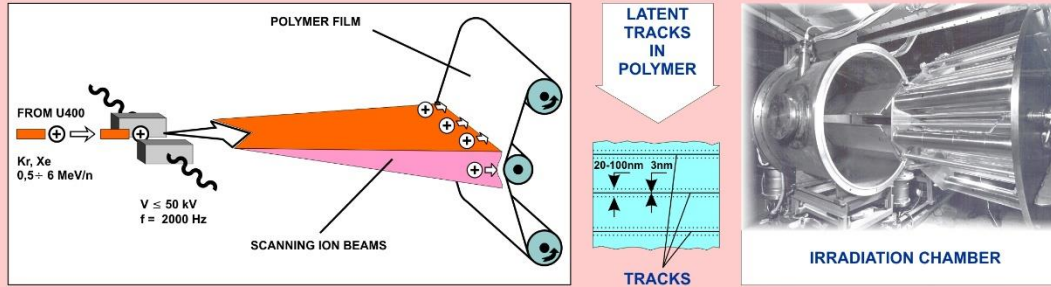
synthesis of nano-objects with unique properties for new applications,

development of hybrid nanotechnologies, combining methods of ion track technology and coating, thin-layer, and surface modification technologies,

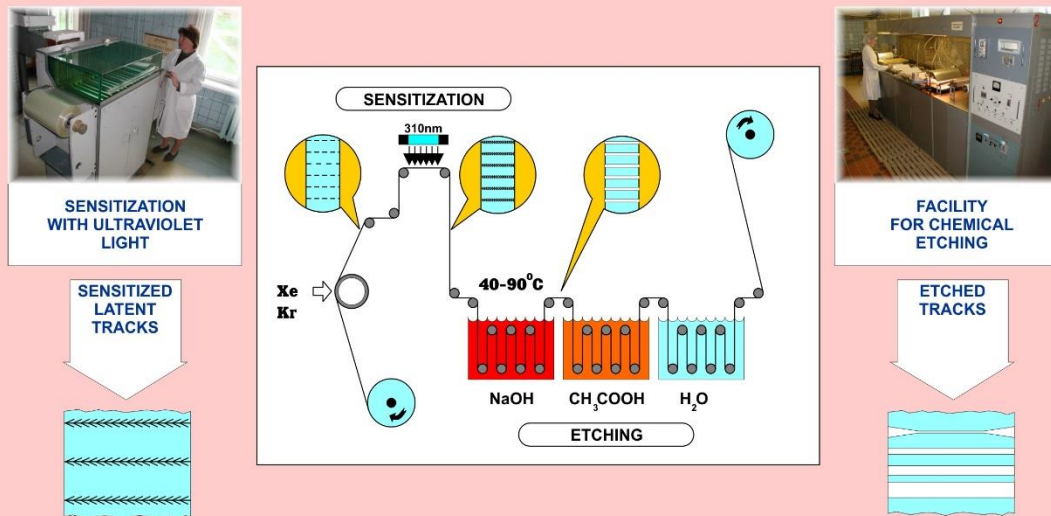
production of radioisotopes for nuclear medicine and radioecological studies with γ -quanta, α -particle-, and heavy-ion beams;

TRACK-ETCH MEMBRANES

I. IRRADIATION WITH ACCELERATED HEAVY IONS



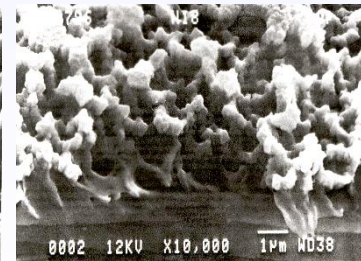
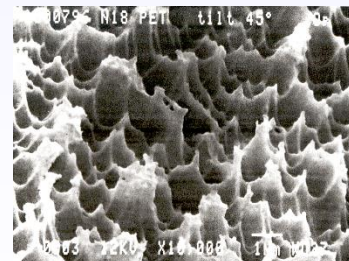
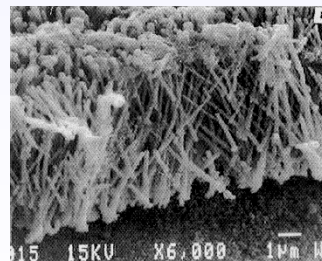
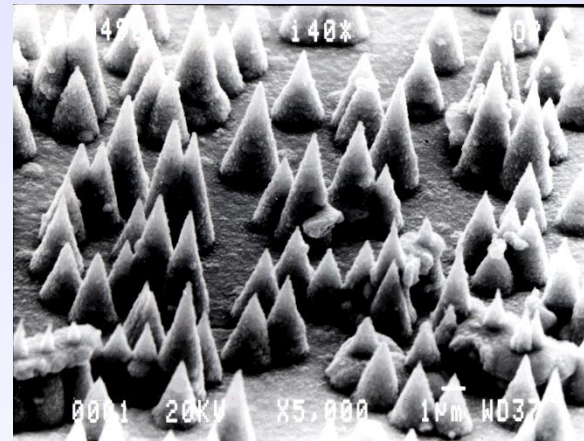
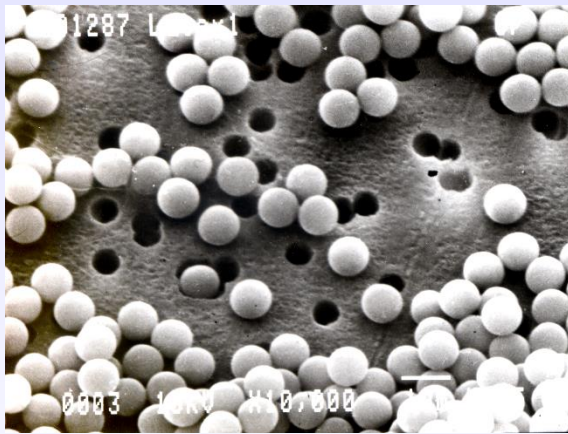
II. SENSITIZATION AND CHEMICAL ETCHING



FLEROV LABORATORY OF NUCLEAR REACTIONS
JOINT INSTITUTE FOR NUCLEAR RESEARCH

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Nano-structures



New research instruments (the Nanotechnology Centre)

- Two scanning electron microscopes (Hitachi SU8020 and SU3400) with a number of options (microanalysis, catodoluminescence, etc)
- Specialized equipment for sample preparation for SEM
- Atomic force microscopy with various optical options
- X Ray photoelectron spectroscopy (K-Alpha instrument)
- Fourier-transform IR spectrophotometer
- UV-Vis spectrophotometers
- Specialized electronics for measurements on single nanopores: Axopatch 200B
- Capillary flow porometer and other equipment for membrane testing
- Versatile equipment for chemical lab, etc.

General view of the new laboratory building (Centre of Nanotechnology)

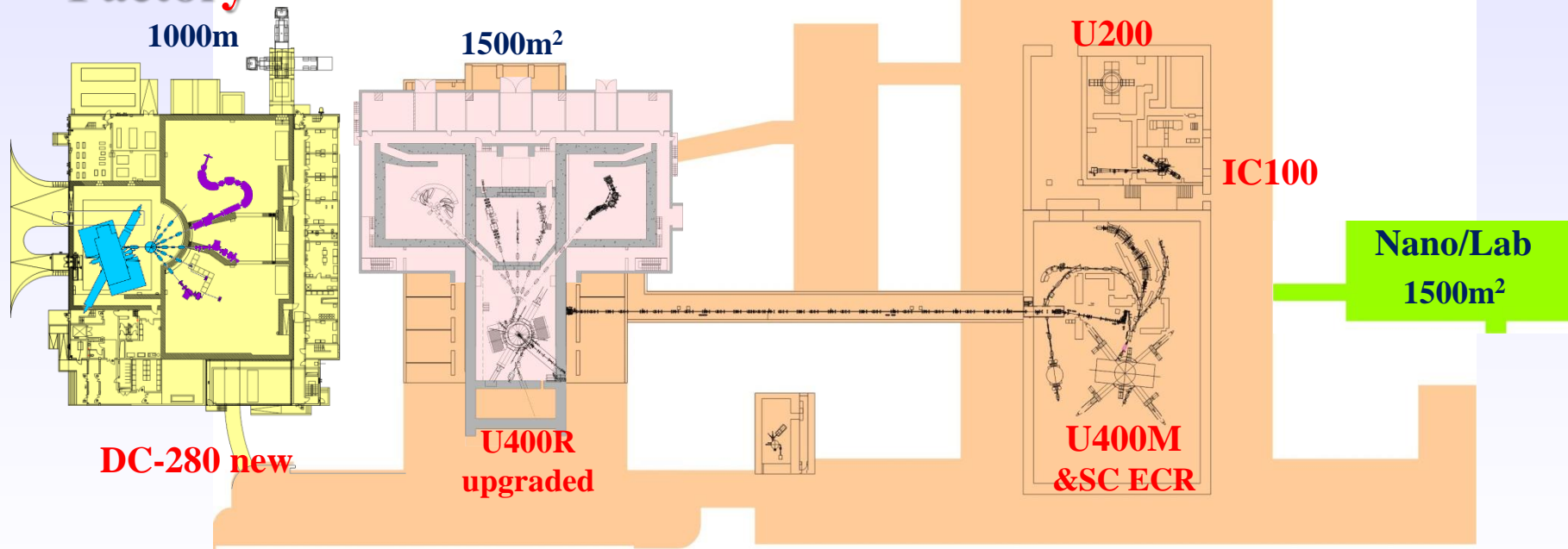


Centre of Nanotechnology, interior view



Full-scale realization off the DRIBs-III -project

SHE Factory



Heavy &
Super heavy

Nuclear
spectroscopy

Light exotic nuclei &
Applied research



**THANK YOU
FOR YOUR
ATTENTION !**

Elerov Laboratory of Nuclear Reactions (JINR)